

US009464368B2

(12) United States Patent

Zussman et al.

(10) Patent No.: US 9,464,368 B2

(45) **Date of Patent:** *Oct. 11, 2016

(54) METHODS OF ATTACHING A MOLECULE-OF-INTEREST TO A MICROTUBE

(75) Inventors: Eyal Zussman, Haifa (IL); Yael Dror, Misgav (IL); Jonathan Charles Kuhn,

Haifa (IL)

(73) Assignee: TECHNION RESEARCH & DEVELOPMENT FOUNDATION LTD., Haifa (IL)

(*) Notice: Subject to any disclaimer, the term of this

patent is extended or adjusted under 35 U.S.C. 154(b) by 2 days.

This patent is subject to a terminal dis-

claimer.

(21) Appl. No.: 12/918,372
(22) PCT Filed: Feb. 12, 2009

(86) PCT No.: **PCT/IL2009/000171**

§ 371 (c)(1),

(2), (4) Date: **Dec. 16, 2010**

(87) PCT Pub. No.: WO2009/104176

PCT Pub. Date: Aug. 27, 2009

(65) Prior Publication Data

US 2011/0081394 A1 Apr. 7, 2011

Related U.S. Application Data

- (60) Provisional application No. 61/064,206, filed on Feb. 21, 2008, provisional application No. 61/064,210, filed on Feb. 21, 2008, provisional application No. 61/064,204, filed on Feb. 21, 2008.
- (51) **Int. Cl. D01D 5/24** (2006.01) **D01F 1/08** (2006.01)
- (Continued)

(52) U.S. CI. CPC D01D 5/003 (2013.01); A61K 9/0092 (2013.01); B82Y 30/00 (2013.01); C02F 3/102 (2013.01);

(Continued)

(58) **Field of Classification Search** CPC A61K 9/0092; A61K 9/00; A61K 9/48;

A61K 9/50; A61K 9/4816; A61K 9/5005; D01D 5/003; D01D 5/007

See application file for complete search history.

(56) References Cited

U.S. PATENT DOCUMENTS

4,921,475 A 5/1990 Sibalis 5,209,734 A 5/1993 Hurley et al. (Continued)

FOREIGN PATENT DOCUMENTS

AU 2007303821 4/2008 CN 1799649 7/2006 (Continued)

OTHER PUBLICATIONS

Zhang et al. Biomimetic and bioactive nanofibrous scaffolds from electrospun composite nanofibers. International Journal of Nanomedicine 2007, vol. 2, No. 4, pp. 623-638.*

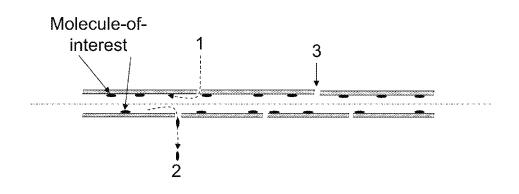
(Continued)

Primary Examiner — Shafiqul Haq (74) Attorney, Agent, or Firm — The Roy Gross Law Firm, LLC; Roy Gross

(57) ABSTRACT

Provided is a method of attaching a molecule-of-interest to a microtube, by co-electrospinning two polymeric solutions through co-axial capillaries, wherein a first polymeric solution of the two polymeric solutions is for forming a shell of the microtube and a second polymeric solution of the two polymeric solutions is for forming a coat over an internal surface of the shell, the first polymeric solution is selected solidifying faster than the second polymeric solution and a solvent of the second polymeric solution is selected incapable of dissolving the first polymeric solution and wherein the second polymeric solution comprises the molecule-ofinterest, thereby attaching the molecule-of-interest to the microtube. Also provided is an electrospun microtube comprising an electrospun shell, an electrospun coat over an internal surface of the shell and a molecule-of-interest attached to the microtube.

18 Claims, 12 Drawing Sheets (6 of 12 Drawing Sheet(s) Filed in Color)



(51)	Int. Cl.	
	A61K 9/00	(2006.01)
	D01D 5/00	(2006.01)
	B82Y 30/00	(2011.01)
	C02F 3/10	(2006.01)
	C02F 3/34	(2006.01)
	D01D 5/247	(2006.01)
	D01F 1/10	(2006.01)
	D01F 8/14	(2006.01)
	G01N 33/543	(2006.01)
	C02F 101/18	(2006.01)
	C02F 101/20	(2006.01)
	C02F 101/30	(2006.01)
	C02F 101/36	(2006.01)
(52)	U.S. Cl.	
	CPC	C 02F 3/34 (2013.01); C
	(2013.01); D	001D 5/247 (2013.01);

Communication Co2F 3/34 (2013.01); Co2F 3/342 (2013.01); Do1D 5/247 (2013.01); Do1F 1/10 (2013.01); Do1F 8/14 (2013.01); Go1N 33/54393 (2013.01); Co2F 2101/18 (2013.01); Co2F 2101/20 (2013.01); Co2F 2101/306 (2013.01); Co2F 2101/36 (2013.01); Co2

(56) References Cited

U.S. PATENT DOCUMENTS

5,688,501	A	11/1997	Merril et al.
5,795,340	A	8/1998	Lang
6,537,195	B2	3/2003	Forman
6,537,241	B1	3/2003	Odland
7,066,922	B2	6/2006	Angel et al.
2001/0014394	$\mathbf{A}1$	8/2001	Soane et al.
2001/0034503	$\mathbf{A}1$	10/2001	Mehier
2003/0098518	A1*	5/2003	Averdung et al 264/10
2003/0135158	$\mathbf{A}1$	7/2003	Gonnelli
2003/0139727	$\mathbf{A}1$	7/2003	Angel et al.
2004/0018226	$\mathbf{A}1$	1/2004	Wnek et al.
2004/0030377	$\mathbf{A}1$	2/2004	Dubson et al.
2004/0147903	$\mathbf{A}1$	7/2004	Latini
2004/0223954	$\mathbf{A}1$	11/2004	Bruessow et al.
2005/0180992	$\mathbf{A}1$	8/2005	Belcher et al.
2006/0119015	$\mathbf{A}1$	6/2006	Wehrspohn et al.
2006/0200232	$\mathbf{A}1$	9/2006	Phaneuf et al.
2006/0226580	$\mathbf{A}1$	10/2006	Xia et al.
2006/0228435	$\mathbf{A}1$	10/2006	Andrady et al.
2009/0061496	$\mathbf{A}1$	3/2009	Kuhn et al.
2010/0129656	$\mathbf{A}1$	5/2010	Zussman et al.
2010/0303881	$\mathbf{A}1$	12/2010	Hoke et al.
2011/0081394	$\mathbf{A}1$	4/2011	Zussman et al.

FOREIGN PATENT DOCUMENTS

EP	2079860	7/2009
WO	WO 03/000381	1/2003
WO	WO 2006/019293	2/2006
WO	WO 2006/108809	10/2006
WO	WO 2008/041183	4/2008
WO	WO 2009/104174	8/2009
WO	WO 2009/104175	8/2009
WO	WO 2009/104176	8/2009

OTHER PUBLICATIONS

Yarin et al. Material encapsulation and transport in core-shell micro/nanofibers, polymer and carbon nanotubes and micro/nanochannels. J. Mater. Chem., 2007, vol. 17, pp. 2585-2599.* Kim et al. Controlled protein release from electrospun biodegradable fiber mesh composed of poly(e-caprolactone) and poly(ethylene oxide). International Journal of Pharmaceutics 2007, vol. 338, pp. 276-283.*

Official Action Dated Jun. 20, 2012 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,377.

Official Action Dated Feb. 14, 2012 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311,601.

Response Dated Feb. 1, 2011 to Official Action of Oct. 1, 2010 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896,130. Response Dated Mar. 24, 2011 to Office Action of Oct. 13, 2010 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1.

Official Action Dated Apr. 7, 2011 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896,130.

Zussman et al. "Formation of Nanofiber Crossbar in Electrospinning", Applied Physics Letters, 82(6): 973-975, Feb. 10, 2003. Response Dated Aug. 2, 2011 to Official Action of Apr. 7, 2011 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896.130.

Larsson et al. "Detection of Number and Viability of *E. coli* and A. Hydrophila With FISH Technique", Techneau, D.3.5.3, p. 1-30, Apr. 30, 2008.

International Preliminary Report on Patentability Dated Apr. 16, 2009 From the International Bureau of WIPO Re.: Application No. PCT/IB2007/054001.

International Search Report and the Written Opinion Dated Sep. 29, 2009 From the International Searching Authority Re.: Application No. PCT/IL2009/000169.

International Search Report and the Written Opinion Dated Sep. 29, 2009 From the International Searching Authority Re.: Application No. PCT/IL2009/000171.

International Search Report and the Written Opinion Dated Sep. 30, 2009 From the International Searching Authority Re.: Application No. PCT/IL 2009/000170.

International Search Report Dated Oct. 14, 2008 From the International Searching Authority Re.: Appliation No. PCT/IB2007/054001.

Written Opinion Dated Oct. 14, 2008 From the International Searching Authority Re.: Application No. PCT/IB2007/054001.

Bognitzki et al. "Polymer, Metal, and Hybrid Nano- and Mesotubes by Coating Degradable Polymer Template Fibers (TUFT Process)", Advanced Materials, 12(9): 637-640, 2000.

Caruso et al. "Titanium Dioxide Tubes From Sol-Gel Coating of Electrospun Polymer Fibers", Advanced Materials, 13: 1577-1579, Oct. 16, 2001.

Dror et al. "One-Step Production of Polymeric Microtubes by Co-Electrospinning", Small, XP002497054, 3(6): 1064-1073, Jun. 4, 2007.

Dror et al. "Viable Encapsulation of Enzymes in Biodegradable Tubular Structures", Faculty of Mechanical Engineering, Faculty of Biology, Technion, Israel Institute of Technology, Haifa, IL, 18 P. Huang et al. "Encapsulating Drugs in Biodegradable Ultrafine Fibers Through Co-Axial Electrospinning", Journal of Biomedical Materials Research, Part A, 77A: 169-179, 169, 2006.

Jiang et al. "A Facile Technique to Prepare Biodegradable Coaxial Electrospun Nanofibers for Controlled Release of Bioactive Agents", Journal of Controlled Release, XP005163067, 108(2-3): 237-243, Nov. 28, 2005.

Jiang et al. "Modulation of Protein Release From Biodegradable Core-Shell Structured Fibers Prepared by Coaxial Electrospinning", Journal of Biomedical Materials Research Part B: Applied Biomaterials, 79B: 50-57, 2006.

Li et al. "Direct Fabrication of Composite and Ceramic Hollow Nanofibers by Electrospinning", Nano Letters, 4(5): 933-938, 2004. Li et al. "Electrospinning of Nanofibers: Reinventing the Wheel?", Advanced Materials, 16(14): 1151-1170, Jul. 19, 2004.

Li et al. "Electrospinning: A Simple and Versatile Technique for Producing Ceramic Nanofibers and Nanotubes", Journal of the American Ceramic Society, 89(6): 1861-1869, 2006.

Li et al. "Use of Electrospinning to Directly Fabricate Hollow Nanofibers With Functionalized Inner and Outer Surfaces", Small, XP002497053, 1(1): 83-86, Jan. 1, 2005.

Loscertales et al. "Electrically Forced Coaxial Nanojets for One-Step Hollow Nanofiber Design", Journal of the American Chemical Society, JACS, 126: 5376-5377, 2004.

(56) References Cited

OTHER PUBLICATIONS

Loscertales et al. "Micro/Nano Encapsulation Via Electrified Coaxial Liquid Jets", Science, 295: 1695-1698, Mar. 16, 2002. Reneker et al. "Electrospinning of Nanofibers From Polymer Solutions and Melts", Advances in Applied Mechanics, 41: 1-3, 103-115, 142-153, 2006.

Reznik et al. "Evolution of A Compound Droplet Attached to a Core-Shell Nozzle Under the Action of a Strong Electric Field", Physics of Fluids, 18: 062101-1 -062101-13, 2006.

Salalha et al. "Encapsulation of Bacteria and Viruses in Electrospun Nanofibres", Nanotechnology, 17: 4675-4681, 2006.

Sun et al. "Compound Core-Shell Polymer Nanofibers by Co-Electrospinning", Advanced Materials, XP002497055, 15(22: 1929-1932, Nov. 17, 2003.

Xie et al. "Ultra-High Surface Fibrous Membranes From Electrospinning of Natural Proteins: Casein and Lipase Enzyme", Journal of Materials Science, 38: 2125-2133, 2003.

Yarin et al. "Material Encapsulation and Transport in Core-Shell Micro/Naonofibers, Polymer and Carbon Nanotubes and Micro/Nanochannels", Journal of Materials Chemistry, XP002546457, 17(25): 2585-2599, Jul. 1, 2007. Chapter III Section (ii).

Yu et al. "Production of Submicrometer Diameter Fibers by Two-Fluid Electrospinning", Advanced Materials, 16(17): 1562-1566, Sep. 3, 2004.

Zhang et al. "Coaxial Electrospinning of (Fluorescein Isothiocyanate-Conjugated Bovine Serum Albumin)-Encapsulated Poly(ε-Caprolactone) Nanofibers for Sustained Release", Biomacromolecules, 7(4): 1049-1057, 2006.

Zussman et al. "Electrospun Polyacrylonitrile/Poly(Methyl Methacrylate)-Derived Turbostratic Carbon Micro-/Nanotubes", Advanced Materials, 18: 348-353, 2006.

Official Action Dated Nov. 3, 2011 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896,130.

Communication Pursuant to Article 94(3) EPC Dated Dec. 2, 2011 From the European Patent Office Re. Application No. 07826621.0. Restriction Official Action Dated Dec. 23, 2011 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311,601.

Official Action Dated Oct. 1, 2010 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896,130.

Lee et al. "Virus-Based Fabrication of Micro- and Nanofibers Using Electrospinning", NanoLetters, 4(3): 387-390, 2004.

Salalha et al. "Encapsulation of Bacteria and Viruses in Electrospun Nanofibres", Nanotechnology, 17: 4675-4681, Aug. 30, 2006.

Theron et al. "Electrostatic Field-Assisted Alignment of Electrospun Nanofibres", Nanotechnology, 12: 384-390, 2001.

Interantional Preliminary Report on Patentability Dated Sep. 2, 2010 From the International Bureau of WIPO Re. Application No. PCT/IL2009/000171.

Translation of Office Action Dated Oct. 13, 2010 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1.

He et al. "Recent Development of the Nanocomposites Prepared by Coaxial Jet Technology", Acta Materiae Compositae Sinica, 22(6): 1-8, Dec. 2005. Abstract in English.

Li et al. "Porous Ultrafine Nanofibers Having a Ultrahigh Specific Surface Area", Chinese Science Bulletin, 49(21): 2160-2163, Nov. 2004. Chinese Only!

International Preliminary Report on Patentability Dated Sep. 2, 2010 From the International Bureau of WIPO Re. Application No. PCT/IL2009/000171.

International Preliminary Report on Patentability Dated Sep. 2, 2010 From the International Bureau of WIPO Re.: Application No. PCT/IL2009/000169.

International Preliminary Report on Patentability Dated Sep. 2, 2010 From the International Bureau of WIPO Re.: Application No. PCT/IL2009/0001790.

Translation of Office Action Dated Mar. 1, 2012 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1.

Communication Pursuant to Article 94(3) EPC Dated Jul. 18, 2012 From the European Patent Office Re. Application No. 09713280.7. Communication Pursuant to Article 94(3) EPC Dated Jul. 19, 2012 From the European Patent Office Re. Application No. 09712148.7. Communication Pursuant to Article 94(3) EPC Dated Jul. 19, 2012 From the European Patent Office Re. Application No. 09713264.1. Official Action Dated Aug. 27, 2012 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311,601.

Official Action Dated Feb. 7, 2013 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311,601.

Translation of Office Action Dated Dec. 4, 2012 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1.

Patent Examination Report Dated Nov. 19, 2012 From the Australian Government, IP Australia Re. Application No. 2007303821. Official Action Dated Jun. 18, 2013 From the US Patent and

Trademark Office Re. U.S. Appl. No. 12/918,365. Advisory Action Before the Filing of an Appeal Brief Dated May 16, 2013 From the US Patent and Trademark Office Re. U.S. Appl.

No. 12/311,601. Restriction Official Action Dated May 9, 2012 From the US Patent

and Trademark Office Re. U.S. Appl. No. 12/918,365. Restriction Official Action Dated Mar. 1, 2013 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,365.

Advisory Action Before the Filing of an Appeal Brief Dated Apr. 12, 2013 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311.601.

Official Action Dated Apr. 8, 2013 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,377.

Decision to Grant a European Patent Pursuant to Article 97(1) EPC Dated Jul. 11, 2013 From the European Patent Office Re. Application No. 07826621.0.

Translation of Office Action Dated Jun. 4, 2013 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1.

Communication Under Rule 71(3) EPC Dated Mar. 5, 2013 From the European Patent Office Re. Application No. 07826621.0.

Communication Pursuant to Article 94(3) EPC Dated Aug. 20, 2013 From the European Patent Office Re. Application No. 09713264.1. Advisory Action Before the Filing of an Appeal Brief Dated May 30, 2014 From the US Patent and Trademark Office Re. U.S Appl. No. 12/918,365.

Requisition by the Examiner Dated May 22, 2014 From the Canadian Intellectual Property Office Re. Application No. 2,664,972.

Official Action Dated Oct. 1, 2014 From the US Patent and Trademark Office Re. U.S. Appl. No. 11/896,130.

Office Action Dated Nov. 21, 2013 From the State Intellectual Property Office of the People's Republic of China Re. Application No. 200780044536.1 and Its Translation Into English.

Official Action Dated Jan. 17, 2014 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,365.

Official Action Dated Jan. 9, 2015 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,377.

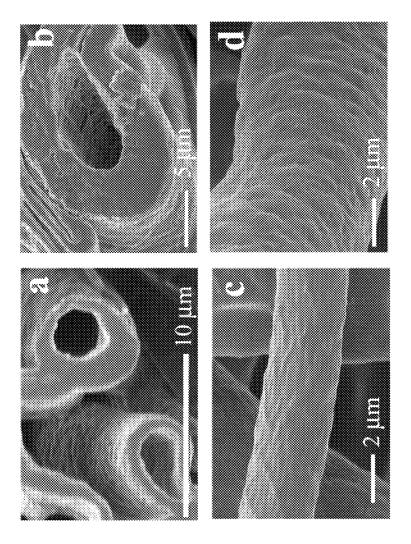
Official Action Dated Jan. 14, 2015 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/311,601.

McCann et al. "Electrospinning of Nanofibers With Core-Sheath, Hollow, or Porous Structures", Journal of Materials Chemistry, 15: 735-738, 2005.

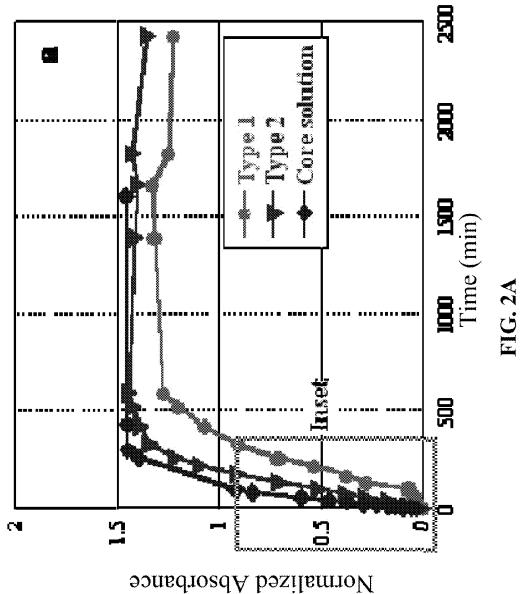
Communication Pursuant to Article 94(3) EPC Dated Aug. 23, 2013 From the European Patent Office Re. Application No. 09713280.7. Official Action Dated Feb. 4, 2015 From the US Patent and Trademark Office Re. U.S. Appl. No. 12/918,365.

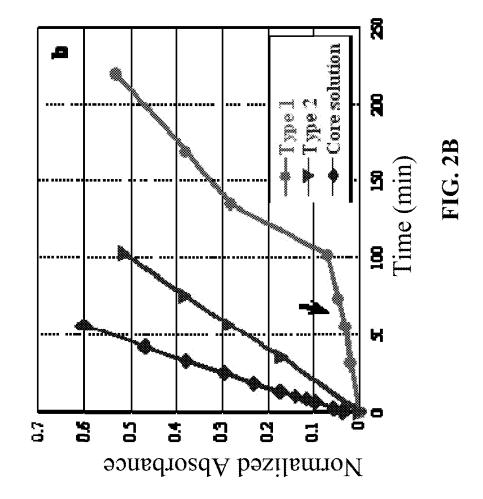
Requisition by the Examiner and Examination Search Report Dated Jan. 23, 2015 From the Canadian Intellectual Property Office Re. Application No. 2,664,972.

* cited by examiner



FIGs. 1A-D





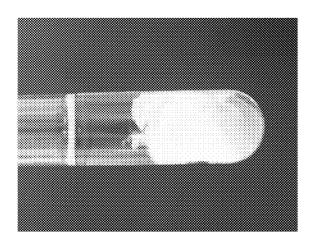
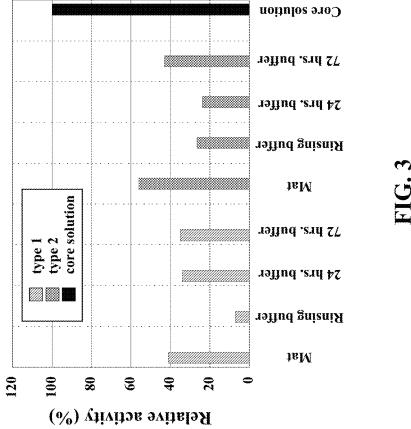
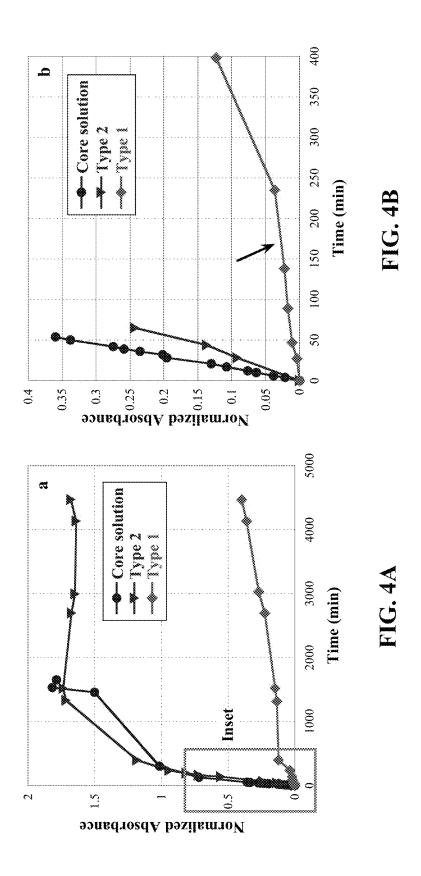
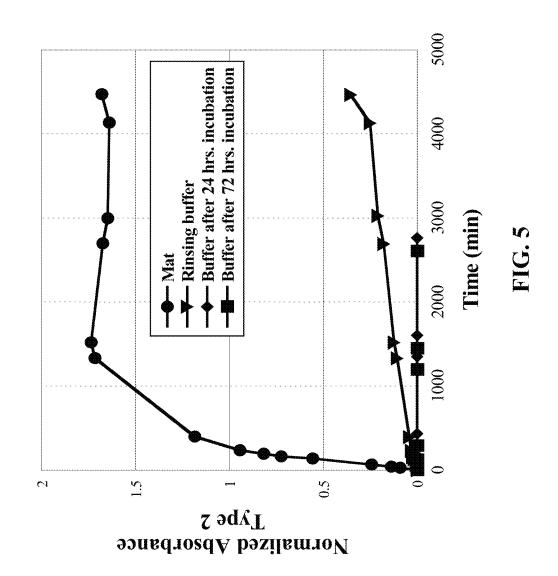


FIG. 2C







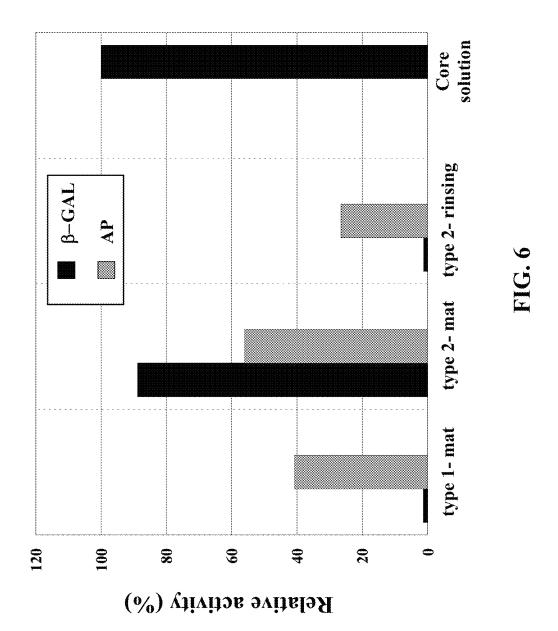
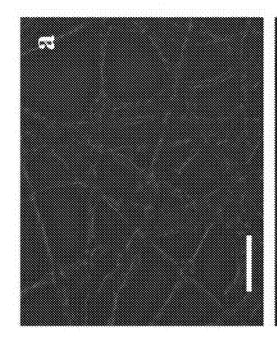
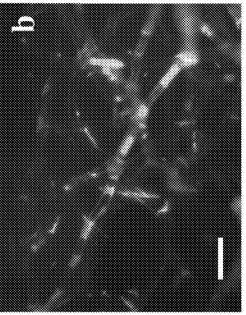
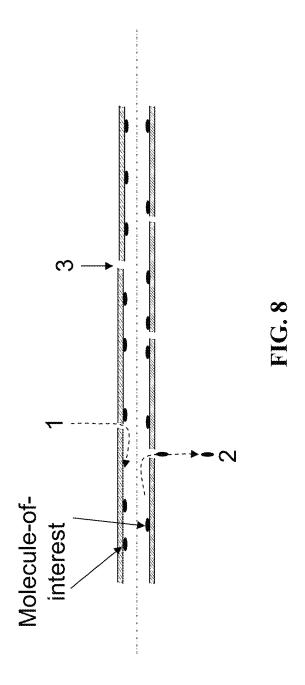


FIG. 7A

IG. 7B







Nutiple Enzyme Pathways

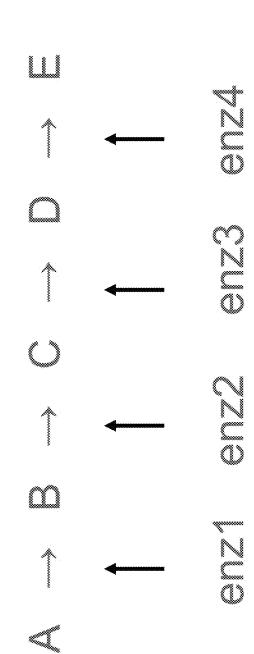
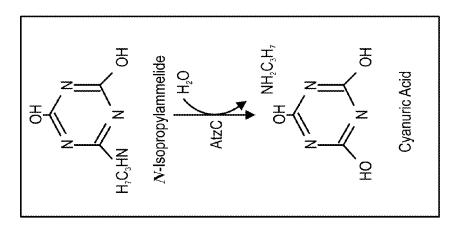
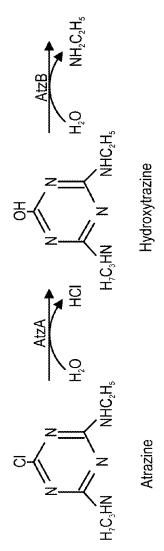


FIG. 9





METHODS OF ATTACHING A MOLECULE-OF-INTEREST TO A MICROTUBE

RELATED APPLICATIONS

This application is a National Phase of PCT Patent Application No. PCT/IL2009/000171 having International filing date of Feb. 12, 2009, which claims the benefit of U.S. Provisional Patent Application Nos. 61/064,210, 61/064,206 10 and 61/064,204 filed on Feb. 21, 2008.

The teachings of PCT/IB2007/054001 are incorporated herein by reference.

The contents of all of the above documents are incorporated by reference as if fully set forth herein.

FIELD AND BACKGROUND OF THE INVENTION

The invention, in some embodiments thereof, relates to a 20 method of attaching a molecule-of-interest to a microtube and, more particularly, but not exclusively, to electrospun microtubes including the molecule-of-interest attached thereto.

catalyze reactions, some of which have industrial use. These include oxidoreductases, transferases, hydrolases, lyases, isomerases, and ligases. Immobilization of enzymes on solid substrates sometimes offers advantages over the use of a free enzyme. For example, immobilization can stabilize 30 enzymes, enable better control of enzymatic reactions, allow the reuse of the enzyme and prevent enzyme loss with time. The material bearing the immobilized enzyme has a significant role in evoking these advantages both from architectural and chemical points of view.

Nanofibers and polymeric nanofibers in particular can be produced by an electrospinning process (Reneker D H., et al., 2006; Ramakrishna S., et al., 2005; Li D., et al., 2004; PCT WO 2006/106506 to the present inventors). Electrospun polymeric nanofibers have been widely used in bio- 40 logical applications such as scaffolds, carriers for biologically active molecules like proteins and enzymes (Xie J., et al., 2003; Zhang Y Z., et al., 2006; Jiang H., et al., 2006; and Patel A C., et al., 2006) and encapsulation of viruses and bacteria (Salalha W., et al., 2006).

Several approaches can be used to entrap or attach enzymes to electrospun fibers. One approach is to immobilize the enzyme on the outer surface of the nanofibers by either covalently attaching the desired enzyme to the functional groups of the polymer surface (Ye P., et al., 2006; Jia 50 H., et al., 2002; Kim T G., et al., 2006) or physically absorbing the enzyme to the surface (Huang X J., et al., 2006). The second approach, which results in encapsulation of enzymes, is based on mixing the enzyme with the polymer solution prior to the electrospinning process (Xie J. 55 and Hsieh Y-L, 2003). However, encapsulation is often associated with leaching of the enzymes, e.g., via fiber dissolution and burst releases (Zhang Y Z., et al., 2006), especially, when the host polymer is a water soluble polymer such as poly(vinyl alcohol) (PVA) or dextran. To prevent 60 immediate dissolution of the fibers in a physiological environment (e.g., blood) and the subsequent enzyme leaching, the electrospun fibers can be crosslinked by chemical or physical agents such as glutaraldehyde or UV irradiation. Alternatively Zeng J, et al. (2005) suggested that PVA fibers 65 can be coated with water insoluble polymers using a chemical vapor deposition (CVD). However, the organic solvents

2

of the water insoluble polymers are harmful to biological material and can lead to loss of enzymatic activity. To overcome this problem, Herricks et al. (2005) suggested to use surfactant-stabilized enzymes in an organic solution of polystyrene (PS) as a spinning solution. In this way the electrospun nanofibers are insoluble in water and the enzymatic activity is retained due to surfactant stabilization (Herricks T E., et al., 2005).

Sun and co-workers (Sun Z, et al., 2003) describe the production of core-shell nanofibers (i.e., filled fibers) by co-electrospinning of two polymeric solutions using a two co-axial capillaries spinneret. US patent application No. 20060119015 to Wehrspohn R., et al. describes the production of hollow fibers by introducing a liquid containing a polymer to a porous template material, and removal of the template following polymer solidification. PCT/IB/2007/ 054001 to the present inventors (which is fully incorporated herein by reference) discloses methods of producing electrospun microtubes (i.e., hollow fibers) which can be further filled with liquids and be used as microfluidics.

SUMMARY OF THE INVENTION

According to an aspect of some embodiments of the In nature there is an enormous variety of enzymes that 25 present invention there is provided a method of attaching a molecule-of-interest to a microtube, the method comprising: co-electrospinning two polymeric solutions through co-axial capillaries, wherein a first polymeric solution of the two polymeric solutions is for forming a shell of the microtube and a second polymeric solution of the two polymeric solutions is for forming a coat over an internal surface of the shell, the first polymeric solution is selected solidifying faster than the second polymeric solution and a solvent of the second polymeric solution is selected incapable of dissolving the first polymeric solution and wherein the second polymeric solution comprises the molecule-of-interest, thereby attaching the molecule-of-interest to the micro-

> According to an aspect of some embodiments of the present invention there is provided a microtube comprising an electrospun shell, an electrospun coat over an internal surface of the shell and a molecule-of-interest attached to the microtube.

According to an aspect of some embodiments of the present invention there is provided a method of processing a substrate-of-interest, comprising contacting the substrateof-interest with the microtube of the invention, wherein the molecule-of-interest is capable of processing the substrate, thereby processing the substrate-of-interest.

According to an aspect of some embodiments of the present invention there is provided a method of depleting a molecule from a solution, comprising contacting the solution with the microtube of the invention, wherein the member of the affinity pair is selected capable of binding the molecule, thereby depleting the molecule from the solution.

According to an aspect of some embodiments of the present invention there is provided a method of isolating a molecule from a solution, comprising: (a) contacting the solution with the microtube of the invention under conditions which allow binding of the molecule to the microtube via the member of the affinity pair which is selected capable of binding the molecule, and; (b) eluting the molecule from the microtube; thereby isolating the molecule from the solution.

According to an aspect of some embodiments of the present invention there is provided a method of detecting a presence of a molecule in a sample, comprising: (a) con-

tacting the sample with the microtube of the invention, wherein the member of the affinity pair is selected capable of binding the molecule, and; (b) detecting binding of the molecule by the member of the affinity pair; thereby detecting the presence of a molecule in the sample.

According to an aspect of some embodiments of the present invention there is provided a method of releasing a molecule-of-interest to cells of a subject in need thereof, comprising implanting in the subject the microtube of the invention, to thereby release the molecule-of-interest to cells 10 of the subject.

According to some embodiments of the invention, the electrospun shell is formed of a first polymeric solution and the electrospun coat is formed of a second polymeric solution.

According to some embodiments of the invention, the first polymeric solution solidifies faster than the second polymeric solution.

According to some embodiments of the invention, a solvent of the second polymeric solution is incapable of 20 dissolving the first polymeric solution.

According to some embodiments of the invention, the electrospun shell comprises a polymer selected from the group consisting of poly (e-caprolactone) (PCL), polyamide, poly(siloxane), poly(silicone), poly(ethylene), poly(vinyl 25 pyrrolidone), poly(2-hydroxy ethylmethacrylate), poly(Nvinyl pyrrolidone), poly(methyl methacrylate), poly(vinyl alcohol), poly(acrylic acid), poly(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(ethylene glycol), poly(methacrylic acid), polylactide, polyglycolide, 30 poly(lactide-coglycolide), polyanhydride, polyorthoester, poly(carbonate), poly(acrylo nitrile), poly(ethylene oxide), polyaniline, polyvinyl carbazole, polystyrene, poly(vinyl phenol), polyhydroxyacid, poly(caprolactone), polyanhydride, polyhydroxyalkanoate, polyurethane, collagen, albu- 35 min, alginate, chitosan, starch, hyaluronic acid, and whereas the electrospun coat comprises a polymer selected from the group consisting of poly(acrylic acid), poly(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(ethylene glycol), poly(methacrylic acid), polylactide polygly- 40 colide, poly(lactide-coglycolide), polyanhydride, polypoly(ethylene poly(carbonate), orthoester. polyaniline, polyvinyl carbazole, polystyrene, poly(vinyl phenol), polyhydroxyacid, alginate, starch, hyaluronic acid.

According to some embodiments of the invention, a 45 solvent of the first polymeric solution evaporates faster than a solvent of the second polymeric solution.

According to some embodiments of the invention, the electrospinning is effected using a rotating collector.

According to some embodiments of the invention, a 50 solvent of the second polymeric solution is capable of evaporating through the internal surface of the shell.

According to some embodiments of the invention, the second polymeric solution is capable of wetting the internal surface of the shell.

According to some embodiments of the invention, a thickness of the shell is from about 100 nm to about 20 micrometer.

According to some embodiments of the invention, an internal diameter of the microtube is from about 50 nm to 60 about 20 micrometer.

According to some embodiments of the invention, the first and the second polymeric solutions are selected from the group consisting of: 10% poly (e-caprolactone) (PCL) in chloroform (CHCl₃) and dimethylforamide (DMF) (80:20 65 by weight) as the first polymeric solution and 4% poly (ethylene oxide) (PEO) in water (H₂O) and ethanol (60:40

4

by weight) as the second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as the first polymeric solution and 6% PEO in $\rm H_2O$ and ethanol (60:40 by weight) as the second polymeric solution, 9% PCL in CHCl₃ and DMF (90:10 by weight) as the first polymeric solution and 7% PEO in $\rm H_2O$ as the second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as the first polymeric solution and 9% poly(vinyl alcohol) (PVA) in water and ethanol (50:50 by weight) as the second polymeric solution, and 10% PCL in CHCl₃ and DMF (90:10 by weight) as the first polymeric solution and 4% (w/w) PEO in ethanol: $\rm H_2O$ (26:74 by weight) as a second polymeric solution.

According to some embodiments of the invention, the microtube is filled with a liquid.

According to some embodiments of the invention, the first and the second polymeric solutions are biocompatible.

According to some embodiments of the invention, the molecule-of-interest is attached to the coat over the internal surface of the shell.

According to some embodiments of the invention, the molecule-of-interest is attached to the shell of the microtube.

According to some embodiments of the invention, the molecule-of-interest comprises a polypeptide, a polynucle-otide, a carbohydrate, a small molecule, or any combination thereof.

According to some embodiments of the invention, the molecule-of-interest comprises a member of an affinity pair.

According to some embodiments of the invention, the polypeptide is an enzyme.

According to some embodiments of the invention, the enzyme is alkaline phosphatase (SEQ ID NO:1 or 8) or beta-galactosidase (SEQ ID NO:2 or 9).

According to some embodiments of the invention, the first polymeric solution comprises polyethylene glycol (PEG).

According to some embodiments of the invention, the shell comprises pores.

According to some embodiments of the invention, the shell prevents diffusion of the molecule-of-interest therethrough.

According to some embodiments of the invention, the substrate-of-interest comprises incorporating the substrate-of-interest in a synthesis reaction catalyzed by the molecule-of-interest.

According to some embodiments of the invention, the substrate-of-interest comprises incorporating the substrate-of-interest in a catabolism reaction catalyzed by the molecule-of-interest.

According to some embodiments of the invention, the method further comprising collecting the solution following the contacting.

According to some embodiments of the invention, the solution comprises blood.

According to some embodiments of the invention, the affinity pair is selected from the group consisting of an enzyme and a substrate, a hormone and a receptor, an antibody and an antigen, a polypeptide and a polynucleotide, a polypucleotide and a cognate polynucleotide, a polypeptide and a metal ion, a polypeptide and a carbohydrate.

According to some embodiments of the invention, a therapeutically effective amount of the molecule-of-interest is capable of treating a pathology in the subject.

According to some embodiments of the invention, the molecule-of-interest comprises a polypeptide, and whereas a therapeutically effective amount of the polypeptide is capable of treating a pathology in the subject.

According to some embodiments of the invention, the pathology is selected from the group consisting of a metabolic disorder, an endocrine disease, an autoimmune disease, and cancer

According to some embodiments of the invention, the polypeptide is selected from the group consisting of insulin (SEQ ID NO:6), phenylalanine hydroxylase (PAH) (SEQ ID NO:3), dystrophin (SEQ ID NO:4), beta-glucosidase (GBA) (SEQ ID NO:5), and ceruloplasmin ferroxidase (CP) (SEQ ID NO:7).

Unless otherwise defined, all technical and/or scientific terms used herein have the same meaning as commonly understood by one of ordinary skill in the art to which the invention pertains. Although methods and materials similar or equivalent to those described herein can be used in the practice or testing of embodiments of the invention, exemplary methods and/or materials are described below. In case of conflict, the patent specification, including definitions, will control. In addition, the materials, methods, and 20 examples are illustrative only and are not intended to be necessarily limiting.

BRIEF DESCRIPTION OF THE DRAWINGS

The patent or application file contains at least one drawing executed in color. Copies of this patent or patent application publication with color drawing(s) will be provided by the Office upon request and payment of the necessary fee.

Some embodiments of the invention are herein described, 30 by way of example only, with reference to the accompanying drawings. With specific reference now to the drawings in detail, it is stressed that the particulars shown are by way of example and for purposes of illustrative discussion of embodiments of the invention. In this regard, the description 35 taken with the drawings makes apparent to those skilled in the art how embodiments of the invention may be practiced.

In the drawings:

FIGS. 1A-D are images depicting high resolution scanning electron microscope (HRSEM) micrographs of (a) 40 Type 1 fibers with alkaline phosphatase (AP); (b) Type 2 fibers with AP; (c) Type 1 fibers with beta-galactosidase (β -GAL); and (d) Type 2 fibers with β -GAL;

FIGS. 2A-C are graphs (FIGS. 2a-b) and a picture (FIG. **2**C) depicting the progress of the AP reactions with time for 45 enzymes attached to (e.g., encapsulated within) the electrospun fibers and the free enzyme in the solution forming the coat over the internal surface of the shell (also referred to herein as a core solution) prior to the electrospinning. FIG. 2A—the progress of the alkaline phosphatase reaction over 50 2500 minutes; FIG. 2B—inset, the progress of the reaction through over the first 300 minutes; Note that the enzymatic reaction of the enzyme encapsulated within type 2 electrospun microtubes is faster than that of enzyme encapsulated within type 1 electrospun microtubes. FIG. 2C-a photo- 55 graph of a piece of mat (Type 1) immersed in the assay solution. The presence of the yellow reaction product, p-nitrophenol, is apparent; the reaction substrate was paranitrophenyl phosphate;

FIG. 3 is a histogram depicting the relative activity of AP 60 enzyme for different types of fibers (type 1 and type 2) and in dwelling buffers. Mat=the electrospun fibers (microtubes) with the attached (encapsulated) enzymes; rinsing buffer=the buffer used to only rinse the fibers, without any additional incubation time; 24 hrs. buffer=the buffer following incubation of the fibers therein for 24 hours; 72 hrs. buffer=the buffer following incubation of the fibers therein

6

for 72 hours; core solution—the enzyme in the core solution prior to the electrospinning process.

FIGS. 4A-B are graphs depicting the progress of the β -GAL reactions with time for the two types of electrospun fibers and the free enzyme in the core solution. The substrate was ortho-nitrophenyl galactoside. FIG. 4A—the progress of the β -GAL reactions over 5000 minutes as measured by the amount of ortho-nitrophenol generated; FIG. 4B—inset of the graph of FIG. 4a, the progress of the β -GAL reactions over the first 50 minutes;

FIG. 5 is a graph depicting the β -GAL reaction versus time for the mat and buffers for Type 2 fibers;

FIG. 6 is a histogram depicting the relative activity of the β -GAL and AP for different types of fibers;

FIGS. 7A-B are fluorescence microscope micrographs depicting Type 1 fibers with AP (FIG. 7A) and β -GAL (FIG. 7B). Size bars: 100 μ m (FIG. 7A) and 50 μ m (FIG. 7B).

FIG. 8 is a schematic illustration depicting the desorption process of the molecule-of-interest from the microtube of the invention. The molecule-of-interest (e.g., a protein, an enzyme) is attached to the coat over the internal surface of the shell. Following contacting the microtube with a solution, the solution enters the microtube via the pores (an exemplary pore is marked by arrow No. 3) by a capillary rise (see arrow No. 1) and gradually wets and fills the microtube inner volume. The desorption of the molecule-of-interest from the internal surface of the microtube shell (which depends mainly on the rate of the release of the molecule-of-interest from the polymer) is shown by arrow No. 2.

FIG. 9 depicts a multi-step enzymatic reaction performed using encapsulated molecules-of-interest (enzymes 1-4). Enzyme 1 (enz1) catalyzes the conversion of compound A to B; Enzyme 2 (enz2) catalyzes the conversion of compound B to C; Enzyme 3 (enz3) catalyzes the conversion of compound C to D; Enzyme 4 (enz4) catalyzes the conversion of compound D to E.

FIG. 10 is a schematic presentation depicting Atrazine degradation by the isolated *Pseudomonas* ADP enzymes: AtzA (atrazine chlorohydrolase, e.g., GenBank Accession No. NP_862474), AtzB (hydroxyatrazine hydrolase, e.g., GenBank Accession No. NP_862481), AtzC(N-isopropylammelide isopropylamino hydrolase, e.g., GenBank Accession No. NP_862508), AtzD (cyanuric acid amidohydrolase, e.g., GenBank Accession No. NP_862537), AtzE (biuret hydrolase, e.g., GenBank Accession No. NP_862538) and AtzF (allophanate hydrolase, e.g., GenBank Accession No. AAK50333) which are attached to the microtube of the invention.

DESCRIPTION OF SPECIFIC EMBODIMENTS OF THE INVENTION

The present invention, in some embodiments thereof, relates to methods of attaching a molecule-of-interest to a microtube and, more particularly, but not exclusively, to electrospun microtubes which include a molecule-of-interest attached thereto which can be used in various therapeutic, diagnostic, purification and synthesis applications.

Before explaining at least one embodiment of the invention in detail, it is to be understood that the invention is not necessarily limited in its application to the details set forth in the following description or exemplified by the Examples. The invention is capable of other embodiments or of being practiced or carried out in various ways.

While reducing the invention to practice, the present inventors have uncovered that a molecule-of-interest can be attached to electrospun microtubes. Thus, as is shown in

FIGS. 1a-b and described in Example 1 of the Examples section which follows, two types of electrospun microtubes containing active molecules-of-interest (e.g., enzymes such as alkaline phosphatase or beta-galactosidase) were formed: Type 1 microtubes which exhibit a non-porous shell, and 5 Type 2 microtubes which exhibit a porous shell due to the presence of PEG in the polymeric solution forming the shell. The enzymatic activity contained within the microtubes was at the same order of magnitude as that of the polymeric solution prior to the electrospinning process (FIGS. 2a-b, 10 **3.4***a*-*b* and **6**; Examples 1-3 of the Examples section which follows) indicating that surprisingly the process of production did not compromise the functionality of the delicate protein material incorporated into the tube. In addition, as is shown in FIG. 3 and described in Example 2 of the 15 Examples section which follows, both the porous and nonporous microtubes were capable of releasing enzymes attached thereto. Moreover, during the electrospinning process, some of the alkaline phosphatase enzyme migrated to the outer surface of the microtube shell and was released 20 process in which at least two polymeric solutions are electherefrom into the aqueous environment (FIG. 3, Example 2), while the β -GAL enzyme remained within the internal surface of the shell (FIG. 5, Example 3). In addition, as is further described in Example 3 of the Examples section which follows, the activity of the enzymes attached to the 25 internal surface of the shell was increased in the presence of a porous shell which enabled the passage of substrates therethrough (FIGS. 4a-b). These results support the use of the microtubes of the invention as micro-reactors (e.g., bioreactors) for various synthesis, hydrolysis, isolation and 30 purification reactions.

According to an aspect of the invention there is provided a method of attaching a molecule-of-interest to a microtube. The method is effected by co-electrospinning two polymeric solutions through co-axial capillaries, wherein a first poly- 35 meric solution of the two polymeric solutions is for forming a shell of the microtube and a second polymeric solution of the two polymeric solutions is for forming a coat over an internal surface of the shell, the first polymeric solution is selected solidifying faster than the second polymeric solu- 40 tion and a solvent of the second polymeric solution is selected incapable of dissolving the first polymeric solution and wherein the second polymeric solution comprises the molecule-of-interest, thereby attaching the molecule-of-interest to the microtube.

As used herein the term "microtube" refers to a hollow tube having an inner diameter of e.g., about 200 nm to about 50 μm and an outer diameter of e.g., about 0.5 μm to about 100 μm.

According to some embodiments of the invention the 50 thickness of the microtube shell can vary from a few nanometers to several micrometers, such as from about 100 nm to about 20 µm, e.g., from about 200 nm to about 10 µm, from about 100 nm to about 5 µm, from about 100 nm to about 1 µm, e.g., about 500 nm.

According to some embodiments of the invention the internal diameter of the microtube shell can vary from a few nanometers to several micrometers, such as from about 50 nm to about 50 µm, e.g., from about 100 nm to about 20 µm, from about 200 nm to about 10 µm, from about 500 nm to 60 about 5 μm, from about 1 μm to about 5 μm, e.g., about 3 μm.

According to some embodiments of the invention, the microtube may have a length which is from about 0.1 millimeter (mm) to about 20 centimeter (cm), e.g., from about 1-20 cm, e.g., from about 5-10 cm.

As used herein the term "attaching" refers to the binding of the molecule-of-interest to the polymer(s) comprised in

the microtube of the invention via covalent or non-covalent binding (e.g., via an electrostatic bond, a hydrogen bond, a van-Der Waals interaction) so as to obtain an absorbed. embedded or immobilized molecule-of-interest to the microtube of the invention.

According to some embodiments of the invention, the length (L) of the microtube can be several orders of magnitude higher (e.g., 10 times, 100 times, 1000 times, 10,000 times) than the microtube's diameter (D). Accordingly, a molecule-of-interest which is attached to such a microtube is referred to as being entrapped or encapsulated within the

According to some embodiments of the invention, covalent attachment of the molecule-of-interest can be via functional groups such as SH groups, amino groups, carboxyl groups which are added to the polymer(s) forming the microtube.

As used herein the phrase "co-electrospinning" refers to a trospun from co-axial capillaries (i.e., at least two capillary dispensers wherein one capillary is placed within the other capillary while sharing a co-axial orientation) forming the spinneret within an electrostatic field in a direction of a collector. The capillary can be, for example, a syringe with a metal needle or a bath provided with one or more capillary apertures from which the polymeric solution can be extruded, e.g., under the action of hydrostatic pressure, mechanical pressure, air pressure and/or high voltage.

The collector serves for collecting the electrospun element (e.g., the electrospun microtube) thereupon. Such a collector can be a rotating collector or a static (non rotating) collector. When a rotating collector is used, such a collector may have a cylindrical shape (e.g., a drum), however, the rotating collector can be also of a planar geometry (e.g., an horizontal disk). The spinneret is typically connected to a source of high voltage, such as of positive polarity, while the collector is grounded, thus forming an electrostatic field between the dispensing capillary (dispenser) and the collector. Alternatively, the spinneret can be grounded while the collector is connected to a source of high voltage, such as with negative polarity. As will be appreciated by one ordinarily skilled in the art, any of the above configurations establishes motion of a positively charged jet from the spinneret to the collector. Reverse polarity for establishing motions of a negatively charged jet from the spinneret to the collector are also contemplated.

For electrospinning, the first polymeric solution is injected into the outer capillary of the co-axial capillaries while the second polymeric solution is injected into the inner capillary of the co-axial capillaries. In order to form a microtube (i.e., a hollow structure, as mentioned above), the first polymeric solution (which is for forming the shell of the microtube) solidifies faster than the second polymeric solution (also referred herein as a core polymeric solution, and is for forming a coat over the internal surface of the shell). In addition, the formation of a microtube also requires that the solvent of the second polymeric solution be incapable of dissolving the first polymeric solution.

The solidification rates of the first and second polymeric solutions are critical for forming the microtube. For example, for a microtube of about 100 µm, the solidification of the first polymer (of the first polymeric solution) can be within about 30 milliseconds (ms) while the solidification of the second polymer (of the second polymeric solution) can be within about 10-20 seconds. The solidification may be a result of polymerization rate and/or evaporation rate.

TABLE 1-continued

According to some embodiments of the invention, the solvent of the first polymeric solution evaporates faster than the solvent of second polymeric solution (e.g., the solvent of the first polymeric solution exhibits a higher vapor pressure than the solvent of the second polymeric solution).

According to some embodiments of the invention, the rate of evaporation of the solvent of the first polymeric solution is at least about 10 times faster than that of the solvent of the second polymeric solution. The evaporation rate of the solvent of the first polymeric solution can be at least about 100 times faster or at least about 1000 times faster than the evaporation rate of the solvent of second polymeric solution. For example, the evaporation of chloroform is significantly faster than the evaporation of an aqueous solution (water) due to the high vapor pressure at room temperature of the chloroform (195 mmHg) vs. that of the aqueous solution (23.8 mmHg).

When selecting a solvent of the second polymeric solution which is incapable of dissolving the first polymeric solution (i.e., a non-solvent of the first polymeric solution), the polymer of the first polymeric solution can solidify (e.g., through precipitation) and form a strong microtube shell which does not collapse, and is characterized by an even thickness. According to some embodiments of the invention, the first polymeric solution (e.g., the solvent of the first polymer) is substantially immiscible in the solvent of the second polymeric solution.

The solvent of the second polymeric solution may evaporate while the polymer (of the second polymeric solution) forms a thin layer on the internal surface of the shell.

According to some embodiments of the invention, the solvent of the second polymeric solution is capable of 35 evaporating through the internal surface of the shell.

The flow rates of the first and second polymeric solutions can determine the microtube outer and inner diameter and thickness of shell. Non-limiting examples of microtubes generated by electrospinning using different flow rates are 40 shown in Table 1 hereinbelow.

Effect of the flow rates of the two polymeric solutions during

TABLE 1

System No.	System: First polymeric solution/ Second polymeric solution	Flow rates (ml/hr)	R Outer Fiber radius (µm)	d Shell thickness (µm)	V Voltage (kV)	Electro- static field kV/cm
M5	First	4	3.0-4.5	0.5 ± 0.1	8.5	0.43
	polymeric solution					
	Second polymeric solution	0.5				
M10	First	10	2.3-4.0	$1.0~\pm~0.1$	8	0.5
	polymeric solution					
	Second	0.3				
	polymeric solution					
M11	First	10	3-6	1.0 ± 0.1	9	0.56
	polymeric solution					

	fect of the flow ectrospinning of					
System No.	System: First polymeric solution/ Second polymeric solution	Flow rates (ml/hr)	R Outer Fiber radius (µm)	d Shell thickness (µm)	V Voltage (kV)	Electro- static field kV/cm
	Second polymeric solution	2				

Table 1: Electrospinning was performed with the following solutions: First polymeric solution (for forming the shell) was 10% PCL in CHCl₃/DMF (8:2 weight/weight); Second polymeric solution (for forming the coat) was 4% PEO in H₂O/E:tOH (6:4, weight/weight). PCL used was PCL 80 K; PEO used was PEO 600 K. The temperature during electrospinning was 22-26° C. The relative humidity during electrospinning was 58%, 52% and 53% for systems M5, M10 and M11, respectively. The flow rates were measured in milliliter per hour (ml/hr); the outer microtube radius (R) and the shell thickness (d) were measured in microns (µm). The voltage was measured in kilo volt (kV), and the electrostatic field was measured in kV per centimeter (cm). The resulting tubes were hollow (good tubes in systems M5 and M11, and mostly good in system M10).

As used herein the phrase "polymeric solution" refers to a soluble polymer, i.e., a liquid medium containing one or more polymers, co-polymers or blends of polymers dissolved in a solvent. The polymer used by the invention can be a natural, synthetic, biocompatible and/or biodegradable polymer.

The phrase "synthetic polymer" refers to polymers that are not found in nature, even if the polymers are made from naturally occurring biomaterials. Examples include, but are not limited to, aliphatic polyesters, poly(amino acids), copoly(ether-esters), polyalkylenes oxalates, polyamides, tyrosine derived polycarbonates, poly(iminocarbonates), polyorthoesters, polyoxaesters, polyamidoesters, polyoxaesters containing amine groups, poly(anhydrides), polyphosphazenes, and combinations thereof.

Suitable synthetic polymers for use by the invention can also include biosynthetic polymers based on sequences found in naturally occurring proteins such as those of collagen, elastin, thrombin, fibronectin, or derivatives thereof or, starches, poly(amino acids), poly(propylene fumarate), gelatin, alginate, pectin, fibrin, oxidized cellulose, chitin, chitosan, tropoelastin, hyaluronic acid, polyethylene, polyethylene terephthalate, poly(tetrafluoroethylene), polycarbonate, polypropylene and poly(vinyl alcohol), ribonucleic acids, deoxyribonucleic acids, polypeptides, proteins, polysaccharides, polynucleotides and combinations thereof.

The phrase "natural polymer" refers to polymers that are naturally occurring. Non-limiting examples of such polymers include, silk, collagen-based materials, chitosan, hyaluronic acid, albumin, fibrinogen, and alginate.

As used herein, the phrase "co-polymer" refers to a polymer of at least two chemically distinct monomers. Non-limiting examples of co-polymers include, polylactic acid (PLA)-polyethyleneglycol (PEG), polyethylene glycol terephthalate (PEGT)/polybutylene terephthalate (PBT), PLA-polyglycolic acid (PGA), PEG-polycaprolactone (PCL) and PCL-PLA.

As used herein, the phrase "blends of polymers" refers to the result of mixing two or more polymers together to create a new material with different physical properties.

The phrase "biocompatible polymer" refers to any polymer (synthetic or natural) which when in contact with cells, tissues or body fluid of an organism does not induce adverse effects such as immunological reactions and/or rejections,

cellular death and the like. It will be appreciated that a biocompatible polymer can also be a biodegradable polymer

According to some embodiments of the invention, the first and the second polymeric solutions are biocompatible.

Non-limiting examples of biocompatible polymers include polyesters (PE), PCL, Calcium sulfate, PLA, PGA, PEG, polyvinyl alcohol, polyvinyl pyrrolidone, Polytetra-fluoroethylene (PTFE, teflon), polypropylene (PP), polyvinylchloride (PVC), Polymethylmethacrylate (PMMA), 10 polyamides, segmented polyurethane, polycarbonate-urethane and thermoplastic polyether urethane, silicone-polyether-urethane, silicone-polycarbonate-urethane collagen, PEG-DMA, alginate, hydroxyapatite and chitosan, blends and copolymers thereof.

The phrase "biodegradable polymer" refers to a synthetic or natural polymer which can be degraded (i.e., broken down) in the physiological environment such as by proteases or other enzymes produced by living organisms such as bacteria, fungi, plants and animals. Biodegradability 20 depends on the availability of degradation substrates (i.e., biological materials or portion thereof which are part of the polymer), the presence of biodegrading materials (e.g., microorganisms, enzymes, proteins) and the availability of oxygen (for aerobic organisms, microorganisms or portions 25 thereof), lack of oxygen (for anaerobic organisms, microorganisms or portions thereof) and/or other nutrients. Examples of biodegradable polymers/materials include, but are not limited to, collagen (e.g., Collagen I or IV), fibrin, hyaluronic acid, polylactic acid (PLA), polyglycolic acid 30 (PGA), polycaprolactone (PCL), polydioxanone (PDO), trimethylene carbonate (TMC), polyethyleneglycol (PEG), collagen, PEG-DMA, alginate, chitosan copolymers or mix-

According to some embodiments, the polymeric solution 35 can be made of one or more polymers, each can be a polymer or a co-polymer such as described hereinabove.

According to some embodiments of the invention, the polymeric solution of the invention is a mixture of at least one biocompatible polymer and a co-polymer (either biodegradable or non-biodegradable).

12

According to some embodiments of the invention, the first polymeric solution for forming the shell can be made of a polymer such as poly (e-caprolactone) (PCL), polyamide, poly(siloxane), poly(silicone), poly(ethylene), poly(vinyl pyrrolidone), poly(2-hydroxy ethylmethacrylate), poly(Nvinyl pyrrolidone), poly(methyl methacrylate), poly(vinyl alcohol), poly(acrylic acid), poly(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(ethylene glycol), poly(methacrylic acid), polylactide, polyglycolide, poly(lactide-coglycolide), polyanhydride, polyorthoester, poly(carbonate), poly(acrylo nitrile), poly(ethylene oxide), polyaniline, polyvinyl carbazole, polystyrene, poly(vinyl phenol), polyhydroxyacid, poly(caprolactone), polyanhydride, polyhydroxyalkanoate, polyurethane, collagen, albumin, alginate, chitosan, starch, hyaluronic acid, and blends and copolymers thereof.

According to some embodiments of the invention, the second polymeric solution for forming the coat over the internal surface of the shell can be made of a polymer such as poly(acrylic acid), poly(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(ethylene glycol), poly (methacrylic acid), polylactide, polyglycolide, poly(lactide-coglycolide), polyanhydride, polyorthoester, poly(carbonate), poly(ethylene oxide), polyaniline, polyvinyl carbazole, polystyrene, poly(vinyl phenol), polyhydroxyacid, alginate, starch, hyaluronic acid, and blends and copolymers thereof.

During the formation of the microtube shell (e.g., following the solidification of the first polymeric solution) the second polymeric solution flows within the internal surface of the shell.

According to some embodiments of the invention, the second polymeric solution is selected capable of wetting the internal surface of the shell.

Various polymeric solutions which are known in the art as capable of wetting other polymeric surfaces (for forming the shell) can be used. Following is a non-limiting list of pairs of polymeric solutions in which the second polymeric solution is capable of wetting the internal surface of the shell formed by the first polymeric solution.

TABLE 2

Pairs of polymeric solutions for pro	ducing the microtube of the invention
First polymeric solution forming the shell	Second polymeric solution capable of wetting the internal surface of the shell
10% poly (e-caprolactone) (PCL); in chloroform (CHCl ₃) and dimethylforamide (DMF) (80:20 by weight)	4% poly(ethylene oxide) (PEO); in water (H_2O) and ethanol (60:40 by weight)
Nylon 6,6 in formic acid 7 to 12 wt %	4% poly(ethylene oxide) (PEO); in water (H ₂ O) and ethanol (60:40 by weight)
Poly(L-lactide-co-glycolide) (PLGA 10:90) in hexafluroisopropanol (HFIP) concentrations ranging from 2 to 7 weight % solution.	4% poly(ethylene oxide) (PEO) in water (H ₂ O) and ethanol (60:40 by weight)
Poly(L-lactide-co-glycolide) (PLGA 15:85) hexafluroisopropanol (HFIP) concentrations ranging from 2 to 7 weight % solution.	4% poly(ethylene oxide) (PEO); in water $(\mathrm{H_2O})$ and ethanol (60:40 by weight)
poly(lactide-co-glycolide) (PLGA; l-lactide/glycolide 50/50) 1,1,1,3,3,3-hexafluoro-2-propanol (HFIP) concentrations ranging from 2 to 7 weight % solution.	4% poly(ethylene oxide) (PEO); in water $(\mathrm{H_2O})$ and ethanol (60:40 by weight)
polyglycolide (PGA) in chloroform 3-10 weight % solution.	9% poly(vinyl alcohol) (PVA); in water and ethanol (50:50 by weight)
poly(L-lactide) (PLA) in chloroform 3-10 weight % solution.	9% poly(vinyl alcohol) (PVA); in water and ethanol (50:50 by weight)

TABLE 2-continued

Pairs of polymeric solutions for producing the microtube of the invention									
First polymeric solution forming the shell	Second polymeric solution capable of wetting the internal surface of the shell								
Segmented polyurethane in DMF and THF (80:20 by weight) Polyurethane in DMF and tetrahydrofuran, THF (80:20 by weight) PLGA (poly lactic-co-glycolic acid); in chloroform and DMSO (dimethyl sulfoxide) in chloroform and DMSO (80:20 by weight).	9% poly(vinyl alcohol) (PVA); in water and ethanol (50:50 by weight) 9% poly(vinyl alcohol) (PVA); in water and ethanol (50:50 by weight) 9% poly(vinyl alcohol) (PVA); in water and ethanol (50:50 by weight)								
10% PCL in CHCl ₃ /DMF (80:20 by weight) 9% PCL in CHCl ₃ /DMSO (90:10 by weight)	6% PEO in $\rm H_2O/EtOH$ (60:40 by weight) 7% PEO in $\rm H_2O$								
weight) 10% PCL in CHCl ₃ /DMF (80:20 by weight) 10% PCL 80 K CHCl ₃ :DMF (90:10 by weight) 10% PCL 80 K + 1% PEG 6 K CHCl ₃ :DMF (90:10 by weight)	9% PVA in ethanol/water (50:50 by weight) 4% (w/w) PEO 600 K; in ethanol:H ₂ O (26:74 by weight) 4% (w/w) PEO 600 K; in ethanol:H ₂ O (26:74 by weight)								

Table 2 (cont.). The polymers forming the solutions and the solvents are provided by weight ratios, i.e., a weight/weight (w/w) ratio.

and the second polymeric solutions are selected from the group consisting of: 10% poly (e-caprolactone) (PCL) in chloroform (CHCl₃) and dimethylforamide (DMF) (80:20 by weight) as the first polymeric solution and 4% poly (ethylene oxide) (PEO) in water (H₂O) and ethanol (60:40 30 by weight) as the second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as the first polymeric solution and 6% PEO in water and ethanol (60:40 by weight) as the second polymeric solution, 9% PCL in CHCl₃ and DMF (90:10 by weight) as the first polymeric solution and 35 7% PEO in water as the second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as the first polymeric solution and 9% poly(vinyl alcohol) (PVA) in water and ethanol (50:50 by weight) as the second polymeric solution and 10% PCL in CHCl₃ and DMF (90:10 by 40 weight) as the first polymeric solution and 4% (w/w) PEO in ethanol:H₂O (26:74 by weight) as a second polymeric solution.

To enable a flow of a liquid-of-interest within the microtube, i.e., along the coat polymer covering the internal 45 surface of the shell (which originates from the second polymer solution), the surface (thin film) formed by the coat polymer should be designed such that it can be wetted by the liquid-of-interest. The ability to wet (wettability) polymer films by liquids is known in the art. For example, silicone oil 50 or water can wet a surface made of a PEO polymer. It will be appreciated that the wettability of the coat polymer covering the internal surface of the shell can be controlled (e.g., improved) for example by attaching functional groups such as hydroxyl groups (OH) which increase the hydro- 55 philicity of the coat by a plasma treatment [see Thurston R M, Clay J D, Schulte M D, Effect of atmospheric plasma treatment on polymer surface energy and adhesion, Journal of Plastic Film & Sheeting 23 (1): 63-78 January 2007; which is incorporated within by reference].

As is further discussed hereinabove and in the Examples section which follows, for certain applications the microtube shell may comprise pores, thus creating a "breathing" tube. Methods of forming "breathing" microtube (i.e., microtubes with pores in the shell thereof) are described in PCT/ 65 IB2007/054001 to the present inventors, which is fully incorporated herein by reference. Briefly, "breathing" tubes

According to some embodiments of the invention, the first 25 can be formed by the inclusion of a high percent (e.g., at least 80%) of a volatile component such as tetrahydrofuran (THF), chloroform, acetone, or trifluoroethanol (TFE) in the first polymeric solution forming the shell, and/or by the inclusion of a water-soluble polymer such as polyethylene glycol (PEG) in the first polymeric solution forming the shell so that the first polymeric solution comprises a blend of polymers in which one is water-soluble and the other is water-insoluble (e.g., a blend of PEG and PCL). Alternatively, "breathing" microtubes can be formed by inducing pores in the shell after the completion of the electrospinning process, essentially as described in PCT WO 2006/106506 to the present inventors, which is fully incorporated herein by reference, such as by passing an electrical spark or a heated puncturing element through the electrospun shell, or by using a pulsed or continuous laser beam through the electrospun shell.

14

According to some embodiments of the invention, the first polymeric solution comprises PEG for inducing pores in the shell. For example, to generate pores greater (>) than 150 nm in diameter, the first polymeric solution may include about 4% PEG MW 35 kDa. Similarly, to generate pores smaller (<) than 150 nm in diameter, the first polymeric solution may include about 2% PEG MW 6 kDa.

The microtube shell of the invention can be designed such that it enables the passage of certain molecules (e.g., a substrate of an enzyme) while preventing the passage of other molecules (e.g., a certain enzyme), depending on the geometry (pore size) and/or the electrical charge of the molecules with respect to the geometry (length and radius), surface energy, electrical charge of the nanopore(s) of the shell, and the viscosity and surface tension of the liquid containing the molecules (e.g., the substrate of the enzyme). In addition, the porosity and pore size of the shell can control the release of the molecule-of-interest which is attached to the microtube. For example, a higher porosity and/or pore size can result in increased rate of release of the moleculeof-interest.

Alternatively, the microtube shell can be made such that it prevents diffusion or any passage of the molecule-ofinterest therethrough (i.e., substantially devoid of pores, or with pores smaller than the molecule-of-interest).

As mentioned, the second polymeric solution comprises the molecule-of-interest. Such a molecule (or molecules) can be any naturally occurring or synthetic molecule such as a polypeptide, a polynucleotide, a carbohydrate or a polysaccharide, a lipid, a drug molecule, a small molecule (e.g., a nucleotide base, an amino acid, a nucleotide, an antibiotic, a vitamin or a molecule which is smaller than 0.15 kDa), or any combination thereof. The molecule-of-interest can be produced by recombinant DNA technology or by known synthesis methods such as solid phase.

According to some embodiments of the invention, the molecule-of-interest comprises a polypeptide such as an enzyme. Such polypeptides (e.g., enzymes) can be naturally occurring (e.g., mammals such as primates, rodents and Homo sapiens, plants, fungi, protozoa, bacteria and viruses) 15 or synthetic (e.g., derived from in vitro evolution) and can be selected according to the desired application.

The following non-limiting list of enzymes can be attached to the microtube of the invention: DNA polymerase (EC 2.7.7.7), DNase (EC 3.1.11.4), RNA polymerase (EC 20 2.7.7.6), DNA ligase (EC 6.5.1.1), RNA ligase (EC 6.5.1.3), alcohol dehydrogenase (EC 1.1.1.1), homoserine dehydrogenase (EC 1.1.1.3), acetoin dehydrogenase (EC 1.1.1.5), glycerol dehydrogenase (EC 1.1.1.6), L-xylulose reductase (EC 1.1.1.10), L-arabinitol 2-dehydrogenase (EC 1.1.1.13), 25 L-iditol 2-dehydrogenase (EC 1.1.1.14), mannitol-1-phosphate 5-dehydrogenase (EC 1.1.1.17), mannitol 2-dehydrogenase (EC 1.1.1.138), glucose oxidase (EC 1.1.3.4), L-sorbose oxidase (EC 1.1.3.11), lactate-malate transhydrogenase (EC 1.1.99.7), formaldehyde dehydrogenase (EC 1.2.1.1), 30 aryl-aldehyde dehydrogenase (EC 1.2.1.29), aldehyde oxidase (EC 1.2.3.1), pyruvate synthase (EC 1.2.7.1), cortisone α-reductase (EC 1.3.1.4), lathosterol oxidase (EC 1.3.3.2), D-proline reductase (EC 1.4.4.1), dihydrofolate reductase (EC 1.5.1.3), methylenetetrahydrofolate reductase (NA- 35 DPH) (EC 1.5.1.20), cystine reductase (NADH) (EC 1.6.1.4), cob(II) alamin reductase (EC 1.6.99.9), sulfite reductase (EC 1.8.1.2), cytochrome-c oxidase (EC 1.9.3.1), NADH peroxidase (EC 1.11.1.1), homogentistate 1,2-dioxygenase (EC 1.13.11.5), Photimus-luciferin 4-monooxy- 40 genase (1.13.12.7), anthranilate 3-monooxygenase (EC 1.14.13.35), steroid 9α -monooxygenase (EC 1.14.99.25), mercury(II) reductase (EC 1.16.1.1), nicotinamide N-methyltransferase (EC 2.1.1.1), thymidylate synthase (EC 2.1.1.45), site-specific DNA-methyltransferase (adenine- 45 specific) (EC 2.1.1.72), tryptophan 2-C-methyltransferase (EC 2.1.1.106), glycine formininotransferase (EC 2.1.2.4), aspartate carbamoyltransferase (EC 2.1.3.2), transaldolase (EC 2.2.1.2), arylamine N-acetyltransferase (EC 2.3.1.5), arginine N-succinyltransferase (EC 2.3.1.109), phosphory- 50 lase (EC 2.4.1.1), glycosaminoglycan galactosyltransferase (EC 2.4.1.74), thymidine phosphorylase (EC 2.4.2.4), β-galactoside α -2,6-sialyltransferase (EC 2.4.99.1), galactose-6sulfurylase (EC 2.5.1.5), aspartate transaminase (2.6.1.1), hexokinase (EC 2.7.1.1), choline kinase (EC 2.7.1.32), 55 acetate kinase (EC 2.7.2.1), creatine kinase (EC 2.7.3.2), adenylate kinase (EC 2.7.4.3), nucleotide pyrophosphokinase (EC 2.7.6.4), sulfate adenylyltransferase (ADP) (EC 2.7.7.5), aryl sulfotransferase (EC 2.8.2.1), carboxylesterase (3.1.1.1), acetyl-CoA hydrolase (EC 3.1.2.1), alkaline phos- 60 phatase (3.1.3.1), phosphodiesterase I (EC 3.1.4.1), dGT-Pase (EC 3.1.5.1), steryl-sulfatase (EC 3.1.6.2), exodeoxyribonuclease I (EC 3.1.11.1), ribonuclease T1 (EC 3.1.27.3), α-amylase (EC 3.2.1.1), purine nucleosidase (EC 3.2.2.1), epoxide hydrolase (EC 3.3.2.3), lysyl aminopeptidase (EC 3.4.11.15), carboxypeptidase A2 (EC 3.4.17.15), trypsin (EC 3.4.21.4), glutaminase (EC 3.5.1.2), barbiturase (EC

16

3.5.2.1), ATP deaminase (EC 3.5.4.18), inorganic pyrophosphatase (EC 3.6.1.1), oxaloacetase (EC 3.7.1.1), oxalate decarboxylase (EC 4.1.1.2), mandelonitrile lyase (EC 4.1.2.10), isocitrate lyase (4.1.3.1), fumarate hydratase (EC 4.2.1.2), pectate lyase (EC 4.2.2.2), histidine ammonia-lyase (EC 4.3.1.3), cvanate lyase (4.3.99.1), cysteine lyase (EC 4.4.1.10), DDT-dehydrochlorinase (EC 4.5.1.1), adenylate cyclase (EC 4.6.1.1), alanine racemase (5.1.1.1). tartrate epimerase (EC 5.1.2.5), retinal isomerase (EC 5.2.1.3), L-rhamnose isomerase (EC 5.3.1.14), prostaglandin-D synthase (EC 5.3.99.2), phosphoglucomutase (EC 5.4.2.2), lanosterol synthase (EC 5.4.99.7), DNA topoisomerase (EC 5.99.1.2), tyrosine-tRNA ligase (EC6.1.1.1), acetate-CoA ligase (EC 6.2.1.1), acetylcholinesterase (EC 3.1.1.7), butyrylcholinesterase (EC 3.1.1.8) and glutathione synthase (EC 6.3.2.3).

According to an embodiment of the invention, the enzyme which is attached to the microtube is alkaline phosphatase (e.g., SEQ ID NO:1 or 8; EC 3.1.3.1) or beta-galactosidase (e.g., SEQ ID NO:2 or 9; EC 3.2.1.23).

The term "polynucleotide" as used herein refers to a single stranded or double stranded oligomer or polymer of ribonucleic acid (RNA) or deoxyribonucleic acid (DNA) or mimetics thereof. This term includes polynucleotides composed of naturally-occurring bases, sugars and covalent internucleoside linkages (e.g., backbone) as well as polynucleotides having non-naturally-occurring portions which function similarly to respective naturally-occurring portions.

The polynucleotide which is attached to the microtube of the invention can be generated according to any oligonucleotide synthesis method known in the art such as enzymatic synthesis, liquid phase or solid phase synthesis (using a commercially available equipment from, for example, Applied Biosystems). Equipment and reagents for executing solid-phase synthesis are commercially available from, for example, Applied Biosystems. Any other means for such synthesis may also be employed; the actual synthesis of the oligonucleotides is well within the capabilities of one skilled in the art and can be accomplished via established methodologies as detailed in, for example, "Molecular Cloning: A laboratory Manual" Sambrook et al., (1989); "Current Protocols in Molecular Biology" Volumes I-III Ausubel, R. M., ed. (1994); Ausubel et al., "Current Protocols in Molecular Biology", John Wiley and Sons, Baltimore, Md. (1989); Perbal, "A Practical Guide to Molecular Cloning", John Wiley & Sons, New York (1988) and "Oligonucleotide Synthesis" Gait, M. J., ed. (1984) utilizing solid phase chemistry, e.g. cyanoethyl phosphoramidite followed by deprotection, desalting and purification by for example, an automated trityl-on method or HPLC. Liquid phase synthesis of oligonucleotides can be performed using methods known in the art (see for example, Bonora G M, et al., 1998, Biol. Proced. Online. 1: 59-69; Padiya K J and Salunkhe M M., 2000, Bioorg. Med. Chem. 8: 337-42). It will be appreciated that for the preparation of multiple labeled polynucleotides, a large scale oligonucleotide synthesis can be utilized essentially as described elsewhere (Anderson N G. et al., Appl Biochem Biotechnol. 1995 July-September; 54 (1-3):19-42; Rahmann S., Proc IEEE Comput Soc Bioinform Conf. 2002; 1:54-63).

Additionally or alternatively, the polynucleotide which is attached to the microtube of the invention can be generated by recombinant DNA techniques using any known DNA replication or transcription system (e.g., using bacterial cells, eukaryotic cells).

As mentioned, the molecule-of-interest can be a drug molecule. Such a drug can be any synthetic, chemical or biological molecule.

Non-limiting examples of biological drug molecules include antisense oligonucleotides, Ribozymes, DNAzymes, 5 siRNA, receptor agonists, antagonists, hormones, growth factors and antibodies. Non-limiting examples of which chemical drug molecules include chemotherapy agents, Paclitaxel (Taxol®), radiation seed particles (e.g., see Hypertext Transfer Protocol://World Wide Web (dot) oncura (dot) com), as well as natural or synthetic vitamins.

The molecule-of-interest which is attached to the microtube of the invention can be labeled. Such a label can be an intrinsic property of the molecule-of-interest (e.g., as in the $_{15}$ case of green fluorescent protein) or can be a label which is attached to the molecule-of-interest using known methods. For example, the label can be a fluorescent labeling in which a fluorophore (i.e., an entity which can be excited by light to emit fluorescence) or a radio-isotope is conjugated via a 20 linker or a chemical bond to the molecule-of-interest. Alternatively, the molecule-of-interest can be indirectly labeled via a covalently conjugated enzyme (e.g., horse radish peroxidase) and a covalently conjugated substrate (e.g., o-phenylenediamine) which upon interaction therebetween 25 yield a colorimetric or fluorescent color.

The molecule-of-interest can also comprise a member of an affinity pair, which is capable of reversibly or nonreversibly binding with high affinity (e.g., less than 10^{-7} M, e.g., less than 10^{-8} M, less than 10^{-9} , less than 10^{-10} M) to 30 a specific molecule. For example, the affinity pair can be an enzyme-substrate pair, a polypeptide-polypeptide pair (e.g., a hormone and receptor, a ligand and receptor, an antibody and an antigen, two chains of a multimeric protein), a with biotin, enzyme-substrate), a polynucleotide and its cognate polynucleotide such as two polynucleotides forming a double strand (e.g., DNA-DNA, DNA-RNA, RNA-DNA), a polypeptide-polynucleotide pair (e.g., a complex formed of a polypeptide and a DNA or RNA e.g., aptamer), a 40 polypeptide-metal pair (e.g., a protein chelator and a metal ion), a polypeptide and a carbohydrate (leptin-carbohydrate), and the like.

The molecule-of-interest, which is comprised within the second polymeric solution, can be attached to the coat over 45 the internal surface of the shell. For example, as shown in FIG. 5 and described in Example 3 of the Examples section which follows, most of the β -GAL enzymatic activity was detected inside the microtube, demonstrating the attachment of the enzyme to the coat over the internal surface of the 50

During the electrospinning process some molecules-ofinterest which are comprised within the second polymeric solution may migrate to the outer surface of the shell (i.e., mixed with the first polymeric solution) depending on their 55 charge state, size and geometry. For example, as shown in FIG. 3 and described in Example 2 of the Examples section which follows, some of the alkaline phosphatase activity was detected in the rinsing buffer of the microtube.

According to some embodiments of the invention attach- 60 ment of the molecule-of-interest is performed following microtube formation. For example, the microtube can be soaked with a solution containing the molecule-of-interest. The molecule-of-interest can diffuse through the shell pores and enter the inner lumen of the microtube. In addition, the 65 microtube can be covalently attached to the molecule-ofinterest (e.g., via SH groups).

18

Regardless of the method of production, the present invention provides a microtube which comprises an electrospun shell, an electrospun coat over an internal surface of the shell and a molecule-of-interest attached to the microtube.

As used herein, the phrase "electrospun shell" refers to a hollow element of a tubular shape, made of one or more polymers, produced by the process of electrospinning as detailed above.

As used herein the phrase "electrospun coat" refers to a thin layer covering the internal surface of the shell of the microtube of the invention which is made of one or more polymers by the process of electrospinning as detailed above.

One of ordinary skill in the art will know how to distinguish an electrospun object from objects made by means which do not comprise electrospinning by the high orientation of the macromolecules, the skin (e.g., shell) morphology, and the typical dimensions of the microtube which are unique to electrospinning.

The microtube of the invention can be an individual (e.g., single or separated) microtube or can form part of a plurality (e.g., an aligned array) of microtubes which can be either connected to each other or separated (as single, not-connected microtubes).

For the production of a single microtube a fork like clip is attached to the edge of the rotating disk. The disk is rotated for 1-2 seconds and individual microtubes are formed between the sides of the clip. In a similar way individual electrospun fibers were collected (see E. Zussman, M. Burman, A. L. Yarin, R. Khalfin, Y. Cohen, "Tensile Deformation of Electrospun Nylon 6,6 Nanofibers," Journal of Polymer Science Part B: Polymer Physics, 44, 1482-1489, 2006, herein incorporated by reference in its entirety).

Alternatively, when using a rotating collector, a plurality polypeptide-small molecule pair (e.g., avidin or streptavidin 35 of microtubes can be formed and collected on the edge of the collector as described elsewhere for electrospun fibers (A. Theron, E. Zussman, A. L. Yarin, "Electrostatic field-assisted alignment of electrospun nanofibers", Nanotechnology J., 12, 3: 384-390, 2001; herein incorporated by reference in its entirety).

> The plurality of microtubes can be arranged on a single layer, or alternatively, the plurality of microtubes define a plurality of layers hence form a three dimensional structure. The microtubes can have a general random orientation, or a preferred orientation, as desired. For example, when the fibers are collected on a cylindrical collector such as a drum, the microtubes can be aligned predominantly axially or predominantly circumferentially. Different layers of the electrospun microtubes can have different orientation characteristics. For example, without limiting the scope of the present invention to any specific ordering or number of layers, the microtubes of a first layer can have a first predominant orientation, the microtubes of a second layer can have a second predominant orientation, and the microtubes of third layer can have general random orientation.

> The microtube of the invention can be available as a dry fibrous mat(s) (e.g., as spun dry microtubes) or as a wetted mat(s) (e.g., following immersing or filling the microtube with a liquid).

> The microtube of the invention which is attached to the molecule-of-interest may be configured as or in a microfluidics device. "Lab-on-a-chip" is described in a series of review articles [see for example, Craighead, H. Future lab-on-a-chip technologies for interrogating individual molecules. Nature 442, 387-393 (2006); deMello, A. J. Control and detection of chemical reactions in microfluidic systems. Nature 442, 394-402 (2006); El-Ali, J., Sorger, P. K. &

Jensen, K. F. Cells on chips. Nature 442, 403-411 (2006); Janasek, D., Franzke, J. & Manz, A. Scaling and the design of miniaturized chemical-analysis systems. Nature 442, 374-380 (2006); Psaltis, D., Quake, S. R. & Yang, C. H. Developing optofluidic technology through the fusion of microfluidics and optics. Nature 442, 381-386 (2006); Whitesides, G. M. The origins and the future of microfluidics. Nature 442, 368-373 (2006); Yager, P. et al. Microfluidic diagnostic technologies for global public health. Nature 442, 412-418 (2006)] each of which is fully incorporated herein by reference].

According to some embodiments of the invention, the liquid which fills in, flows in or surrounds the microtube enables the desorption (detachment) of the molecule-ofinterest from the microtube (e.g., from the polymer included in the coat over the internal surface of the shell). According to of some embodiments of the invention the desorption process facilitates the interaction between the molecule-ofinterest and a substrate. According to some embodiments of 20 the invention the desorption process enables the flow and/or the release of the molecule-of-interest within and/or from the microtube.

According to some embodiments of the invention, the molecule-of-interest which is attached to the microtube of 25 the invention remains active, i.e., maintains the activity, or at least a portion thereof, which it possessed prior to the attachment (e.g., of the same molecule-of-interest before electrospinning, or when not-attached to the microtube). The term "activity" as used herein refers to any of a catalytic 30 activity, kinetics, and/or affinity to a substrate, a ligand or an affinity member of the molecule. Such an activity can be any biological activity such as catalysis, binding (with a specific affinity), hybridization, chelation, degradation, synthesis, catabolism, hydrolysis, polymerization, transcription, and 35

As used herein the phrase "at least a portion of the activity" refers to at least about 10%, at least about 20-50%, e.g., more than about 50%, e.g., more than about 60%, e.g., more than about 70%, e.g., more than about 75%, e.g., more 40 than about 80%, e.g., more than about 90%, e.g., more than about 95% of the activity which the molecule-of-interest possessed prior to the attachment to the microtube.

For example, as mentioned before and described in the Examples section which follows, the enzymes contained 45 within the microtubes preserved the specific activity to their substrates at a kinetic which is comparable (i.e., within the same order of magnitude) to that of the enzyme in the polymeric solution prior to electrospinning.

The microtube of the invention which is attached to an 50 active molecule-of-interest can be used in various applications which require the attachment of active molecules (e.g., enzymes, DNA, RNA) to a support and optionally also the controlled release therefrom.

microtube of the invention is attached to more than one type of molecule-of-interest. The combination of molecules can be selected according to the intended use. For example, several molecules (e.g., enzymes) which are involved in complex reactions (e.g., processing of a substrate or a 60 mixture of substrates) can be used.

Thus, according to an aspect of the invention, there is provided a method of processing a substrate-of-interest. The method is effected by contacting the substrate-of-interest with the microtube of the invention, wherein the molecule- 65 of-interest is capable of processing the substrate, thereby processing the substrate-of-interest.

20

As used herein the term "processing" refers to a catalytic activity performed by the molecule-of-interest which is attached to the microtube on its cognate substrate.

According to some embodiments of the invention, such a process can concomitantly incorporate of the substrate-ofinterest in a synthesis reaction catalyzed by the moleculeof-interest.

For example, the microtube of some embodiments the invention can be used as a micro-reactor (e.g., bioreactor) for chemical transition reactions requiring high concentrations of several enzymes. As described in Example 4 of the Examples section which follows and schematically illustrated in FIG. 9, the microtube of the invention can be attached to certain molecules (enzymes in this case), which together catalyze a multi-step synthesis reaction (e.g., cascade) which converts an initial substrate (e.g., compound A) to an end-product (e.g., compound E). As mentioned, the selective shell of the microtube can be designed such that it prevents the leakage (escape by diffusion) of the intermediate compounds (e.g., compounds B, C and D) therethrough and thus enables sufficiently high concentrations of such compounds as needed for the synthesis of the end product. The local concentration of the intermediate molecules formed from the initial substrate (entrapped at the time of spinning or externally added after electro spinning to the formed microtube) are about 2-10 orders of magnitude greater than the concentrations formed in an open system. Thus, the microtube of some embodiments of the invention exhibits a great kinetic advantage in multi-step reactions as compared to an open system. Microtubes are in this way similar to living cells which function on the same principle.

For example, to synthesize an indole-glycerol phosphate, an intermediate compound in tryptophan synthesis within cells, a microtube of some embodiments of the invention can be attached to the enzymes anthranilate-phosphoribosyl transferase (EC 2.4.2.18), phosphoribosylanthranilate isomerase (EC 5.3.1.24) and indole-3-glycerol-phosphate synthase (EC 4.1.1.18), and the reaction commences when anthranilate and phosphoribosyl-pyrophosphate interact with the attached enzymes. The substrates (anthranilate and phosphoribosyl-pyrophosphate) can be either added externally to the reaction medium (within which the microtube is placed) or can be attached to the microtube by mixing them within the second polymeric solution. When the substrates are supplied externally, pores of about 2-20 nm in diameter should exist in the shell to allow diffusion of anthranilate and phosphoribosyl-pyrophosphate therethrough.

According to some embodiments of the invention, such a process can be the incorporation of the substrate-of-interest in a catabolism reaction catalyzed by the molecule-of-

A catabolism reaction can be the degradation (e.g., by According to some embodiments of the invention, the 55 hydrolysis) of a toxic molecule for the purpose of detoxification (e.g., detoxifying water) or decomposition of an unwanted molecule. Examples include, but are not limited to, the removal of the chlorine entity from atrazine (see FIG. 10) and the degradation of cyanide resulting from silver mining.

According to an aspect of the invention, there is provided a method of depleting a molecule from a solution. The method is effected by contacting the solution with the microtube of the invention, wherein the member of the affinity pair (which is attached to the microtube) is selected capable of binding the molecule (which is to be removed), thereby depleting the molecule from the solution.

According to an embodiment of the invention, the method further comprising collecting the solution following the contacting.

As used herein the phrase "depleting" refers to removing an amount e.g., at least about 50%, at least about 60%, at 5 least about 70%, at least about 80%, at least about 90%, at least about 95%, e.g., 99%, e.g., 100% of the molecule from the solution.

According to some embodiments of the invention, the depletion (removal) of the molecule from the solution is effected within a short time period, such as within minutes (e.g., 1-30 minutes), hours (e.g., 1-10 hours) or several days (e.g., 1-5 days).

As used herein the phrase "contacting" refers to enabling the interaction between the molecule and the member of the affinity pair, which is attached to the microtube, for a time period which is sufficient for depleting the molecule from the solution. Such a contact can take place while the solution is passing through (e.g., via capillary forces) the end(s) of the hollow structure of the microtube and/or through the 20 shell pores. Additionally or alternatively, such a contact between the molecule and the member of the affinity pair can take place by incubating the microtube in the solution (e.g., by placing the microtube in a container including the solution).

The solution can be any water-based solution which includes inorganic or organic molecules, such as a biological sample or a sample from a non-living source such as stream or ocean waters. As used herein the phrase "biological sample" refers to any sample derived from a living organism 30 such as plant, bacteria or mammals, and can include cells or alternatively be cell-free (i.e., include only a biological fluid). For example, a biological sample of an individual can include body fluids such as blood or components thereof (e.g., white blood cells, red blood cells, coagulation factors, 35 leukocytes, neutrophils, serum, plasma), cerebrospinal fluid, urine, lymph fluids, and various external secretions of the respiratory, intestinal and genitourinary tracts, tears, saliva, milk, amniotic fluid and chorionic villi, a tissue biopsy, a tissue section, a malignant tissue, and the like. The sample 40 can be derived from the individual and be further tested in vitro or ex vivo, or alternatively, can be not physically removed from the subject (e.g., for in situ detection and/or

According to some embodiments of the invention, the 45 solution is an aqueous solution such as a drinking water, a groundwater and/or an industrial waste water. According to some embodiments of the invention, the microtube of the invention forms part of an aqueous system designed for treatment of the aqueous solution (e.g., for depleting, eliminating or removing toxic moieties therefrom).

For example, to remove a certain metal ion (e.g., copper, gold, nickel, zinc, lead, mercury, cadmium, silver, iron, manganese, palladium, and platinum) from water, the microtube of the invention can be attached to a water soluble 55 ethylene dichloride ammonia polymer, which contains dithiocarbamate salt groups and is capable of chelating the metal ion (U.S. Pat. No. 5,346,627). Thus, by contacting the water with the microtube the ethylene dichloride ammonia polymer binds to the metal ion and removes it from the water. Water collected after being in contact with the microtube is substantially devoid of the metal ion. Alternatively, these metal ions may be removed by attaching a protein chelator of such metal ions to the microtube.

Alternatively, to remove a ligand (e.g., a hormone, a 65 substrate, a co-factor or a vitamin such as biotin) from a solution containing a biological sample, the microtube can

22

be attached to a polypeptide which is member of an affinity pair such as an enzyme, a hormone or streptavidin, and following contacting the solution with the microtube, the ligand remains attached to the microtube while the solution is substantially devoid of the ligand (e.g., includes less than 0.5%, e.g., less than 0.1%, e.g., less than 0.01% of the ligand).

According to some embodiments of the invention, the molecule which is to be removed from the solution comprises an antigen and the member of the affinity pair comprises the antibody capable of specifically binding the antigen.

For example, the microtube of the invention can be used to remove virus particles from a blood sample. Briefly, an anti-virus antibody (e.g., anti-HIV antibodies such as those described in Tullis, R H., et al., Therapeutic Apheresis and Dialysis, 6: 213-220) can be attached to the microtube and a blood sample containing the virus particles (e.g., HIV particles) can be in contact with the microtube such that the virus particles bind to their respective antibodies and the collected blood sample (after being in contact with the microtube) is substantially devoid of the viral particles.

The member of the affinity pair which is attached to the microtube of the invention can be also used to isolate a molecule from a solution.

According to an aspect of the invention, there is provided a method of isolating a molecule from a solution. The method is effected by: (a) contacting the solution with the microtube of the invention under conditions which allow binding of the molecule to the microtube via the member of the affinity pair which is selected capable of binding the molecule, and (b) eluting the molecule from the microtube, thereby isolating the molecule from the solution.

As used herein the term "isolating" refers to physically separating the molecule from the solution or its other components by binding the molecule to the member of the affinity pair that is attached to the microtube and eluting the bound molecule therefrom. As used herein the term "eluting" refers to dissociating the bound molecule from the microtube. Those of skills in the art are capable of adjusting the conditions required for eluting (e.g., releasing) the molecule from the microtube and/or separating the molecule from the other member of the affinity pair.

As is further described in Example 5 of the Examples section which follows, the present inventors have envisaged the use of the microtube of the invention, which is attached to a member of an affinity pair, as a biosensor, for the detection of molecules in a sample. Such a biosensor can be advantageous over known open field biosensors (e.g., sensors in which the member of the affinity pair is conjugated to a solid support not having a tubular structure, such as a flat support) due to the increased ratio between the size of the microtube surface (which attaches the member of the affinity pair) and the volume of the sample being in contact therewith

According to an aspect of the invention, there is provided a method of detecting a presence of a molecule in a sample. The method is effected by (a) contacting the sample with the microtube of the invention, wherein the member of the affinity pair is selected capable of binding the molecule, and; (b) detecting binding of the molecule by the member of the affinity pair, thereby detecting the presence of a molecule in the sample.

As used herein the phrase "detecting binding" refers to identifying a change in the concentration, conformation, spectrum or electrical charge of the molecule in the sample and/or the member of the affinity pair that is attached to the

microtube following the binding therebetween. Identification of such binding can be performed using methods known in the art such as following the fluorescence or the color of the sample, radioactivity in the sample, the electrical conductivity of the sample and the like.

As mentioned hereinabove and described in Example 2 of the Examples section which follows, the microtube of the invention can release the molecule-of-interest attached thereto (a releasing apparatus).

The microtube of some embodiments the invention (e.g., 10 a microtube made of biocompatible polymers) can be implanted in a subject in need thereof.

As used herein the phrase a "subject in need thereof" refers to any animal subject e.g., a mammal, e.g., a human being which suffers from a pathology (disease, disorder or 15 condition) which can be treated by the molecule that is attached to or flows through the microtube of the invention.

The term "treating" as used herein refers to inhibiting, preventing or arresting the development of a pathology and/or causing the reduction, remission, or regression of a 20 pathology. Those of skill in the art will understand that various methodologies and assays can be used to assess the development of a pathology, and similarly, various methodologies and assays may be used to assess the reduction, remission or regression of a pathology.

Methods of implanting grafts such as the microtube of the invention into a subject are known in the art. For example, the microtube can be implanted subcutaneously, intradermally, or into any body cavity (e.g., abdomen), as well as into the vascular system (using e.g., a hollow catheter 30 delivery system). Alternatively, the microtube of the invention can be connected to a body conduit (e.g., a blood vessel such as a vein or an artery) such that it enables the flow of a fluid therethrough.

For example, the microtube of the invention which is 35 capable of depleting a molecule from a solution as described above, can be connected to a blood vessel of the subject. For example, the proximal end of the microtube (or of a plurality of microtubes) can be connected to a feeding blood vessel and the distal end of the microtube(s) can be connected to a 40 receiving blood vessel. Such a configuration can be used, for example, for hemodialysis and depletion of a specific molecule (e.g., a virus particle such as HIV, hepatitis virus such as HCV) from the blood stream of the subject.

In addition, a microtube which is attached to a drug 45 molecule can be implanted in a subject in need thereof to thereby release a therapeutically effective amount of the drug to cells of the subject.

As used herein the phrase "therapeutically effective amount" means an amount of the molecule-of-interest (e.g., 50 the drug, the active molecule) effective to prevent, alleviate or ameliorate symptoms of a pathology or prolong the survival of the subject being treated. Determination of a therapeutically effective amount is well within the capability of those skilled in the art.

For example, in case the molecule-of-interest comprises a polynucleotide, such a polynucleotide can be used in gene therapy applications to either increase an expression level or activity of a desired polypeptide needed for treating the pathology, or to decrease or inhibit the expression level of a 60 polynucleotide causing the pathology (e.g., antisense technology). Alternatively, the polynucleotide can be used to immunize the subject by inducing an immune response thereagainst.

Alternatively, in case the molecule-of-interest is a polypeptide, such a polypeptide can be used in a subject in need of polypeptide therapy, such as a subject having a decreased

24

or no activity of the polypeptide (e.g., due to a genetic disease, auto-antibodies, pathogen infection, degeneration or a decrease in tissue functioning), as well as for other therapeutic applications such as immunization with the polypeptide.

Non-limiting examples of pathologies which require polypeptide therapy and can be treated using the microtube of the invention include, metabolic disorders such as phenylketonuria (PKU), Gaucher disease, muscular dystrophy [Duchenne (DMD) and Becker (BMD) Muscular Dystrophies], Aceruloplasminemia (an iron metabolic disorder), endocrine diseases such as diabetes, autoimmune diseases such as multiple sclerosis (MS), rheumatoid arthritis (RA), and psoriasis, and various cancers (e.g., lymphoma).

Following is a non-limiting list of polypeptides which can be attached to the microtube of the invention in order to treat pathologies requiring polypeptide therapy. Phenylalanine hvdroxvlase [(PAH); GenBank Accession NM_000277.1 (nucleic acid sequence) and NP_000268.1 (SEO ID NO:3: amino acid sequence)] for treating phenylketonuria (PKU), dystrophin [(DMD); GenBank Accession Nos. NM_000109.2 (nucleic acid sequence) and NP_000100.2 (SEQ ID NO:4; amino acid sequence)] for treating Duchenne (DMD) and Becker (BMD) Muscular Dystrophies, beta-glucosidase [(GBA); GenBank Accession Nos. NM_001005741.1 (nucleic acid sequence) and NP_001005741.1 (amino acid sequence; SEQ ID NO:5)] for treating Gaucher disease, insulin [GenBank Accession Nos. NM_000207.1 (nucleic acid sequence) and NP_000198.1 (amino acid sequence; SEQ ID NO:6)] for treating diabetes, and ceruloplasmin ferroxidase [(CP); GenBank Accession Nos. NM_000096.1 (nucleic acid sequence) and NP_000087.1 (SEQ ID NO:7; amino acid sequence)] for treating aceruloplasminemia, CD20 monoclonal antibodies for treating non-Hodgkin's lymphoma and autoimmune disease (Yazawa N, et al., 2005, Proc Natl Acad Sci USA. 102:15178-83) and T-cell receptor peptides for treating of multiple sclerosis (MS), rheumatoid arthritis (RA), and psoriasis (Vandenbark A A, et al., 2001, Neurochem Res. 26:713-30).

Targeted delivery of a drug molecule to a tissue-of-interest is desired in various pathologies, especially in cases where the effect of the drug is deleterious to non-diseased tissues or when high concentrations of drug molecules are required to achieve a therapeutic effect on the diseased tissue (the tissue-of-interest). Thus, for example it is highly desired to have a targeted delivery of a chemotherapy agent or a radiation seed particle to the liver in case of hepatic cancer, or an angiogenic factor to coronary blood vessels, heart or carotid blood vessels in case of ischemia.

According to an embodiment of the invention, for targeted delivery of a drug molecule to a tissue-of-interest via the opening of the microtube (at the targeted tissue), the microtube of the invention is designed such that the electrospun shell is semi-permeable (i.e., prevents passage of the drug molecule but enables the penetration of water or a physiological solution therethrough) and the coat over the internal surface of the shell is attached to the drug molecule.

Such a microtube can be implanted in a subject such that the distal end of the microtube is implanted in or in close proximity to the tissue-of-interest. As used herein the term "proximity" refers to being in a cavity defined by the tissue, for example, if the tissue in which the drug is released is a blood vessel (artery or vein) the cavity is a lumen of such a blood vessel, or if the tissue in which the medication is released is a heart chamber, then the cavity is an atrium or a ventricle. It will be appreciated that the other end of the

25

microtube can be also implanted in proximity to the tissue-of-interest. Alternatively, the proximal end of the microtube can be either sealed using e.g., a laser beam to prevent delivery of the drug to undesired cells/tissues of the subject, or if needed, could be placed outside the body, or subcutaneously such that the microtube can be replenished with additional drug molecules using extra thin needles (e.g., which can penetrate a 5 μ m lumen of the microtube).

Once the microtube of some embodiments of the invention (e.g., a microtube with a semi permeable shell and a 10 drug molecule attached to the coat over the internal surface of the shell) is implanted in the subject it can be filled with a physiological fluid (e.g., of the subject) which is capable of dissolving the water-soluble polymer of the coat over the internal surface of the shell to thereby release the drug 15 molecule therefrom. The released drug molecule flows by capillary forces within the microtube until reaching the end of the open lumen, which is in proximity of the tissue-of-interest

If needed, the microtube according to this embodiment of 20 the invention, can be also replenished with additional drug molecules or other molecules which can increase the effect of the drug molecule released by the microtube. For example, if the drug molecule attached to the microtube is an angiogenic factor, a solution saturated with gasses (e.g., 25 oxygen) can be administered to the microtube (e.g., after implantation in the subject) to thereby increase the anti-ischemic effect of the angiogenic factor.

Targeted delivery of a drug to a tissue-of-interest can be also effected using a microtube in which the shell enables 30 diffusion of the drug molecule therethrough and accordingly, the drug molecule can be released through the shell pores and/or the distal opening of the microtube at the desired tissue.

The invention further envisages the use of the microtube 35 of the invention, which include a molecule-of-interest attached thereto, for guiding cell growth ex vivo or in vivo. For example, neuronal cells can be placed near or in direct contact with the microtube which is attached to necessary growth factors and nutrients needed for neuronal growth. It 40 will be appreciated that once an initial neuronal growth has occurred ex vivo, such a system (i.e., the microtube and the neuronal cells) can be implanted in a subject in need thereof (e.g., a subject with degenerated, damaged or injured neuronal cells) to thereby enable neuronal growth and guidance. 45

The microtube of some embodiments of the invention can be included in a kit/article of manufacture along with a packaging material and/or instructions for use in any of the above described methods or applications.

As used herein the term "about" refers to \pm 10%.

The terms "comprises", "comprising", "includes", "including", "having" and their conjugates mean "including but not limited to". This term encompasses the terms "consisting of" and "consisting essentially of".

The phrase "consisting essentially of" means that the 55 composition or method may include additional ingredients and/or steps, but only if the additional ingredients and/or steps do not materially alter the basic and novel characteristics of the claimed composition or method.

As used herein, the singular form "a", "an" and "the" 60 include plural references unless the context clearly dictates otherwise. For example, the term "a compound" or "at least one compound" may include a plurality of compounds, including mixtures thereof.

Throughout this application, various embodiments of this 65 invention may be presented in a range format. It should be understood that the description in range format is merely for

26

convenience and brevity and should not be construed as an inflexible limitation on the scope of the invention. Accordingly, the description of a range should be considered to have specifically disclosed all the possible subranges as well as individual numerical values within that range. For example, description of a range such as from 1 to 6 should be considered to have specifically disclosed subranges such as from 1 to 3, from 1 to 4, from 1 to 5, from 2 to 4, from 2 to 6, from 3 to 6 etc., as well as individual numbers within that range, for example, 1, 2, 3, 4, 5, and 6. This applies regardless of the breadth of the range.

Whenever a numerical range is indicated herein, it is meant to include any cited numeral (fractional or integral) within the indicated range. The phrases "ranging/ranges between" a first indicate number and a second indicate number and "ranging/ranges from" a first indicate number "to" a second indicate number are used herein interchangeably and are meant to include the first and second indicated numbers and all the fractional and integral numerals therebetween.

As used herein the term "method" refers to manners, means, techniques and procedures for accomplishing a given task including, but not limited to, those manners, means, techniques and procedures either known to, or readily developed from known manners, means, techniques and procedures by practitioners of the chemical, pharmacological, biological, biochemical and medical arts.

An electrospinning apparatus can include a controller programmed with parameters as described herein or measuring output and automatically modifying. Controller can be hardware, software, firmware, with CPU, volatile memory, optional non-volatile memory.

It is appreciated that certain features of the invention, which are, for clarity, described in the context of separate embodiments, may also be provided in combination in a single embodiment. Conversely, various features of the invention, which are, for brevity, described in the context of a single embodiment, may also be provided separately or in any suitable subcombination or as suitable in any other described embodiment of the invention. Certain features described in the context of various embodiments are not to be considered essential features of those embodiments, unless the embodiment is inoperative without those elements.

Various embodiments and aspects of the present invention as delineated hereinabove and as claimed in the claims section below find experimental support in the following examples.

EXAMPLES

Reference is now made to the following examples, which together with the above descriptions, illustrate some embodiments of the invention in a non limiting fashion.

Generally, the nomenclature used herein and the laboratory procedures utilized in the present invention include molecular, biochemical, microbiological and recombinant DNA techniques. Such techniques are thoroughly explained in the literature. See, for example, "Molecular Cloning: A laboratory Manual" Sambrook et al., (1989); "Current Protocols in Molecular Biology" Volumes I-III Ausubel, R. M., ed. (1994); Ausubel et al., "Current Protocols in Molecular Biology", John Wiley and Sons, Baltimore, Md. (1989); Perbal, "A Practical Guide to Molecular Cloning", John Wiley & Sons, New York (1988); Watson et al., "Recombinant DNA", Scientific American Books, New York; Birren et al. (eds) "Genome Analysis: A Laboratory Manual

Series", Vols. 1-4, Cold Spring Harbor Laboratory Press, New York (1998); methodologies as set forth in U.S. Pat. Nos. 4,666,828; 4,683,202; 4,801,531; 5,192,659 and 5,272, 057; "Cell Biology: A Laboratory Handbook", Volumes I-III Cellis, J. E., ed. (1994); "Current Protocols in Immunology" Volumes I-III Coligan J. E., ed. (1994); Stites et al. (eds), "Basic and Clinical Immunology" (8th Edition), Appleton & Lange, Norwalk, Conn. (1994); Mishell and Shiigi (eds), "Selected Methods in Cellular Immunology", W.H. Freeman and Co., New York (1980); available immunoassays are extensively described in the patent and scientific literature, see, for example, U.S. Pat. Nos. 3,791,932; 3,839,153; 3,850,752; 3,850,578; 3,853,987; 3,867,517; 3,879,262; 3,901,654; 3,935,074; 3,984,533; 3,996,345; 4,034,074; 15 4,098,876; 4,879,219; 5,011,771 and 5,281,521; "Oligonucleotide Synthesis" Gait, M. J., ed. (1984); "Nucleic Acid Hybridization" Hames, B. D., and Higgins S. J., eds. (1985); "Transcription and Translation" Hames, B. D., and Higgins S. J., Eds. (1984); "Animal Cell Culture" Freshney, R. I., ed. 20 (1986); "Immobilized Cells and Enzymes" IRL Press, (1986); "A Practical Guide to Molecular Cloning" Perbal, B., (1984) and "Methods in Enzymology" Vol. 1-317, Academic Press; "PCR Protocols: A Guide To Methods And Applications", Academic Press, San Diego, Calif. (1990); 25 Marshak et al., "Strategies for Protein Purification and Characterization—A Laboratory Course Manual" CSHL Press (1996); all of which are incorporated by reference as if fully set forth herein. Other general references are provided throughout this document. The procedures therein are 30 believed to be well known in the art and are provided for the convenience of the reader. All the information contained therein is incorporated herein by reference.

General Materials and Experimental Methods

Enzymes and Solutions-

The compositions of the shell and core solutions are given in Table 3, hereinbelow. All polymers and solvents were purchased from Sigma-Aldrich and were used as is. Alkaline phosphatase (AP) and beta-galactosidase (β -GAL) from E. coli were also purchased from Sigma-Aldrich. AP cleaves monophosphate esters and has a molecular weight of about 80,000 Da. β -GAL is a tetrameric enzyme of 465,396 Da consisting of four identical subunits each (Zabin I., et al., 1980) and catalyzes the hydrolysis of the terminal galactosidyl group of β -galactosides. Both enzymes were initially dissolved in water and then mixed with the core solution.

TABLE 3

Tw	Two types of core-shell microtubes: composition of the solutions									
Type	Shell solution	Core solution								
1	10% PCL 80 K; in CHCl ₃ :DMF (90:10 by weight)	4% (w/w) PEG 600 K + 0.0733 mg/ml AP or 2.38 units/ml β-GAL; in ethanol:H ₂ O (26:74 by weight)								
2	10% PCL 80 K + 1% PEG 6 K; in CHCl ₃ :DMF (90:10 by weight)	4% (w/w) PEG 600 K + 0.0733 mg/ml AP or 2.38 units/ml β-GAL; in ethanol: $\mathrm{H}_2\mathrm{O}$ (26:74 by weight)								

Table 3. Microtubes were formed by co-electrospinning of the shell solution (a first polymeric solution for forming the shell) and a core solution (a second polymeric solution for forming the coat over the internal surface of the shell).

Electrospinning-

Hollow microtubes (core-shell hollow fibers) were fabricated by a co-electrospinning process using the set up

28

described by Sun et al. 2003 and Zussman et al. 2006. All experiments were conducted at room temperature (about 22° C.) and a relative humidity of about 35%. The spinning parameters were as follow: the electrostatic field used was approximately 0.44 kV/cm and the distance between the spinneret and collector plate was 16 cm. The flow rates of both the core and shell solutions were controlled by two syringe pumps and were 3.5 ml/hour for the shell solution and 1 ml/hour for the core solution. The fibers were collected as a strip on the edge of a vertical rotating wheel (Theron A., et al., 2001) having a velocity of 1.2 m/second. For fluorescence microscopy, a few fibers were collected directly onto a microscope slide

Imaging-

50

Images of the fibers were obtained using a Leo Gemini high resolution scanning electron microscope (HRSEM) at an acceleration voltage of 3 kV and a sample to detector distance of 3-5 mm. The specimens were coated with a thin gold film to increase their conductivity. Fluorescence microscope Leica DM IRE2 at excitation and emission wave lengths of 359 and 361 nm, respectively, was used for the imaging of fibers filled with fluorescent product.

Measurement of the Enzyme Activity—

To measure enzyme activity, pieces of mat were weighed and dipped in assay solution according to Table 4, hereinbelow. At each time of sampling, the solution was mixed with a vortex mixer, and 1 ml of the assay mixture was transferred to spectrophotometer cuvette. The absorbance of the solution was measured in a Perkins-Elmer spectrophotometer at a wavelength of 410 nm. For both enzymatic reactions, the substrates are colorless but the products, para-nitrophenol for AP and ortho-nitrophenol for 13-GAL, are yellow with an absorption maximum at 410 nm. After the absorbance was measured, the liquid was returned to the assay vessel. Units, activity and relative activity are defined as follow:

$$Unit = \frac{\Delta A}{\Delta t} \cdot 1000 \tag{1}$$

Activity of the mat = unit·
$$C$$
 where $C = \frac{\text{mass of total mat}}{\text{mass of piece}}$ (2)

Relative Activity (%) =
$$\frac{\text{Activity of the mat}}{\text{Activity of the core solution}} \cdot 100$$
 (3)

Where: ΔA is the difference is the absorbance at difference times, Dt, difference between t_1 and t_2 (two time points) taken in the linear region of the reaction curve, and C is a normalization factor which takes into account the different weight of each piece.

For the fluorescence microscope imaging a drop of the assay solution was put directly on the microscope slide on which a few fibers had been deposited. The fluorescent substrates were methylumbelliferyl-phosphate for alkaline phosphatase (AP) and methylumbelliferyl-galactoside for beta-galactosidase (β -GAL). These were used at the same concentrations as their nitro-phenyl analogs.

TABLE 4

Composition of the assays										
Enzyme	Substrate	Buffer	${\rm H_2O}$	Products						
AP	0.7 mg/ml p-nitrophenyl- phosphate (MW = 217 Da) 1 ml	TRIS- HCl buffer	0.5 ml	p-nitrophenol + PO ₄						
	4-methylumbelliferyl- phosphate: fluorescence microscopy	1.5 ml		4-methylumbelliferone + PO_4						
β-GAL	4 mg/ml o-nitrophenyl-β- D-galactoside 0.2 ml (MW = 301 Da)	Z-buffer 0.7 ml	0.3 ml	o-nitrophenol + galactose						
	4-methylumbelliferyl-β-D-galactoside: fluorescence microscopy			4-methylumbelliferone + galactose						

Table 4.

Example 1

Attachment of Enzymes to Electrospun Microtubes

Experimental Results

Formation of Micro-Tubes—

Two types of electrospun hollow fibers have been fabri- 25 cated with the polymers listed in Table 3, hereinabove. The resultant fibers are hollow structures, namely micro-tubes, as previously described by the present inventors (Dror, Y et al., 2007 and PCT/IB2007/054001, which is fully incorporated herein by reference). The hollow nature of these structures 30 and the different morphologies of the tube walls are demonstrated in the high resolution scanning electron microscope (HRSEM) micrographs presented in FIGS. 1a-d. Type 1 fibers are made with only PCL in the shell and exhibit a rough surface due to the rapid evaporation of the solvent 35 (FIGS. 1a and c). However, this roughness doesn't affect the intact nature of the walls. As PEG is added to the shell solution (Type 2), the walls become increasingly porous (FIGS. 1b and d) and pores can be seen even in the interior surface of the tubes (FIG. 1b). PEG and PCL are partially 40 miscible due to favorable, but weak, intermolecular polarinteractions (Coleman M M, et al., 1991). During fiber solidification along with the evaporation of the solvents, the concentration of the components increases and phase separation takes place. However, since the PEG has a surfactant- 45 like character it deposits an adherent film around the PCL domains resulting in the formation of pores rather than forming solid domains of PEG.

After electrospinning the tubes form a fibrous non-woven, aligned or non-aligned mat. When the mat material is placed 50 in an aqueous environment, the coat over the internal surface of the shell dissolves and the enzymes are released (through desorption) and become active on their substrate(s). For a further description of the desorption process see FIG. 8 and Example 8 hereinbelow. This arrangement of material 55 and the exposure of the enzyme to harmful solvents. allows for flow-through technologies without subjecting the enzyme to the external environment and without the need for chemical attachment and can prevent loss of the entrapped enzyme.

Enzymes Attached to Fibers Maintain Normal Biological 60 Activity-

The kinetics of the enzymatic reaction for alkaline phosphatase was measured as described in the experimental section and is presented in FIGS. 2a-b. The enzymatic reaction with the fibers was compared to the free enzyme in 65 the core solution (before electrospinning) and normalized with respect to the weight of the analyzed pieces. The results

strongly indicate that the enzyme attached in the fiber (e.g., 20 encapsulated) maintains its biological activity after electrospinning and exhibits a reaction curve similar to the free enzyme. The curves are characteristic of enzymatic reactions when there is large excess of substrate. The colored product, p-nitrophenol, diffuses out of the fibers into the surrounding medium as shown in FIG. 2c (Type 1). The reaction rates of the enzyme attached (e.g., contained within) the microtubes are slightly reduced in comparison to the free enzyme. Without being bound by any theory, it is possible that the reaction rates are reduced because of the following reasons: (1) the substrate has to diffuse into the fibers in order to reach the active site of the enzyme; and/or (2) the enzyme has to diffuse outside; (3) the product has to diffuse outside in order to be detected; (4) some enzyme activity was lost during the spinning process itself. The attainment of maximum reaction velocity occurs more rapidly with Type 2 fibers, undoubtedly due to their highly porous character. Upon closer examination of the initial kinetics, it can be seen that Type 1 fibers, which do not contain PEG in the shell, exhibit a linear reaction rate that is much reduced (inset—FIG. 2b, marked by dark arrows) than Type 2 fibers (which include PEG in the shell polymer, and consequently, pores in the microtube shell). Without being bound by any theory, the initial low rate seems to result from the time required for the penetration of the substrate into the fibers since these fibers have a less porous morphology and are more hydrophobic. In the porous fibers (Type 2) the penetration of the substrate is barely hampered since the presence of PEG facilitates the wetting of the outer surface of the fibers and thereby allows the access of the aqueous substrate.

Altogether, these results demonstrate that enzymes attached to (e.g., encapsulated within) electrospun microtubes maintain their enzymatic activity.

Thus, the present technology overcomes both the problem of fiber dissolution and subsequent leaching of the enzyme

Example 2

Electrospun Microtubes can Release Enzymes Attached Thereto

Experimental Results

Enzyme can be Leached Out of the Electrospun Microtubes-

In order to determined whether the substrate diffuses into the fibers or the enzyme diffuses out of the fibers, 3 specimens containing alkaline phosphatase (AP) were cut out

from the non-woven mat and placed in buffer. The buffer from the first sample was taken immediately and assayed (initial rinse). For the second and third samples the buffer was taken and assayed after 24 hours and 72 hours, respectively. The first sample was tested in order to evaluate if any 5 removable enzyme resides on the outer surface of the fibers after the spinning. The relative activity of the escaped enzyme in the buffers for the two types of fibers is presented in FIG. 3. The results clearly indicate that a significant fraction of the enzyme has leached out of the fibers within 10 the first 24 hours. Hence it can be concluded that the reaction monitored for the mat (the electrospun microtubes) is a result of both in-fiber and out-fiber reactions. Interestingly however, for Type 1 fibers, which do not contain PEG in the shell, 82% of enzyme has diffused outside the fibers within 15 24 hours (compared to the mat). For Type 2 fibers, which are more hydrophilic due to PEG, the leaching of the enzyme is less although by no means negligible. The results also point out, as was already mentioned, that Type 2 fibers are the most active system and attains 55% of the activity of the free 20

Alkaline Phosphatase can Migrate to the Outer Surface of the Microtubes During Electrospinning—

Moreover, it is clear that some enzyme is located on the outer surface of the fibers (i.e., attached to the microtube 25 shell) and can immediately enter the surrounding buffer upon rinsing. The migration of the enzyme to the outer surface of the fibers during the spinning is more pronounced in Type 2 and is probably due to PEG which might serve as hydrophilic conduit for the enzyme. It has been already found (Reznik S N, et al., 2006) that in the core-shell process all charges immediately accumulate at the outer surface of the shell in the drop. Ions are preferentially subjected to this migration as the electric field is applied. In the present case, the protein, which is a charged molecule, can also migrate to 35 the outer surface during electrospinning. This may explain the relatively large quantity of enzyme that is released by rinsing.

Altogether, these results demonstrate that type 2 fibers attain more enzymatic activity as compared to type 1 fibers. 40 In addition, the enzyme within the type 2 fibers (which include PEG in the shell) is present in both the outer surface and the inner surface of the microtubes. Thus, these results demonstrate that the even though the enzyme is mixed with the core solution, it is capable of migrating into the outer 45 surface of the shell during the electrospinning process.

Example 3

Electrospun Microtubes can Serve as Microreactors

Experimental Results

β-Galactosidase Remains within the Hollow Fiber Micro-Reactor—

The kinetics of the β -GAL reaction were determined as 55 described hereinabove for AP and the results are presented in FIGS. **4***a-b*. Briefly, pieces of mat (eletrospun microtubes) were immersed in buffer for a quick rinse, 24 or 72 hours to determine if there is any leaching of the enzyme (FIG. **5**, shown are results of Type 2 as an example). Two striking differences between the AP and β -GAL series were observed (FIG. **6**): (i) the β -GAL reaction rate of the Type 1 mat is much slower relative to that found for AP, and the activity of AP in the type 1 mat is slower relative to that found for the activity of the free enzyme in the core solution (prior to electrospinning). In type 2, conversely, the activity of the β -GAL is higher than that of AP and the reaction velocity of

32

the β -GAL enzyme is comparable to the free enzyme. This intense response of Type 2 is attributed to the high porosity of these fibers and their hydrophilic nature; (ii) the results shown in FIG. 5 for Type 2 fibers demonstrate that the β -GAL enzyme doesn't diffuse out of the fibers even after 72 hours even though Type 2 might be thought to be the more likely to allow the enzyme to escape. That is, the reaction takes place only within the fibers. Thus, for β -GAL, the hollow fibers act as a micro-reactor; the substrate which enters the "reactor" through the entire porous shell is cleaved by the encapsulated enzyme and the reaction product then diffuses out into the surrounding medium. Thus, without being bound by any theory it seems that the reduction in the reaction rate for Type 1 is related solely to the slow diffusion of the substrate into the fibers. The size of the enzyme seems to affect both the amount of enzyme that can migrate during the spinning to the outer surface of the fibers and its subsequent escape into the surrounding medium. In the case of β -GAL only a small amount of enzyme was detected in the rinsing buffer for Type 2 while for Type 1 no enzyme was detected at all (data not shown). As was argued before, in the fibers which do not contain PEG in the shell (Type 1) an additional moderate slope at the very beginning of the reaction can be observed (FIG. 4b-inset, marked by dark arrow) due to the relatively high hydrophobicity of the surface which hinders the access of the aqueous dissolved substrate.

Altogether, these results demonstrate that the Type 1 system results in lowered enzymatic activity, especially for large proteins such as $\beta\text{-}GAL$, which cannot diffuse through the shell pores. The fact that Type 1 fibers are hydrophobic and non-porous and thus inhibit the entrance of these substrates seems to make Type 1 fibers less efficacious for their use as flow-through reactors. This is in contrast to the remarkably efficient system obtained with Type 2 fibers.

The Enzymatic Reaction Occurs within Type I Microtubes (Hollow Fibers)—

Visual evidence that the enzymatic reaction occurs within the type 1 fibers was obtained by using a substrate in which one of the products is fluorescent. For both AP and 13-GAL enzymes the substrates used liberate 4-methyumbelliferone after hydrolysis which allows imaging by fluorescence microscopy. As is clearly seen in FIGS. 7a-b, the interior of the fibers is fluorescent while the surrounding medium is dark. Thus, these results clearly show that both enzymatic reactions (of AP and β -GAL), in fact, take place within the fibers. It is important to emphasize that, in contrast to the mat immersion experiments, this method is very sensitive and enables the detection of very small amounts of product which accumulates within a relatively short time. Indeed, these images were acquired within 1-2 minutes after the substrate was applied, a time scale which is larger than the characteristic time [about 10 seconds, as was calculated by the present inventors (Dror Y., et al., 2007)] of the diffusion through the micro-tubes wall. Hence, these results are not in contradiction to those of the mat immersion experiments in which the kinetics were followed over a much longer period during which both the product and the enzyme (in the AP case) can diffuse outside the fibers. In FIG. 7b short slugs (sections) of fluorescent liquid are observed. This phenomenon has been previously found in such fibers (Dror Y., et al., 2007).

The results shown in Examples 1-3 demonstrate the direct incorporation of enzymes into micro-tubes fabricated by co-electrospinning by introduction of the enzymes into the aqueous core solution of the microtube (e.g., PEO). The shell solution in this case was made of PCL dissolved in

mixture of chloroform and DMF. The separation between the outer organic and inner aqueous phases was found to preserve enzyme activity during and after spinning when the electrospun fibers (mats) were subsequently placed in an aqueous environment.

Two types of micro-tubes were fabricated which differ in their shell morphology. Type 2 shells were produced by adding PEG to the shell solution. By using a mixture of PEG and PCL in the shell, pores were formed during the solidification process and this, in turn, directly affected the transfer of molecules into and out of the fibers. As a consequence, the more porous fibers (Type 2) exhibited higher rates of enzymatic reaction. In addition two enzymes differing in their molecular weight were incorporated: AP and β -GAL. The difference in the molecular weight between the enzymes 15 was well reflected in the kinetics of the enzymatic reactions for both types of micro-tubes. While AP could diffuse outside the fibers, β-GAL remained in the fibers without any leaching of the enzyme and the progress of the reaction depended only on the arrival of the substrate from the 20 surroundings. Thus, the AP fibers act as an enzyme release device in which the release rate can be tailored by modifying the morphology of the shell and, on the other hand, the β-GAL fibers act as an enzymatic micro-reactor with an efficient provision of the substrate through the entire surface 25 area and efficient discharge of the product. Thus, by manipulating the morphology of the shell, the substrate supply and product release rate can be controlled. This method of encapsulation can be used when a separation between the enzyme and an external aqueous environment is desired (e.g. 30 with living tissue to avoid immunological reactions). The remarkable retention of the enzyme activity for β -GAL Type 2 fibers clearly demonstrates that this approach preserves the activity of the enzyme.

Another advantage of the core-shell fiber method is the small volume within the micro-tubes which enables the quick buildup of the product. This is important for enzymes working in sequence where the local concentration of the product of the first reaction serves as the substrate for a subsequent reaction. In this regard, these nanotubes are somewhat analogous to living cells except that any manner of enzymes may be added to the fibers without regard to their biological origin. Another advantage of this system is that unlike living cells, there is no discrimination as to which type of small molecules may enter these tubes. For example, 45 phosphorylated molecules (like p-nitrophenyl phosphate) which do not enter *Escherichia coli* cells can enter the microtubes described herein.

Example 4

Use of the Electrospun Microtubes as Microreactors for Chemical Transitions

The Microtubes of the Invention as Reactions for Multi- 55 Step Enzymatic Processes—

In order to synthesize or catabolize molecules which require multi-step enzymatic processes the present inventors have devised electrospun microtubes which include the enzymes participating in the multi-step enzymatic process, 60 attached thereto, as follows.

For a biochemical pathway which involves the conversion of A to E via compounds B, C and D, the second polymeric solution for forming the coat over the internal surface of the shell (also referred to as a core solution) is mixed with the 65 following enzymes: the enzyme that converts the starting substrate A to intermediate compound B, the enzyme that

34

converts B to C, the enzyme that converts C to D, and the enzyme that converts D to the end product E (see for example, FIG. 9). It should be noted that due to the proximity of the enzymes to each other in the microtube (which can be a closed micro-reactor), the local concentrations of each of the intermediate molecules, i.e., compounds B, C, and D is relatively high, which enables the kinetics of the reactions to occur, similarly to their concentrations in cells or cell compartments (e.g., mitochondria). The shell solution is made from hydrophobic polymers (water insoluble polymers), with or without the addition of PEG.

Thus, the microtube of the invention enables higher local concentrations of intermediate compounds which can not be reached from the same starting material (substrate A) if an open system (such as any solid substrate to which an enzyme is immobilized) is used.

The creation of micro- and nano-fibers containing enzymes simulates the cellular structure because two or more different enzymes involved in a particular synthesis or degradation can be put into proximity of one another. The interior of the tube is quite parallel to that of cells except that the borders of the tube are made from a water insoluble substance whereas living cells are encompassed by lipid membranes. In addition the microtubes are much longer than cells but quite similar in other dimensions to a bacterial cell. In the electrospun fibers, any small molecule can pass through the water insoluble barrier (pores can be made) regardless of its chemistry with the only provision that the small molecule be water soluble.

Thus, the present technology allows the entrapment of high concentrations of an enzyme or several enzymes within a confined space. Single or multi-step reactions can then take place where the product of one reaction is the substrate of a second and the second enzyme is spatially nearby. While such multi-step reactions can occur in an open system, the time necessary to reach the end product is orders of magnitude greater than within the microtube of the invention.

The Microtubes of the Invention can Include Enzymes from Different Species—

Another very important advantage of these electrospun fibers is that there is no limitation of which enzymes can be embedded. In nature, cells contain enzymes useful for their growth and reproduction. Organisms have not been designed or selected for industrial processes desired by humans. The microtubes of the invention allow any desired combination of enzymes to be brought together. This might mean that enzymes from totally different organisms (e.g. flies and humans) could be placed together for some use while in nature no organism exists with this combination.

The Microtubes of the Invention as Micro-Reactors for the Production of Molecules which are Intermediate Compounds of a Natural Process—

The enzymes encapsulated might be those carrying out part of pathway making the end product a substance that is usually an intermediate molecule in living organisms. This allows one to synthesize molecules that cannot be obtained in any amount from living material because the concentration of intermediates in cells is usually very low (in the order of $10~\mu M$ or less). For example, to synthesize indoleglycerol phosphate which is an intermediate in tryptophan synthesis within the cells of lower organisms, the enzymes that participate in the conversion of anthranilate (an inexpensive compound) to indole-glycerol phosphate should be included in the microtube, while the enzymes that continue the synthesis of tryptophan from indole-glycerol phosphate are excluded from the microtube. In summary, many combinations of enzymes from different organisms may be put

together without any genetic engineering and partial sets of enzymes can also be used. The number of possible useful combinations is therefore very large.

Thus, the microtubes of the invention can be used as enzymatic micro-reactors where the inner space enables a 5 confined but free reaction space. The substrate diffuses through the shell to the inner space where the enzymatic reaction takes place and the product can then diffuse out.

Example 5

The Electrospun Microtubes as Biosensors

Since the electrospun microtubes of the invention are insoluble in aqueous solutions, they can provide an excellent 15 tool for the construction of biosensors.

Since any enzyme or combination of enzymes can be encapsulated in the electrospun microtubes, a variety of biosensors can be devised. For example, enzymes that are presence of heavy metals. Another example, firefly luciferase, for example can be electrospun with its luciferin cofactor and any reaction affecting ATP production can be used in conjunction with light output the signal.

Example 6

The Electrospun Microtubes for Flow-Through Applications

Water Purification or Detoxification-

The electrospun microtubes of the invention, which are made of a water-insoluble outer shell, enable the flow of liquids. It will be appreciated that enzymes embedded in such microtubes can be used to purify the liquid flowing past 35 the microtubes as molecules diffuse in and out of them.

Thus the present inventors have devised water purification or detoxification apparatuses, as follows. The second polymeric solution which forms the coat over the internal surface of the shell (also referred to as a core solution) [which is 40 made of water-soluble polymer(s)] includes enzymes which remove a toxic moiety from water, such as the gene product of the atzA gene from Pseudomonas ADP that removes the chlorine from atrazine, a toxic substance. The shell solution [which is made of water-insoluble polymer(s)] is designed 45 so as to enable water flow within the microtube. The effluent would thereby be rendered free of atrazine and safe for animal and human consumption.

Dialysis-

The microtubes of the invention can be used in various 50 applications which remove certain compounds, such as dialysis procedures on humans. Thus, the electrospun microtubes can be made using a shell polymer which prevents the diffusion of enzymes therethrough, yet enables passage of water and substances that need to be purified. It will be 55 appreciated that such microtubes can be also implanted into a subject in need thereof (e.g., a subject in need of dialysis), and due to the structure of a closed conduit, which prevents passage of embedded enzymes through the shell, there is no immune response to the implanted conduit.

Example 7

The Electrospun Microtubes for Enzyme Therapy

Since the electrospun microtubes of the invention are insoluble in aqueous solutions, they should provide excel36

lent tool for the construction of material for enzyme therapy. Some individuals lack certain enzymes, usually as a result of their being homozygous for recessive alleles that lead to synthesis of an inactive enzyme. Gene therapy attempts to introduce the missing active gene which thereby leads to the production of an active enzyme. However, this technique is still quite inefficient. A different way of treating such patients is enzyme therapy, in which the missing enzyme is exogenously supplied to the subject. The main drawbacks of the second method is that injection of enzymes often leads to the formation of antibodies against them and the half-life of the enzymes within the body may be quite short.

Thus, the present inventors have devised an apparatus for enzyme therapy, as follows. Briefly, the electrospun microtubes which include a water-insoluble shell can include enzymes which are needed for enzyme therapy, and be further implanted in a subject in need thereof.

Enzyme Therapy for PKU-

Phenylketonuria (PKU) occurs in slightly less than 1 per sensitive to heavy metals exhibit loss of activity in the 20 10000 individuals and is an autosomal recessive genetic disease caused by homozygosity of alleles encoding defective enzymes. Phenylalanine and tyrosine are amino acids that are found in most proteins. In humans, the source of these two amino acids is dietary protein. In normal individuals, excess phenylalanine is converted to tyrosine. Excess tyrosine, in turn, is broken down to fumarate and acetoacetate. Both tyrosine and phenylalanine are essential for human protein synthesis. In addition, tyrosine is the precursor of melanin (skin and eye pigment) and for certain hormone like substances such as thyroxine. Phenylketonuria is caused by the lack of the enzyme (phenylalanine 4-monooxygease EC 1.14.16.1) that converts phenylalanine to tyrosine. The result of this defect is the accumulation of phenylalanine in the blood along with a number of compounds that are derived from it (e.g. phenylpyruvic acid and phenyllacetic acid). The result is brain damage (and an IQ of 30-70) as some of these compounds are toxic. The current method of preventing deterioration of the disease is to limit the intake of phenylalanine. The present inventors have envisaged that PKU can be treated by implanting electrospun microtubes containing the missing enzyme, phenylalanine hydroxylase (PAH; GenBank Accession No. NP_00026), in a subject diagnosed with PKU, and thereby enabling the breakdown of excess phenylalanine in the subject. It will be appreciated that in this case, the microtube can be designed so as to enable release of enzyme from the inner surface through the outer shell (e.g., using PEG in the outer shell) or alternatively can be designed such that the enzyme is entrapped (or remains) within the microtubes and effects its activity there (e.g., by diffusion of the substrate or end-product through the shell pores, or microtube opening(s)). As phenylalanine hydroxylase can be phosphorylated (with a molecular weight of consists of 50,000 Da) or dephosphorylated (with a molecular weight of 49,000 Da), the size of the pores in the shell should enable passage (by diffusion) of each of these forms (e.g., about 5 nm in diameter). Microtubes can be made with smaller pores that will prevent the loss of the enzyme which will remain within the microtube.

Example 8

60

The Desorption Process

FIG. 8 schematically depicts the desorption of a molecule-of-interest to the microtube of the invention. After the electrospinning process, the molecule-of-interest (e.g., a

protein, an enzyme) is adsorbed to the inner side of the microtubes. As mentioned, the porosity of the microtubes can be controlled (e.g., adding PEG to the shell polymer), therefore the shell consists of nanopores (see #3 in FIG. 8) with an opening to the outer surface of the microtube, which herein the pores are considered to have a cylindrical shape. When immersing the microtubes in a solution (e.g., a tissue culture medium, a physiological solution or any buffer) most nanopores opening are accessible to the solution.

Once the microtubes are immersed in the solution the nanopores are filled by the solution through capillary rise (see arrow #1 in FIG. 8). It will be appreciated that the time of the capillary rise depends on the solution rheological properties (viscosity and surface tension), the wetting angle and the geometry of the nanopore (length and radius). The solution penetrates to the microtube and start wetting and filling its entire inner volume. Desorption of the moleculeof-interest from the microtube wall depends mainly on the rate of the release of the molecule-of-interest from the polymer of the second polymeric solution. Finally, the molecule-of-interest (e.g., protein/enzymes) diffuses (see arrow #2 in FIG. 8) into the solution and released to the surroundings of the microtubes (assuming that the major release is through the microtubes envelope.

Note that the geometry (radius and length) of the nanopore is controllable, by adjusting the shell thickness, or by blending, more PEG to the shell polymer. Therefore, the release from or confinement in the microtube of the invention is controllable, e.g., in certain cases the molecule-ofinterest is released from the microtube, whereas in other cases, the molecule-of-interest remains in the inner volume of the microtube. As shown in FIG. 2a, a molecule-ofinterest is released from the microtube of the invention in a controlled manner, which can be extended beyond 2500, minutes, e.g., for several days and months.

Although the invention has been described in conjunction with specific embodiments thereof, it is evident that many alternatives, modifications and variations will be apparent to those skilled in the art. Accordingly, it is intended to embrace all such alternatives, modifications and variations 40 that fall within the spirit and broad scope of the appended

All publications, patents and patent applications mentioned in this specification are herein incorporated in their entirety by reference into the specification, to the same 45 extent as if each individual publication, patent or patent application was specifically and individually indicated to be incorporated herein by reference. In addition, citation or identification of any reference in this application shall not be construed as an admission that such reference is available as 50 prior art to the present invention. To the extent that section headings are used, they should not be construed as necessarily limiting.

38

REFERENCES

Additional References are Cited in Text

- (1) Xie, J.; Hsieh, Y. L. J. Mater. Sci. 2003, 38, 2125-2133. (2) Zhang, Y. Z.; Wang, X.; Lim, C. T.; Ramakrishna, S. Biomacromolecules 2006, 7, 1049-1057.
 - (3) Jiang, H.; Hu, Y.; Zhao, P.; Li, Y.; Zhu, K. J. Biomed. Mater. Res. Part B: Appl. Biomater. 2006, 79B, 50-57.
 - (4) Patel, A. C.; Li, S.; Yuan, J. M.; Wei, Y. Nano Letters 2006, 6, 1042-1046.
- (5) Salalha, W.; Kuhn, J.; Dror, Y.; Zussman, E. Nanotech-
- nology 2006, 17, 4675-4681. (6) Sun, Z.; Zussman, E.; Yarin, A. L.; Wendorff, J. H.; Greiner, A. Adv. Mater. 2003, 15, 1929-1932.
- (7) Ye, P.; Xu, Z. K.; Wu, J.; Innocent, C.; Seta, P. Macromolecules 2006, 39, 1041-1045.
 - (8) Jia, H.; Zhu, G.; Vugrinovich, B.; Kataphinan, W.; Reneker, D. H.; Wang, P. Biotechnol. Prog. 2002, 18, 1027-1032
- (9) Kim, T. G.; Park, T. G. Biotechnol. Prog. 2006, 22, 1108-1113.
- (10) Huang, X. J.; Xu, Z. K.; Wan, L. S.; Innocent, C.; Seta, P. Macromol. Rapid Commun. 2006, 27, 1341-1345.
- (11) Wu, L.; Yuan, X.; Sheng, J. J. Membr. Sci. 2005, 250, 167-173.
- 25 (12) Jiang, H.; Fang, D.; Hsiao, B. S.; Chu, B.; Chen, W. Biomacromolecules 2004, 4, 326-333.
 - (13) Zeng, J.; Aigner, A.; Czubayko, F.; Kissel, T.; Wendorff, J. H.; Greiner, A. Biomacromolecules 2005, 6, 1484-
- (14) Herricks, T. E.; Kim, S. H.; Kim, J.; Li, D.; Kwak, J. H.; Grate, J. W.; Kim, S. H.; Xia, Y. J. Mater. Chem. 2005, 15,
 - (15) Dror, Y.; Salalha, W.; Avrahami, R.; Zussman, E.; Yarin, A. L.; Dersch, R.; Greiner, A.; Wendorff, J. H. 2007, Small 3:1064-73.
- (16) Zabin, I.; Flower, A. V. b-Galactosidase, the Lactose Permease Protein and Thiogalactoside Transacetylase; in The Operon pp 89-121, Cold Spring Harbor Laboratory: NY, 1980.
- (17) Zussman, E.; Yarin, A. L.; Bazilevsky, V.; Avrahami, R.; Feldman, M. Adv. Mater. 2006, 18, 348-353.
- (18) Theron, A.; Zussman, E.; Yarin, A. L. Nanotechnology 2001, 12, 384.
- (19) Coleman, M. M.; Graf, J. F.; Painter, P. C. Specific Interactions and the Miscibility of Polymer Blends; Technomic Publishing Company: Pennsylvania, 1991.
- (20) Reznik, S, N.; Yarin, A. L.; Zussman, E.; Bercovici, L. Physics of Fluids 2006, 18, 1-13.
- (21) D. H. Reneker, A. L. Yarin, E. Zussman, H. Xu, Advances in Applied Mechanics 2006, 40.
- (22) S. Ramakrishna, K. Fujihara, W.-e. Teo, Lim, T. C., Z. Ma, An Introduction to Electrospinning and Nanofibers, 1 ed., World Scientific Publishing Company, 2005.
 - (23) D. Li, Y. N. Xia, Advanced Materials 2004, 16, 1151.

SEQUENCE LISTING

```
<160> NUMBER OF SEO ID NOS: 9
<210> SEQ ID NO 1
<211> LENGTH: 524
<212> TYPE: PRT
<213> ORGANISM: Homo sapiens
<400> SEOUENCE: 1
Met Ile Ser Pro Phe Leu Val Leu Ala Ile Gly Thr Cys Leu Thr Asn
                5
                                    10
                                                         15
```

Ser	Leu	Val	Pro 20	Glu	Lys	Glu	Lys	Asp 25	Pro	Lys	Tyr	Trp	Arg 30	Asp	Gln
Ala	Gln	Glu 35	Thr	Leu	Lys	Tyr	Ala 40	Leu	Glu	Leu	Gln	Lys 45	Leu	Asn	Thr
Asn	Val 50	Ala	Lys	Asn	Val	Ile 55	Met	Phe	Leu	Gly	Asp 60	Gly	Met	Gly	Val
Ser 65	Thr	Val	Thr	Ala	Ala 70	Arg	Ile	Leu	Lys	Gly 75	Gln	Leu	His	His	Asn 80
Pro	Gly	Glu	Glu	Thr 85	Arg	Leu	Glu	Met	Asp 90	Lys	Phe	Pro	Phe	Val 95	Ala
Leu	Ser	Lys	Thr 100	Tyr	Asn	Thr	Asn	Ala 105	Gln	Val	Pro	Asp	Ser 110	Ala	Gly
Thr	Ala	Thr 115	Ala	Tyr	Leu	CAa	Gly 120	Val	Lys	Ala	Asn	Glu 125	Gly	Thr	Val
Gly	Val 130	Ser	Ala	Ala	Thr	Glu 135	Arg	Ser	Arg	Cys	Asn 140	Thr	Thr	Gln	Gly
Asn 145	Glu	Val	Thr	Ser	Ile 150	Leu	Arg	Trp	Ala	Lys 155	Asp	Ala	Gly	Lys	Ser 160
Val	Gly	Ile	Val	Thr 165	Thr	Thr	Arg	Val	Asn 170	His	Ala	Thr	Pro	Ser 175	Ala
Ala	Tyr	Ala	His 180	Ser	Ala	Asp	Arg	Asp 185	Trp	Tyr	Ser	Asp	Asn 190	Glu	Met
Pro	Pro	Glu 195	Ala	Leu	Ser	Gln	Gly 200	Cys	Lys	Asp	Ile	Ala 205	Tyr	Gln	Leu
Met	His 210	Asn	Ile	Arg	Asp	Ile 215	Asp	Val	Ile	Met	Gly 220	Gly	Gly	Arg	Lys
Tyr 225	Met	Tyr	Pro	Lys	Asn 230	Lys	Thr	Asp	Val	Glu 235	Tyr	Glu	Ser	Asp	Glu 240
Lys	Ala	Arg	Gly	Thr 245	Arg	Leu	Asp	Gly	Leu 250	Asp	Leu	Val	Asp	Thr 255	Trp
ràs	Ser	Phe	Lys 260	Pro	Arg	Tyr	Lys	His 265	Ser	His	Phe	Ile	Trp 270	Asn	Arg
Thr	Glu	Leu 275	Leu	Thr	Leu	Asp	Pro 280	His	Asn	Val	Asp	Tyr 285	Leu	Leu	Gly
Leu	Phe 290	Glu	Pro	Gly	Asp	Met 295	Gln	Tyr	Glu	Leu	Asn 300	Arg	Asn	Asn	Val
Thr 305	Asp	Pro	Ser		Ser 310	Glu	Met	Val		Val 315		Ile	Gln	Ile	Leu 320
Arg	Lys	Asn	Pro	Lys 325	Gly	Phe	Phe	Leu	Leu 330	Val	Glu	Gly	Gly	Arg 335	Ile
Asp	His	Gly	His 340	His	Glu	Gly	Lys	Ala 345	Lys	Gln	Ala	Leu	His 350	Glu	Ala
Val	Glu	Met 355	Asp	Arg	Ala	Ile	Gly 360	Gln	Ala	Gly	Ser	Leu 365	Thr	Ser	Ser
Glu	Asp 370	Thr	Leu	Thr	Val	Val 375	Thr	Ala	Asp	His	Ser 380	His	Val	Phe	Thr
Phe 385	Gly	Gly	Tyr	Thr	Pro 390	Arg	Gly	Asn	Ser	Ile 395	Phe	Gly	Leu	Ala	Pro 400
Met	Leu	Ser	Asp	Thr 405	Asp	Lys	Lys	Pro	Phe 410	Thr	Ala	Ile	Leu	Tyr 415	Gly
Asn	Gly	Pro	Gly 420	Tyr	Lys	Val	Val	Gly 425	Gly	Glu	Arg	Glu	Asn 430	Val	Ser

-continued

Met Val Asp Tyr Ala His Asn Asn Tyr Gln Ala Gln Ser Ala Val Pro Leu Arg His Glu Thr His Gly Glu Asp Val Ala Val Phe Ser Lys Gly Pro Met Ala His Leu Leu His Gly Val His Glu Gln Asn Tyr Val Pro His Val Met Ala Tyr Ala Ala Cys Ile Gly Ala Asn Leu Gly His Cys Ala Pro Ala Ser Ser Ala Gly Ser Leu Ala Ala Gly Pro Leu Leu Leu Ala Leu Ala Leu Tyr Pro Leu Ser Val Leu Phe <210> SEQ ID NO 2 <211> LENGTH: 1024 <212> TYPE: PRT <213 > ORGANISM: Homo sapiens <400> SEQUENCE: 2 Met Thr Met Ile Thr Asp Ser Leu Ala Val Val Leu Gln Arg Arg Asp Trp Glu Asn Pro Gly Val Thr Gln Leu Asn Arg Leu Ala Ala His Pro Pro Phe Ala Ser Trp Arg Asn Ser Glu Glu Ala Arg Thr Asp Arg Pro Ser Gln Gln Leu Arg Ser Leu Asn Gly Glu Trp Arg Phe Ala Trp Phe Pro Ala Pro Glu Ala Val Pro Glu Ser Trp Leu Glu Cys Asp Leu Pro 70 Glu Ala Asp Thr Val Val Val Pro Ser Asn Trp Gln Met His Gly Tyr Asp Ala Pro Ile Tyr Thr Asn Val Thr Tyr Pro Ile Thr Val Asn Pro Pro Phe Val Pro Thr Glu Asn Pro Thr Gly Cys Tyr Ser Leu Thr Phe 120 Asn Val Asp Glu Ser Trp Leu Gln Glu Gly Gln Thr Arg Ile Ile Phe Asp Gly Val Asn Ser Ala Phe His Leu Trp Cys Asn Gly Arg Trp Val Gly Tyr Gly Gln Asp Ser Arg Leu Pro Ser Glu Phe Asp Leu Ser Ala Phe Leu Arg Ala Gly Glu Asn Arg Leu Ala Val Met Val Leu Arg Trp Ser Asp Gly Ser Tyr Leu Glu Asp Gln Asp Met Trp Arg Met Ser Gly Ile Phe Arg Asp Val Ser Leu Leu His Lys Pro Thr Thr Gln Ile Ser 215 Asp Phe His Val Ala Thr Arg Phe Asn Asp Asp Phe Ser Arg Ala Val 230 235 Leu Glu Ala Glu Val Gln Met Cys Gly Glu Leu Arg Asp Tyr Leu Arg Val Thr Val Ser Leu Trp Gln Gly Glu Thr Gln Val Ala Ser Gly Thr 265 Ala Pro Phe Gly Gly Glu Ile Ile Asp Glu Arg Gly Gly Tyr Ala Asp 280

Arg	Val 290	Thr	Leu	Arg	Leu	Asn 295	Val	Glu	Asn	Pro	300 Tàa	Leu	Trp	Ser	Ala
Glu 305	Ile	Pro	Asn	Leu	Tyr 310	Arg	Ala	Val	Val	Glu 315	Leu	His	Thr	Ala	Asp 320
Gly	Thr	Leu	Ile	Glu 325	Ala	Glu	Ala	Cys	Asp 330	Val	Gly	Phe	Arg	Glu 335	Val
Arg	Ile	Glu	Asn 340	Gly	Leu	Leu	Leu	Leu 345	Asn	Gly	Lys	Pro	Leu 350	Leu	Ile
Arg	Gly	Val 355	Asn	Arg	His	Glu	His 360	His	Pro	Leu	His	Gly 365	Gln	Val	Met
Asp	Glu 370	Gln	Thr	Met	Val	Gln 375	Asp	Ile	Leu	Leu	Met 380	Lys	Gln	Asn	Asn
Phe 385	Asn	Ala	Val	Arg	Сув 390	Ser	His	Tyr	Pro	Asn 395	His	Pro	Leu	Trp	Tyr 400
Thr	Leu	Cys	Asp	Arg 405	Tyr	Gly	Leu	Tyr	Val 410	Val	Asp	Glu	Ala	Asn 415	Ile
Glu	Thr	His	Gly 420	Met	Val	Pro	Met	Asn 425	Arg	Leu	Thr	Asp	Asp 430	Pro	Arg
Trp	Leu	Pro 435	Ala	Met	Ser	Glu	Arg 440	Val	Thr	Arg	Met	Val 445	Gln	Arg	Asp
Arg	Asn 450	His	Pro	Ser	Val	Ile 455	Ile	Trp	Ser	Leu	Gly 460	Asn	Glu	Ser	Gly
His 465	Gly	Ala	Asn	His	Asp 470	Ala	Leu	Tyr	Arg	Trp 475	Ile	Lys	Ser	Val	Asp 480
Pro	Ser	Arg	Pro	Val 485	Gln	Tyr	Glu	Gly	Gly 490	Gly	Ala	Asp	Thr	Thr 495	Ala
Thr	Asp	Ile	Ile 500	CÀa	Pro	Met	Tyr	Ala 505	Arg	Val	Asp	Glu	Asp 510	Gln	Pro
Phe	Pro	Ala 515	Val	Pro	Lys	Trp	Ser 520	Ile	Lys	ГÀз	Trp	Leu 525	Ser	Leu	Pro
Gly	Glu 530	Thr	Arg	Pro	Leu	Ile 535	Leu	CAa	Glu	Tyr	Ala 540	His	Ala	Met	Gly
Asn 545	Ser	Leu	Gly	Gly	Phe 550	Ala	Lys	Tyr	Trp	Gln 555	Ala	Phe	Arg	Gln	Tyr 560
Pro	Arg	Leu	Gln	Gly 565	Gly	Phe	Val	Trp	Asp 570	Trp	Val	Asp	Gln	Ser 575	Leu
Ile	Lys		Asp 580	Glu	Asn	Gly		Pro 585		Ser	Ala	Tyr	Gly 590	Gly	Asp
Phe	Gly	Asp 595	Thr	Pro	Asn	Asp	Arg 600	Gln	Phe	Cys	Met	Asn 605	Gly	Leu	Val
Phe	Ala 610	Asp	Arg	Thr	Pro	His 615	Pro	Ala	Leu	Thr	Glu 620	Ala	Lys	His	Gln
Gln 625	Gln	Phe	Phe	Gln	Phe 630	Arg	Leu	Ser	Gly	Gln 635	Thr	Ile	Glu	Val	Thr 640
Ser	Glu	Tyr	Leu	Phe 645	Arg	His	Ser	Asp	Asn 650	Glu	Leu	Leu	His	Trp 655	Met
Val	Ala	Leu	Asp 660	Gly	Lys	Pro	Leu	Ala 665	Ser	Gly	Glu	Val	Pro 670	Leu	Asp
Val	Ala	Pro 675	Gln	Gly	Lys	Gln	Leu 680	Ile	Glu	Leu	Pro	Glu 685	Leu	Pro	Gln
Pro	Glu 690	Ser	Ala	Gly	Gln	Leu 695	Trp	Leu	Thr	Val	Arg 700	Val	Val	Gln	Pro

-continued

Asn Ala Thr Ala Trp Ser Glu Ala Gly His Ile Ser Ala Trp Gln Gln 710 Trp Arg Leu Ala Glu Asn Leu Ser Val Thr Leu Pro Ala Ala Ser His Ala Ile Pro His Leu Thr Thr Ser Glu Met Asp Phe Cys Ile Glu Leu 745 Gly Asn Lys Arg Trp Gln Phe Asn Arg Gln Ser Gly Phe Leu Ser Gln Met Trp Ile Gly Asp Lys Lys Gln Leu Leu Thr Pro Leu Arg Asp Gln Phe Thr Arg Ala Pro Leu Asp Asn Asp Ile Gly Val Ser Glu Ala Thr Arg Ile Asp Pro Asn Ala Trp Val Glu Arg Trp Lys Ala Ala Gly His Tyr Gln Ala Glu Ala Ala Leu Leu Gln Cys Thr Ala Asp Thr Leu Ala Asp Ala Val Leu Ile Thr Thr Ala His Ala Trp Gln His Gln Gly Lys Thr Leu Phe Ile Ser Arg Lys Thr Tyr Arg Ile Asp Gly Ser Gly Gln 855 Met Ala Ile Thr Val Asp Val Glu Val Ala Ser Asp Thr Pro His Pro 870 875 Ala Arg Ile Gly Leu Asn Cys Gln Leu Ala Gln Val Ala Glu Arg Val 890 Asn Trp Leu Gly Leu Gly Pro Gln Glu Asn Tyr Pro Asp Arg Leu Thr Ala Ala Cys Phe Asp Arg Trp Asp Leu Pro Leu Ser Asp Met Tyr Thr 920 Pro Tyr Val Phe Pro Ser Glu Asn Gly Leu Arg Cys Gly Thr Arg Glu 935 Leu Asn Tyr Gly Pro His Gln Trp Arg Gly Asp Phe Gln Phe Asn Ile Ser Arg Tyr Ser Gln Gln Gln Leu Met Glu Thr Ser His Arg His Leu Leu His Ala Glu Glu Gly Thr Trp Leu Asn Ile Asp Gly Phe His Met 985 Gly Ile Gly Gly Asp Asp Ser Trp Ser Pro Ser Val Ser Ala Glu Phe Gln Leu Ser Ala Gly Arg Tyr His Tyr Gln Leu Val Trp Cys Gln Lys <210> SEQ ID NO 3 <211> LENGTH: 452 <212> TYPE: PRT <213> ORGANISM: Homo sapiens <400> SEQUENCE: 3 Met Ser Thr Ala Val Leu Glu Asn Pro Gly Leu Gly Arg Lys Leu Ser Asp Phe Gly Gln Glu Thr Ser Tyr Ile Glu Asp Asn Cys Asn Gln Asn 25 Gly Ala Ile Ser Leu Ile Phe Ser Leu Lys Glu Glu Val Gly Ala Leu

Ala	Lys 50	Val	Leu	Arg	Leu	Phe 55	Glu	Glu	Asn	Asp	Val 60	Asn	Leu	Thr	His
Ile 65	Glu	Ser	Arg	Pro	Ser 70	Arg	Leu	Lys	Lys	Asp 75	Glu	Tyr	Glu	Phe	Phe 80
Thr	His	Leu	Asp	Lуз 85	Arg	Ser	Leu	Pro	Ala 90	Leu	Thr	Asn	Ile	Ile 95	Lys
Ile	Leu	Arg	His 100	Asp	Ile	Gly	Ala	Thr 105	Val	His	Glu	Leu	Ser 110	Arg	Asp
ГÀз	ГЛа	Lys 115	Asp	Thr	Val	Pro	Trp 120	Phe	Pro	Arg	Thr	Ile 125	Gln	Glu	Leu
Asp	Arg 130	Phe	Ala	Asn	Gln	Ile 135	Leu	Ser	Tyr	Gly	Ala 140	Glu	Leu	Asp	Ala
Asp 145	His	Pro	Gly	Phe	Lys 150	Asp	Pro	Val	Tyr	Arg 155	Ala	Arg	Arg	Lys	Gln 160
Phe	Ala	Asp	Ile	Ala 165	Tyr	Asn	Tyr	Arg	His 170	Gly	Gln	Pro	Ile	Pro 175	Arg
Val	Glu	Tyr	Met 180	Glu	Glu	Glu	Lys	Lys 185	Thr	Trp	Gly	Thr	Val 190	Phe	Lys
Thr	Leu	Lys 195	Ser	Leu	Tyr	Lys	Thr 200	His	Ala	Cys	Tyr	Glu 205	Tyr	Asn	His
Ile	Phe 210	Pro	Leu	Leu	Glu	Lуs 215	Tyr	Сув	Gly	Phe	His 220	Glu	Asp	Asn	Ile
Pro 225	Gln	Leu	Glu	Asp	Val 230	Ser	Gln	Phe	Leu	Gln 235	Thr	CAa	Thr	Gly	Phe 240
Arg	Leu	Arg	Pro	Val 245	Ala	Gly	Leu	Leu	Ser 250	Ser	Arg	Asp	Phe	Leu 255	Gly
Gly	Leu	Ala	Phe 260	Arg	Val	Phe	His	Сув 265	Thr	Gln	Tyr	Ile	Arg 270	His	Gly
Ser	Lys	Pro 275	Met	Tyr	Thr	Pro	Glu 280	Pro	Asp	Ile	CAa	His 285	Glu	Leu	Leu
	290					295	Asp				300				
305					310		Gly			315					320
				325			Thr		330					335	
			340					345					350		
		355					360					365			Glu
	370					375	Asn				380				
Leu 385	Tyr	Tyr	Val	Ala	Glu 390	Ser	Phe	Asn	Asp	Ala 395	Lys	Glu	Lys	Val	Arg 400
Asn	Phe	Ala	Ala	Thr 405	Ile	Pro	Arg	Pro	Phe 410	Ser	Val	Arg	Tyr	Asp 415	Pro
Tyr	Thr	Gln	Arg 420	Ile	Glu	Val	Leu	Asp 425	Asn	Thr	Gln	Gln	Leu 430	Lys	Ile
Leu	Ala	Asp 435	Ser	Ile	Asn	Ser	Glu 440	Ile	Gly	Ile	Leu	Сув 445	Ser	Ala	Leu
Gln	Lys 450	Ile	Lys												

<21	<210> SEQ ID NO 4 <211> LENGTH: 3677 <212> TYPE: PRT														
	3 > OI			Homo	sa]	pien	s								
< 40	0 > SI	EQUEI	NCE :	4											
Met 1	Glu	Asp	Glu	Arg 5	Glu	Asp	Val	Gln	Lys 10	Lys	Thr	Phe	Thr	Lys 15	Trp
Val	Asn	Ala	Gln 20	Phe	Ser	Lys	Phe	Gly 25	Lys	Gln	His	Ile	Glu 30	Asn	Leu
Phe	Ser	Asp 35	Leu	Gln	Asp	Gly	Arg 40	Arg	Leu	Leu	Asp	Leu 45	Leu	Glu	Gly
Leu	Thr 50	Gly	Gln	Lys	Leu	Pro 55	Lys	Glu	Lys	Gly	Ser 60	Thr	Arg	Val	His
Ala 65	Leu	Asn	Asn	Val	Asn 70	Lys	Ala	Leu	Arg	Val 75	Leu	Gln	Asn	Asn	Asn 80
Val	Asp	Leu	Val	Asn 85	Ile	Gly	Ser	Thr	Asp 90	Ile	Val	Asp	Gly	Asn 95	His
Lys	Leu	Thr	Leu 100	Gly	Leu	Ile	Trp	Asn 105	Ile	Ile	Leu	His	Trp	Gln	Val
Lys	Asn	Val 115	Met	Lys	Asn	Ile	Met 120	Ala	Gly	Leu	Gln	Gln 125	Thr	Asn	Ser
Glu	Lys 130	Ile	Leu	Leu	Ser	Trp 135	Val	Arg	Gln	Ser	Thr 140	Arg	Asn	Tyr	Pro
Gln 145	Val	Asn	Val	Ile	Asn 150	Phe	Thr	Thr	Ser	Trp 155	Ser	Asp	Gly	Leu	Ala 160
Leu	Asn	Ala	Leu	Ile 165	His	Ser	His	Arg	Pro 170	Asp	Leu	Phe	Asp	Trp 175	Asn
Ser	Val	Val	Cys 180	Gln	Gln	Ser	Ala	Thr 185	Gln	Arg	Leu	Glu	His 190	Ala	Phe
Asn	Ile	Ala 195	Arg	Tyr	Gln	Leu	Gly 200	Ile	Glu	Lys	Leu	Leu 205	Asp	Pro	Glu
Asp	Val 210	Asp	Thr	Thr	Tyr	Pro 215	Asp	ГÀз	ГÀЗ	Ser	Ile 220	Leu	Met	Tyr	Ile
Thr 225	Ser	Leu	Phe	Gln	Val 230	Leu	Pro	Gln	Gln	Val 235	Ser	Ile	Glu	Ala	Ile 240
Gln	Glu	Val	Glu	Met 245	Leu	Pro	Arg	Pro	Pro 250	Lys	Val	Thr	Lys	Glu 255	Glu
His	Phe	Gln	Leu 260	His	His	Gln	Met	His 265	Tyr	Ser	Gln	Gln	Ile 270	Thr	Val
Ser	Leu	Ala 275	Gln	Gly	Tyr	Glu	Arg 280	Thr	Ser	Ser	Pro	Lys 285	Pro	Arg	Phe
Lys	Ser 290	Tyr	Ala	Tyr	Thr	Gln 295	Ala	Ala	Tyr	Val	Thr 300	Thr	Ser	Asp	Pro
Thr 305	Arg	Ser	Pro	Phe	Pro 310	Ser	Gln	His	Leu	Glu 315	Ala	Pro	Glu	Asp	Lys 320
Ser	Phe	Gly	Ser	Ser 325	Leu	Met	Glu	Ser	Glu 330	Val	Asn	Leu	Asp	Arg 335	Tyr
Gln	Thr	Ala	Leu 340	Glu	Glu	Val	Leu	Ser 345	Trp	Leu	Leu	Ser	Ala 350	Glu	Asp
Thr	Leu	Gln 355	Ala	Gln	Gly	Glu	Ile 360	Ser	Asn	Asp	Val	Glu 365	Val	Val	Lys
Asp	Gln 370	Phe	His	Thr	His	Glu 375	Gly	Tyr	Met	Met	Asp 380	Leu	Thr	Ala	His

Gln 385	Gly	Arg	Val	Gly	Asn 390	Ile	Leu	Gln	Leu	Gly 395	Ser	ГÀв	Leu	Ile	Gly 400
Thr	Gly	ГÀв	Leu	Ser 405	Glu	Asp	Glu	Glu	Thr 410	Glu	Val	Gln	Glu	Gln 415	Met
Asn	Leu	Leu	Asn 420	Ser	Arg	Trp	Glu	Cys 425	Leu	Arg	Val	Ala	Ser 430	Met	Glu
Lys	Gln	Ser 435	Asn	Leu	His	Arg	Val 440	Leu	Met	Asp	Leu	Gln 445	Asn	Gln	Lys
Leu	Lys 450	Glu	Leu	Asn	Asp	Trp 455	Leu	Thr	Lys	Thr	Glu 460	Glu	Arg	Thr	Arg
Lys 465	Met	Glu	Glu	Glu	Pro 470	Leu	Gly	Pro	Asp	Leu 475	Glu	Asp	Leu	Lys	Arg 480
Gln	Val	Gln	Gln	His 485	rys	Val	Leu	Gln	Glu 490	Asp	Leu	Glu	Gln	Glu 495	Gln
Val	Arg	Val	Asn 500	Ser	Leu	Thr	His	Met 505	Val	Val	Val	Val	Asp 510	Glu	Ser
Ser	Gly	Asp 515	His	Ala	Thr	Ala	Ala 520	Leu	Glu	Glu	Gln	Leu 525	Lys	Val	Leu
Gly	Asp 530	Arg	Trp	Ala	Asn	Ile 535	Cys	Arg	Trp	Thr	Glu 540	Asp	Arg	Trp	Val
Leu 545	Leu	Gln	Asp	Ile	Leu 550	Leu	Lys	Trp	Gln	Arg 555	Leu	Thr	Glu	Glu	Gln 560
Cys	Leu	Phe	Ser	Ala 565	Trp	Leu	Ser	Glu	Lys 570	Glu	Asp	Ala	Val	Asn 575	Lys
Ile	His	Thr	Thr 580	Gly	Phe	Lys	Asp	Gln 585	Asn	Glu	Met	Leu	Ser 590	Ser	Leu
Gln	Lys	Leu 595	Ala	Val	Leu	Lys	Ala 600	Asp	Leu	Glu	Lys	Lys	Lys	Gln	Ser
Met	Gly 610	Lys	Leu	Tyr	Ser	Leu 615	Lys	Gln	Asp	Leu	Leu 620	Ser	Thr	Leu	Lys
Asn 625	Lys	Ser	Val	Thr	Gln 630	Lys	Thr	Glu	Ala	Trp 635	Leu	Asp	Asn	Phe	Ala 640
Arg	Сув	Trp	Asp	Asn 645	Leu	Val	Gln	Lys	Leu 650	Glu	ГЛа	Ser	Thr	Ala 655	Gln
Ile	Ser	Gln	Ala 660	Val	Thr	Thr	Thr	Gln 665	Pro	Ser	Leu	Thr	Gln 670	Thr	Thr
Val	Met	Glu 675	Thr	Val	Thr	Thr	Val 680	Thr	Thr	Arg	Glu	Gln 685	Ile	Leu	Val
Lys	His 690	Ala	Gln	Glu	Glu	Leu 695	Pro	Pro	Pro	Pro	Pro 700	Gln	ГЛа	Lys	Arg
Gln 705	Ile	Thr	Val	Asp	Ser 710	Glu	Ile	Arg	Lys	Arg 715	Leu	Asp	Val	Asp	Ile 720
Thr	Glu	Leu	His	Ser 725	Trp	Ile	Thr	Arg	Ser 730	Glu	Ala	Val	Leu	Gln 735	Ser
Pro	Glu	Phe	Ala 740	Ile	Phe	Arg	Lys	Glu 745	Gly	Asn	Phe	Ser	Asp 750	Leu	Lys
Glu	Lys	Val 755	Asn	Ala	Ile	Glu	Arg 760	Glu	Lys	Ala	Glu	Lув 765	Phe	Arg	ГÀз
Leu	Gln 770	Asp	Ala	Ser	Arg	Ser 775	Ala	Gln	Ala	Leu	Val 780	Glu	Gln	Met	Val
Asn 785	Glu	Gly	Val	Asn	Ala 790	Asp	Ser	Ile	Lys	Gln 795	Ala	Ser	Glu	Gln	Leu 800
Asn	Ser	Arg	Trp	Ile	Glu	Phe	Cys	Gln	Leu	Leu	Ser	Glu	Arg	Leu	Asn

- C	ontinued
- C	ontinuea

												COII	CIII	aca	
				805					810					815	
Trp	Leu	Glu	Tyr 820	Gln	Asn	Asn	Ile	Ile 825	Ala	Phe	Tyr	Asn	Gln 830	Leu	Gln
Gln	Leu	Glu 835		Met	Thr	Thr	Thr 840	Ala	Glu	Asn	Trp	Leu 845	Lys	Ile	Gln
Pro	Thr 850	Thr	Pro	Ser	Glu	Pro 855	Thr	Ala	Ile	Lys	Ser 860	Gln	Leu	Lys	Ile
865 865		Asp	Glu	Val	Asn 870	Arg	Leu	Ser	Gly	Leu 875	Gln	Pro	Gln	Ile	Glu 880
Arg	Leu	Lys	Ile	Gln 885	Ser	Ile	Ala	Leu	Lys	Glu	ГÀа	Gly	Gln	Gly 895	
Met	Phe	Leu	Asp 900	Ala	Asp	Phe	Val	Ala 905	Phe	Thr	Asn	His	Phe 910	ГЛа	Gln
Val	Phe	Ser 915	Asp	Val	Gln	Ala	Arg 920	Glu	Lys	Glu	Leu	Gln 925	Thr	Ile	Phe
Asp	Thr 930	Leu	Pro	Pro	Met	Arg 935	Tyr	Gln	Glu	Thr	Met 940	Ser	Ala	Ile	Arg
Thr 945	Trp	Val	Gln	Gln	Ser 950	Glu	Thr	Lys	Leu	Ser 955	Ile	Pro	Gln	Leu	Ser 960
Val	Thr	Asp	Tyr	Glu 965	Ile	Met	Glu	Gln	Arg 970	Leu	Gly	Glu	Leu	Gln 975	Ala
Leu	Gln	Ser	Ser 980	Leu	Gln	Glu	Gln	Gln 985	Ser	Gly	Leu	Tyr	Tyr 990	Leu	Ser
Thr	Thr	Val 995		Glu	Met	Ser	Lys 100		a Ala	a Pro	Se:	r Gl		le S	er Arg
Lys	Tyr 1010		n Se:	r Glu	ı Phe	Gl: 101		lu I	le G	lu G		rg ' 020	Trp :	Lys 1	Lys
Leu	Ser 1025		r Glı	n Let	ı Val	Gl: 103		is Cy	ys G	ln Ly		eu 035	Glu (Glu (Gln
Met	Asn 1040		s Lei	ı Arç	g Lys	11e		ln As	sn H	is I		ln 050	Thr :	Leu 1	Lys
ГÀа	Trp 1055		t Ala	a Glu	ı Val	As ₁		al Ph	ne Le	eu Ly		lu 065	Glu '	Trp 1	Pro
Ala	Leu 1070		y Asj	Sei	r Glu	. Ile 10		eu Ly	va P	ys GI		eu :	Lys (Gln (Cys
Arg		Let		l Sei	r Asp		e G: 90		nr I	le G		ro 095	Ser :	Leu 1	Asn
Ser	Val 1100		n Glı	ı Gly	/ Gly	Gl1		ys I	le Ly	ys A:		lu . 110	Ala (Glu I	Pro
Glu	Phe 1115		a Se:	r Arg	g Leu	. Gli 112		hr G	lu Le	eu Ly		lu : 125	Leu I	Asn '	Thr
Gln	Trp 1130		o Hi	s Met	Cys	Gli 113		ln Va	al Ty	yr Al		rg :	Lys (Glu i	Ala
Leu	Lys 1149	_	y Gly	y Let	ı Glu	. Ly:		hr Va	al Se	er Le		ln : 155	Lys 1	Asp 1	Leu
Ser	Glu 1160		t Hi:	s Glı	ı Trp	Met		hr G	ln A	la G		lu (Glu '	Tyr i	Leu
Glu	Arg 1175		p Phe	e Glu	ı Tyr	Ly:		hr Pi	ro As	ab G		eu 1	Gln :	Lys 1	Ala
Val	Glu 1190		ı Met	t Ly:	a Arg	Ala 119		ys G	lu G	lu Al		ln (Gln :	Lys (Glu
Ala	Lys 1205		l Ly:	s Lei	ı Leu	. Th:		lu Se	er Va	al As		er ' 215	Val :	Ile Z	Ala

Gln	Ala 1220	Pro	Pro	Val	Ala	Gln 1225	Glu	Ala	Leu	Lys	Lys 1230	Glu	Leu	Glu
Thr	Leu 1235	Thr	Thr	Asn	Tyr	Gln 1240	Trp	Leu	СЛа	Thr	Arg 1245	Leu	Asn	Gly
Lys	Cys 1250	ГЛа	Thr	Leu	Glu	Glu 1255	Val	Trp	Ala	CAa	Trp 1260	His	Glu	Leu
Leu	Ser 1265	Tyr	Leu	Glu	Lys	Ala 1270	Asn	Lys	Trp	Leu	Asn 1275	Glu	Val	Glu
Phe	Lys 1280	Leu	ГЛа	Thr	Thr	Glu 1285	Asn	Ile	Pro	Gly	Gly 1290	Ala	Glu	Glu
Ile	Ser 1295	Glu	Val	Leu	Asp	Ser 1300	Leu	Glu	Asn	Leu	Met 1305	Arg	His	Ser
Glu	Asp 1310	Asn	Pro	Asn	Gln	Ile 1315	Arg	Ile	Leu	Ala	Gln 1320	Thr	Leu	Thr
Asp	Gly 1325	Gly	Val	Met	Asp	Glu 1330	Leu	Ile	Asn	Glu	Glu 1335	Leu	Glu	Thr
Phe	Asn 1340	Ser	Arg	Trp	Arg	Glu 1345	Leu	His	Glu	Glu	Ala 1350	Val	Arg	Arg
Gln	Lys 1355	Leu	Leu	Glu	Gln	Ser 1360	Ile	Gln	Ser	Ala	Gln 1365	Glu	Thr	Glu
Lys	Ser 1370	Leu	His	Leu	Ile	Gln 1375	Glu	Ser	Leu	Thr	Phe 1380	Ile	Asp	Lys
Gln	Leu 1385	Ala	Ala	Tyr	Ile	Ala 1390	Asp	Lys	Val	Asp	Ala 1395	Ala	Gln	Met
Pro	Gln 1400	Glu	Ala	Gln	Lys	Ile 1405	Gln	Ser	Asp	Leu	Thr 1410	Ser	His	Glu
Ile	Ser 1415	Leu	Glu	Glu	Met	Lys 1420	Lys	His	Asn	Gln	Gly 1425	Lys	Glu	Ala
Ala	Gln 1430	Arg	Val	Leu	Ser	Gln 1435	Ile	Asp	Val	Ala	Gln 1440	Lys	ГÀЗ	Leu
Gln	Asp 1445	Val	Ser	Met	Lys	Phe 1450	Arg	Leu	Phe	Gln	Lys 1455	Pro	Ala	Asn
Phe	Glu 1460	Gln	Arg	Leu	Gln	Glu 1465	Ser	Lys	Met	Ile	Leu 1470	Asp	Glu	Val
ГÀа	Met 1475	His	Leu	Pro	Ala	Leu 1480	Glu	Thr	ГÀа	Ser	Val 1485	Glu	Gln	Glu
Val	Val 1490	Gln	Ser	Gln	Leu	Asn 1495	His	CÀa	Val	Asn	Leu 1500	Tyr	Lys	Ser
Leu	Ser 1505	Glu	Val	Lys	Ser	Glu 1510	Val	Glu	Met	Val	Ile 1515	ГÀа	Thr	Gly
Arg	Gln 1520	Ile	Val	Gln	Lys	Lys 1525	Gln	Thr	Glu	Asn	Pro 1530	ГÀа	Glu	Leu
Asp	Glu 1535	Arg	Val	Thr	Ala	Leu 1540	ГÀв	Leu	His	Tyr	Asn 1545	Glu	Leu	Gly
Ala	Lys 1550	Val	Thr	Glu	Arg	Lys 1555	Gln	Gln	Leu	Glu	Lys 1560	Cya	Leu	ГÀа
Leu	Ser 1565	Arg	Lys	Met	Arg	Lys 1570	Glu	Met	Asn	Val	Leu 1575	Thr	Glu	Trp
Leu	Ala 1580	Ala	Thr	Asp	Met	Glu 1585	Leu	Thr	Lys	Arg	Ser 1590	Ala	Val	Glu
Gly	Met 1595	Pro	Ser	Asn	Leu	Asp 1600	Ser	Glu	Val	Ala	Trp 1605	Gly	Lys	Ala

Thr	Gln 1610	Lys	Glu	Ile	Glu	Lys 1615	Gln	Lys	Val	His	Leu 1620	Lys	Ser	Ile
Thr	Glu 1625	Val	Gly	Glu	Ala	Leu 1630	Lys	Thr	Val	Leu	Gly 1635	Lys	Lys	Glu
Thr	Leu 1640	Val	Glu	Asp	Lys	Leu 1645	Ser	Leu	Leu	Asn	Ser 1650	Asn	Trp	Ile
Ala	Val 1655	Thr	Ser	Arg	Ala	Glu 1660	Glu	Trp	Leu	Asn	Leu 1665	Leu	Leu	Glu
Tyr	Gln 1670	Lys	His	Met	Glu	Thr 1675	Phe	Asp	Gln	Asn	Val 1680	Asp	His	Ile
Thr	Lys 1685	Trp	Ile	Ile	Gln	Ala 1690	Asp	Thr	Leu	Leu	Asp 1695	Glu	Ser	Glu
Lys	Lys 1700	Lys	Pro	Gln	Gln	Lys 1705	Glu	Asp	Val	Leu	Lys 1710	Arg	Leu	Lys
Ala	Glu 1715	Leu	Asn	Aap	Ile	Arg 1720	Pro	ГÀа	Val	Asp	Ser 1725	Thr	Arg	Asp
Gln	Ala 1730	Ala	Asn	Leu	Met	Ala 1735	Asn	Arg	Gly	Asp	His 1740	Cys	Arg	ГÀа
Leu	Val 1745	Glu	Pro	Gln	Ile	Ser 1750	Glu	Leu	Asn	His	Arg 1755	Phe	Ala	Ala
Ile	Ser 1760	His	Arg	Ile	Lys	Thr 1765	Gly	Lys	Ala	Ser	Ile 1770	Pro	Leu	ГÀа
Glu	Leu 1775	Glu	Gln	Phe	Asn	Ser 1780	Asp	Ile	Gln	Lys	Leu 1785	Leu	Glu	Pro
Leu	Glu 1790	Ala	Glu	Ile	Gln	Gln 1795	Gly	Val	Asn	Leu	Lys 1800	Glu	Glu	Asp
Phe	Asn 1805	Lys	Asp	Met	Asn	Glu 1810	Asp	Asn	Glu	Gly	Thr 1815	Val	Lys	Glu
Leu	Leu 1820	Gln	Arg	Gly	Asp	Asn 1825	Leu	Gln	Gln	Arg	Ile 1830	Thr	Asp	Glu
Arg	Lys 1835	Arg	Glu	Glu	Ile	Lys 1840	Ile	Lys	Gln	Gln	Leu 1845	Leu	Gln	Thr
Lys	His 1850	Asn	Ala	Leu	Lys	Asp 1855	Leu	Arg	Ser	Gln	Arg 1860	Arg	Lys	ГÀв
Ala	Leu 1865	Glu	Ile	Ser	His	Gln 1870	Trp	Tyr	Gln	Tyr	Lys 1875	Arg	Gln	Ala
Asp	Asp 1880		Leu			Leu 1885					Lys 1890		Leu	Ala
Ser	Leu 1895	Pro	Glu	Pro	Arg	Asp 1900	Glu	Arg	Lys	Ile	Lys 1905	Glu	Ile	Asp
Arg	Glu 1910	Leu	Gln	Lys	Lys	Lys 1915	Glu	Glu	Leu	Asn	Ala 1920	Val	Arg	Arg
Gln	Ala 1925	Glu	Gly	Leu	Ser	Glu 1930	Asp	Gly	Ala	Ala	Met 1935	Ala	Val	Glu
Pro	Thr 1940	Gln	Ile	Gln	Leu	Ser 1945	Lys	Arg	Trp	Arg	Glu 1950	Ile	Glu	Ser
Lys	Phe 1955	Ala	Gln	Phe	Arg	Arg 1960	Leu	Asn	Phe	Ala	Gln 1965	Ile	His	Thr
Val	Arg 1970	Glu	Glu	Thr	Met	Met 1975	Val	Met	Thr	Glu	Asp 1980	Met	Pro	Leu
Glu	Ile 1985	Ser	Tyr	Val	Pro	Ser 1990	Thr	Tyr	Leu	Thr	Glu 1995	Ile	Thr	His
Val	Ser	Gln	Ala	Leu	Leu	Glu	Val	Glu	Gln	Leu	Leu	Asn	Ala	Pro

_														
	2000					2005					2010			
Asp	Leu 2015	CAa	Ala	Lys	Asp	Phe 2020	Glu	Asp	Leu	Phe	Lys 2025	Gln	Glu	Glu
Ser	Leu 2030	ràa	Asn	Ile	Lys	Asp 2035	Ser	Leu	Gln	Gln	Ser 2040	Ser	Gly	Arg
Ile	Asp 2045	Ile	Ile	His	Ser	Lys 2050	Lys	Thr	Ala	Ala	Leu 2055	Gln	Ser	Ala
Thr	Pro 2060	Val	Glu	Arg	Val	Lys 2065	Leu	Gln	Glu	Ala	Leu 2070	Ser	Gln	Leu
Asp	Phe 2075	Gln	Trp	Glu	Lys	Val 2080	Asn	Lys	Met	Tyr	Lys 2085	Asp	Arg	Gln
Gly	Arg 2090	Phe	Asp	Arg	Ser	Val 2095	Glu	Lys	Trp	Arg	Arg 2100	Phe	His	Tyr
Asp	Ile 2105	ràa	Ile	Phe	Asn	Gln 2110	Trp	Leu	Thr	Glu	Ala 2115	Glu	Gln	Phe
Leu	Arg 2120	Lys	Thr	Gln	Ile	Pro 2125	Glu	Asn	Trp	Glu	His 2130	Ala	Lys	Tyr
Lys	Trp 2135	Tyr	Leu	Lys	Glu	Leu 2140	Gln	Asp	Gly	Ile	Gly 2145	Gln	Arg	Gln
Thr	Val 2150	Val	Arg	Thr	Leu	Asn 2155	Ala	Thr	Gly	Glu	Glu 2160	Ile	Ile	Gln
Gln	Ser 2165	Ser	Lys	Thr	Asp	Ala 2170	Ser	Ile	Leu	Gln	Glu 2175	Lys	Leu	Gly
Ser	Leu 2180	Asn	Leu	Arg	Trp	Gln 2185	Glu	Val	Cys	Lys	Gln 2190	Leu	Ser	Asp
Arg	Lys 2195	rys	Arg	Leu	Glu	Glu 2200	Gln	Lys	Asn	Ile	Leu 2205	Ser	Glu	Phe
Gln	Arg 2210	Asp	Leu	Asn	Glu	Phe 2215	Val	Leu	Trp	Leu	Glu 2220	Glu	Ala	Asp
Asn	Ile 2225	Ala	Ser	Ile	Pro	Leu 2230	Glu	Pro	Gly	Lys	Glu 2235	Gln	Gln	Leu
Lys	Glu 2240	rys	Leu	Glu	Gln	Val 2245	Lys	Leu	Leu	Val	Glu 2250	Glu	Leu	Pro
Leu	Arg 2255	Gln	Gly	Ile	Leu	Lys 2260	Gln	Leu	Asn	Glu	Thr 2265	Gly	Gly	Pro
Val	Leu 2270	Val	Ser	Ala	Pro	Ile 2275	Ser	Pro	Glu	Glu	Gln 2280	Asp	Lys	Leu
Glu	Asn 2285	Lys	Leu	Lys	Gln	Thr 2290	Asn	Leu	Gln	Trp	Ile 2295	Lys	Val	Ser
Arg	Ala 2300	Leu	Pro	Glu	Lys	Gln 2305	Gly	Glu	Ile	Glu	Ala 2310	Gln	Ile	Lys
Asp	Leu 2315	Gly	Gln	Leu	Glu	Lys 2320	Lys	Leu	Glu	Asp	Leu 2325	Glu	Glu	Gln
Leu	Asn 2330	His	Leu	Leu	Leu	Trp 2335	Leu	Ser	Pro	Ile	Arg 2340	Asn	Gln	Leu
Glu	Ile 2345	Tyr	Asn	Gln	Pro	Asn 2350	Gln	Glu	Gly	Pro	Phe 2355	Asp	Val	Gln
Glu	Thr 2360	Glu	Ile	Ala	Val	Gln 2365	Ala	Lys	Gln	Pro	Asp 2370	Val	Glu	Glu
Ile	Leu 2375	Ser	Lys	Gly	Gln	His 2380	Leu	Tyr	Lys	Glu	Lys 2385	Pro	Ala	Thr
Gln	Pro 2390	Val	Lys	Arg	Lys	Leu 2395	Glu	Asp	Leu	Ser	Ser 2400	Glu	Trp	Lys

-cont:	inued
-cont:	inued

Ala	Val 2405	Asn	Arg	Leu	Leu	Gln 2410	Glu	Leu	Arg	Ala	Lys 2415	Gln	Pro	Asp
Leu	Ala 2420	Pro	Gly	Leu	Thr	Thr 2425	Ile	Gly	Ala	Ser	Pro 2430	Thr	Gln	Thr
Val	Thr 2435	Leu	Val	Thr	Gln	Pro 2440	Val	Val	Thr	Lys	Glu 2445	Thr	Ala	Ile
Ser	Lys 2450	Leu	Glu	Met	Pro	Ser 2455	Ser	Leu	Met	Leu	Glu 2460	Val	Pro	Ala
Leu	Ala 2465	Asp	Phe	Asn	Arg	Ala 2470	Trp	Thr	Glu	Leu	Thr 2475	Asp	Trp	Leu
Ser	Leu 2480	Leu	Asp	Gln	Val	Ile 2485	Lys	Ser	Gln	Arg	Val 2490	Met	Val	Gly
Asp	Leu 2495	Glu	Asp	Ile	Asn	Glu 2500	Met	Ile	Ile	Lys	Gln 2505	Lys	Ala	Thr
Met	Gln 2510	Asp	Leu	Glu	Gln	Arg 2515	Arg	Pro	Gln	Leu	Glu 2520	Glu	Leu	Ile
Thr	Ala 2525	Ala	Gln	Asn	Leu	Lув 2530	Asn	ГЛа	Thr	Ser	Asn 2535	Gln	Glu	Ala
Arg	Thr 2540	Ile	Ile	Thr	Asp	Arg 2545	Ile	Glu	Arg	Ile	Gln 2550	Asn	Gln	Trp
Asp	Glu 2555	Val	Gln	Glu	His	Leu 2560	Gln	Asn	Arg	Arg	Gln 2565	Gln	Leu	Asn
Glu	Met 2570	Leu	ГÀв	Asp	Ser	Thr 2575	Gln	Trp	Leu	Glu	Ala 2580	ГÀв	Glu	Glu
Ala	Glu 2585	Gln	Val	Leu	Gly	Gln 2590	Ala	Arg	Ala	ГÀЗ	Leu 2595	Glu	Ser	Trp
ГÀз	Glu 2600	Gly	Pro	Tyr	Thr	Val 2605	Asp	Ala	Ile	Gln	Lys 2610	Lys	Ile	Thr
Glu	Thr 2615	ГÀз	Gln	Leu	Ala	Lys 2620	Asp	Leu	Arg	Gln	Trp 2625	Gln	Thr	Asn
Val	Asp 2630	Val	Ala	Asn	Asp	Leu 2635	Ala	Leu	Lys	Leu	Leu 2640	Arg	Asp	Tyr
Ser	Ala 2645	Asp	Asp	Thr	Arg	Lys 2650	Val	His	Met	Ile	Thr 2655	Glu	Asn	Ile
Asn	Ala 2660	Ser	Trp	Arg	Ser	Ile 2665	His	Lys	Arg	Val	Ser 2670	Glu	Arg	Glu
Ala	Ala 2675	Leu	Glu	Glu	Thr	His 2680	Arg	Leu	Leu	Gln	Gln 2685	Phe	Pro	Leu
Asp	Leu 2690	Glu	ГÀа	Phe	Leu	Ala 2695	Trp	Leu	Thr	Glu	Ala 2700	Glu	Thr	Thr
Ala	Asn 2705	Val	Leu	Gln	Asp	Ala 2710	Thr	Arg	ГЛа	Glu	Arg 2715	Leu	Leu	Glu
Asp	Ser 2720	ГÀв	Gly	Val	Lys	Glu 2725	Leu	Met	Lys	Gln	Trp 2730	Gln	Asp	Leu
Gln	Gly 2735	Glu	Ile	Glu	Ala	His 2740	Thr	Asp	Val	Tyr	His 2745	Asn	Leu	Asp
Glu	Asn 2750	Ser	Gln	Lys	Ile	Leu 2755	Arg	Ser	Leu	Glu	Gly 2760	Ser	Asp	Asp
Ala	Val 2765	Leu	Leu	Gln	Arg	Arg 2770	Leu	Asp	Asn	Met	Asn 2775	Phe	Lys	Trp
Ser	Glu 2780	Leu	Arg	Lys	Lys	Ser 2785	Leu	Asn	Ile	Arg	Ser 2790	His	Leu	Glu

Ala	Ser 2795	Ser	Asp	Gln	Trp	Lys 2800	Arg	Leu	His	Leu	Ser 2805	Leu	Gln	Glu
Leu	Leu 2810	Val	Trp	Leu	Gln	Leu 2815	ГÀв	Asp	Asp	Glu	Leu 2820	Ser	Arg	Gln
Ala	Pro 2825	Ile	Gly	Gly	Asp	Phe 2830	Pro	Ala	Val	Gln	Lys 2835	Gln	Asn	Asp
Val	His 2840	Arg	Ala	Phe	Lys	Arg 2845	Glu	Leu	ГÀа	Thr	Lys 2850	Glu	Pro	Val
Ile	Met 2855	Ser	Thr	Leu	Glu	Thr 2860	Val	Arg	Ile	Phe	Leu 2865	Thr	Glu	Gln
Pro	Leu 2870	Glu	Gly	Leu	Glu	Lys 2875	Leu	Tyr	Gln	Glu	Pro 2880	Arg	Glu	Leu
Pro	Pro 2885	Glu	Glu	Arg	Ala	Gln 2890	Asn	Val	Thr	Arg	Leu 2895	Leu	Arg	ГÀа
Gln	Ala 2900	Glu	Glu	Val	Asn	Thr 2905	Glu	Trp	Glu	ГЛа	Leu 2910	Asn	Leu	His
Ser	Ala 2915	Asp	Trp	Gln	Arg	Lys 2920	Ile	Asp	Glu	Thr	Leu 2925	Glu	Arg	Leu
Gln	Glu 2930	Leu	Gln	Glu	Ala	Thr 2935	Asp	Glu	Leu	Asp	Leu 2940	Lys	Leu	Arg
Gln	Ala 2945	Glu	Val	Ile	ГЛа	Gly 2950	Ser	Trp	Gln	Pro	Val 2955	Gly	Asp	Leu
Leu	Ile 2960	Asp	Ser	Leu	Gln	Asp 2965	His	Leu	Glu	Lys	Val 2970	Lys	Ala	Leu
Arg	Gly 2975	Glu	Ile	Ala	Pro	Leu 2980	Lys	Glu	Asn	Val	Ser 2985	His	Val	Asn
Asp	Leu 2990	Ala	Arg	Gln	Leu	Thr 2995	Thr	Leu	Gly	Ile	Gln 3000	Leu	Ser	Pro
Tyr	Asn 3005	Leu	Ser	Thr	Leu	Glu 3010	Asp	Leu	Asn	Thr	Arg 3015	Trp	Lys	Leu
Leu	Gln 3020	Val	Ala	Val	Glu	Asp 3025	Arg	Val	Arg	Gln	Leu 3030	His	Glu	Ala
His	Arg 3035	Asp	Phe	Gly	Pro	Ala 3040	Ser	Gln	His	Phe	Leu 3045	Ser	Thr	Ser
Val	Gln 3050	Gly	Pro	Trp	Glu	Arg 3055	Ala	Ile	Ser	Pro	Asn 3060	Lys	Val	Pro
Tyr	Tyr 3065	Ile	Asn	His		Thr 3070		Thr	Thr	CÀa	Trp 3075	Asp	His	Pro
ГÀа	Met 3080	Thr	Glu	Leu	Tyr	Gln 3085	Ser	Leu	Ala	Asp	Leu 3090	Asn	Asn	Val
Arg	Phe 3095	Ser	Ala	Tyr	Arg	Thr 3100	Ala	Met	Lys	Leu	Arg 3105	Arg	Leu	Gln
Lys	Ala 3110	Leu	Cys	Leu	Asp	Leu 3115	Leu	Ser	Leu	Ser	Ala 3120	Ala	Cys	Asp
Ala	Leu 3125	Asp	Gln	His	Asn	Leu 3130	Lys	Gln	Asn	Asp	Gln 3135	Pro	Met	Asp
Ile	Leu 3140	Gln	Ile	Ile	Asn	Cys 3145	Leu	Thr	Thr	Ile	Tyr 3150	Asp	Arg	Leu
Glu	Gln 3155	Glu	His	Asn	Asn	Leu 3160	Val	Asn	Val	Pro	Leu 3165	Сув	Val	Asp
Met	Cys 3170	Leu	Asn	Trp	Leu	Leu 3175	Asn	Val	Tyr	Asp	Thr 3180	Gly	Arg	Thr
Gly	Arg	Ile	Arg	Val	Leu	Ser	Phe	Lys	Thr	Gly	Ile	Ile	Ser	Leu

	3185					3190					3195			
Cys	Lys 3200	Ala	His	Leu	Glu	Asp 3205	Lys	Tyr	Arg	Tyr	Leu 3210	Phe	Lys	Gln
Val	Ala 3215	Ser	Ser	Thr	Gly	Phe 3220		Asp	Gln	Arg	Arg 3225	Leu	Gly	Leu
Leu	Leu 3230	His	Asp	Ser	Ile	Gln 3235		Pro	Arg	Gln	Leu 3240	Gly	Glu	Val
Ala	Ser 3245	Phe	Gly	Gly	Ser	Asn 3250		Glu	Pro	Ser	Val 3255	Arg	Ser	CÀa
Phe	Gln 3260	Phe	Ala	Asn	Asn	Lys 3265	Pro	Glu	Ile	Glu	Ala 3270	Ala	Leu	Phe
Leu	Asp 3275	Trp	Met	Arg	Leu	Glu 3280	Pro	Gln	Ser	Met	Val 3285	Trp	Leu	Pro
Val	Leu 3290	His	Arg	Val	Ala	Ala 3295	Ala	Glu	Thr	Ala	3300 Lys	His	Gln	Ala
Lys	Cys 3305	Asn	Ile	CÀa	ГÀа	Glu 3310	CÀa	Pro	Ile	Ile	Gly 3315	Phe	Arg	Tyr
Arg	Ser 3320	Leu	ГÀа	His	Phe	Asn 3325		Asp	Ile	CÀa	Gln 3330	Ser	CÀa	Phe
Phe	Ser 3335	Gly	Arg	Val	Ala	Lys 3340	Gly	His	ГÀв	Met	His 3345	Tyr	Pro	Met
Val	Glu 3350	Tyr	CAa	Thr	Pro	Thr 3355	Thr	Ser	Gly	Glu	3360 Yab	Val	Arg	Asp
Phe	Ala 3365	ГÀа	Val	Leu	ГÀв	Asn 3370	ГÀа	Phe	Arg	Thr	Lys 3375	Arg	Tyr	Phe
Ala	3380 TÀa	His	Pro	Arg	Met	Gly 3385	Tyr	Leu	Pro	Val	Gln 3390	Thr	Val	Leu
Glu	Gly 3395	Asp	Asn	Met	Glu	Thr 3400	Pro	Val	Thr	Leu	Ile 3405	Asn	Phe	Trp
Pro	Val 3410	Asp	Ser	Ala	Pro	Ala 3415	Ser	Ser	Pro	Gln	Leu 3420	Ser	His	Asp
Asp	Thr 3425	His	Ser	Arg	Ile	Glu 3430	His	Tyr	Ala	Ser	Arg 3435	Leu	Ala	Glu
Met	Glu 3440	Asn	Ser	Asn	Gly	Ser 3445	Tyr	Leu	Asn	Asp	Ser 3450	Ile	Ser	Pro
Asn	Glu 3455	Ser	Ile	Asp	Asp	Glu 3460		Leu	Leu	Ile	Gln 3465	His	Tyr	СЛа
Gln	Ser 3470	Leu	Asn	Gln	Asp	Ser 3475	Pro	Leu	Ser	Gln	Pro 3480	Arg	Ser	Pro
Ala	Gln 3485	Ile	Leu	Ile	Ser	Leu 3490	Glu	Ser	Glu	Glu	Arg 3495	Gly	Glu	Leu
Glu	Arg 3500	Ile	Leu	Ala	Asp	Leu 3505	Glu	Glu	Glu	Asn	Arg 3510	Asn	Leu	Gln
Ala	Glu 3515	Tyr	Asp	Arg	Leu	Lys 3520	Gln	Gln	His	Glu	His 3525	Lys	Gly	Leu
Ser	Pro 3530	Leu	Pro	Ser	Pro	Pro 3535	Glu	Met	Met	Pro	Thr 3540	Ser	Pro	Gln
Ser	Pro 3545	Arg	Asp	Ala	Glu	Leu 3550	Ile	Ala	Glu	Ala	Lув 3555	Leu	Leu	Arg
Gln	His 3560	Lys	Gly	Arg	Leu	Glu 3565	Ala	Arg	Met	Gln	Ile 3570	Leu	Glu	Asp
His	Asn 3575	Lys	Gln	Leu	Glu	Ser 3580		Leu	His	Arg	Leu 3585	Arg	Gln	Leu

-continued

Leu Glu Gln Pro Gln Ala Glu Ala Lys Val Asn Gly Thr Thr Val 3595 Ser Ser Pro Ser Thr Ser Leu Gln Arg Ser Asp Ser Ser Gln Pro 3605 3610 Met Leu Leu Arg Val Val Gly Ser Gln Thr Ser Asp Ser Met Gly Glu Glu Asp Leu Leu Ser Pro Pro Gln Asp Thr Ser Thr Gly Leu 3640 Glu Glu Val Met Glu Gln Leu Asn Asn Ser Phe Pro Ser Ser Arg 3655 Gly Arg Asn Thr Pro Gly Lys Pro Met Arg Glu Asp Thr Met 3670 <210> SEQ ID NO 5 <211> LENGTH: 536 <212> TYPE: PRT <213 > ORGANISM: Homo sapiens <400> SEQUENCE: 5 Met Glu Phe Ser Ser Pro Ser Arg Glu Glu Cys Pro Lys Pro Leu Ser 10 Arg Val Ser Ile Met Ala Gly Ser Leu Thr Gly Leu Leu Leu Gln 25 Ala Val Ser Trp Ala Ser Gly Ala Arg Pro Cys Ile Pro Lys Ser Phe 40 Gly Tyr Ser Ser Val Val Cys Val Cys Asn Ala Thr Tyr Cys Asp Ser Phe Asp Pro Pro Thr Phe Pro Ala Leu Gly Thr Phe Ser Arg Tyr Glu Ser Thr Arg Ser Gly Arg Arg Met Glu Leu Ser Met Gly Pro Ile Gln Ala Asn His Thr Gly Thr Gly Leu Leu Leu Thr Leu Gln Pro Glu Gln Lys Phe Gln Lys Val Lys Gly Phe Gly Gly Ala Met Thr Asp Ala Ala Ala Leu Asn Ile Leu Ala Leu Ser Pro Pro Ala Gln Asn Leu Leu Leu Lys Ser Tyr Phe Ser Glu Glu Gly Ile Gly Tyr Asn Ile Ile Arg Val Pro Met Ala Ser Cys Asp Phe Ser Ile Arg Thr Tyr Thr Tyr Ala Asp Thr Pro Asp Asp Phe Gln Leu His Asn Phe Ser Leu Pro Glu Glu Asp Thr Lys Leu Lys Ile Pro Leu Ile His Arg Ala Leu Gln Leu Ala Gln Arg Pro Val Ser Leu Leu Ala Ser Pro Trp Thr Ser Pro Thr Trp Leu 215 Lys Thr Asn Gly Ala Val Asn Gly Lys Gly Ser Leu Lys Gly Gln Pro 230 235 Gly Asp Ile Tyr His Gln Thr Trp Ala Arg Tyr Phe Val Lys Phe Leu Asp Ala Tyr Ala Glu His Lys Leu Gln Phe Trp Ala Val Thr Ala Glu Asn Glu Pro Ser Ala Gly Leu Leu Ser Gly Tyr Pro Phe Gln Cys Leu

69 70

-continued

280

Gly	Phe 290	Thr	Pro	Glu	His	Gln 295	Arg	Asp	Phe	Ile	Ala 300	Arg	Asp	Leu	Gly
Pro 305	Thr	Leu	Ala	Asn	Ser 310	Thr	His	His	Asn	Val 315	Arg	Leu	Leu	Met	Leu 320
Asp	Asp	Gln	Arg	Leu 325	Leu	Leu	Pro	His	Trp 330	Ala	Lys	Val	Val	Leu 335	Thr
Asp	Pro	Glu	Ala 340	Ala	Lys	Tyr	Val	His 345	Gly	Ile	Ala	Val	His 350	Trp	Tyr
Leu	Asp	Phe 355	Leu	Ala	Pro	Ala	360	Ala	Thr	Leu	Gly	Glu 365	Thr	His	Arg
Leu	Phe 370	Pro	Asn	Thr	Met	Leu 375	Phe	Ala	Ser	Glu	Ala 380	CÀa	Val	Gly	Ser
382 Tàa	Phe	Trp	Glu	Gln	Ser 390	Val	Arg	Leu	Gly	Ser 395	Trp	Asp	Arg	Gly	Met 400
Gln	Tyr	Ser	His	Ser 405	Ile	Ile	Thr	Asn	Leu 410	Leu	Tyr	His	Val	Val 415	Gly
Trp	Thr	Asp	Trp 420	Asn	Leu	Ala	Leu	Asn 425	Pro	Glu	Gly	Gly	Pro 430	Asn	Trp
Val	Arg	Asn 435	Phe	Val	Asp	Ser	Pro 440	Ile	Ile	Val	Asp	Ile 445	Thr	Lys	Asp
Thr	Phe 450	Tyr	Lys	Gln	Pro	Met 455	Phe	Tyr	His	Leu	Gly 460	His	Phe	Ser	Lys
Phe 465	Ile	Pro	Glu	Gly	Ser 470	Gln	Arg	Val	Gly	Leu 475	Val	Ala	Ser	Gln	Lys 480
Asn	Asp	Leu	Asp	Ala 485	Val	Ala	Leu	Met	His 490	Pro	Asp	Gly	Ser	Ala 495	Val
Val	Val	Val	Leu 500	Asn	Arg	Ser	Ser	Lys 505	Asp	Val	Pro	Leu	Thr 510	Ile	Lys
Asp	Pro	Ala 515	Val	Gly	Phe	Leu	Glu 520	Thr	Ile	Ser	Pro	Gly 525	Tyr	Ser	Ile
His	Thr 530	Tyr	Leu	Trp	Arg	Arg 535	Gln								
			ои с												
<212	2 > T	PE:				,									
			ISM:		sar	piens	3								
			ICE:												
Met 1	Ala	Leu	Trp	Met 5	Arg	Leu	Leu	Pro	Leu 10	Leu	Ala	Leu	Leu	Ala 15	Leu
Trp	Gly	Pro	Asp 20	Pro	Ala	Ala	Ala	Phe 25	Val	Asn	Gln	His	Leu 30	CÀa	Gly
Ser	His	Leu 35	Val	Glu	Ala	Leu	Tyr 40	Leu	Val	Cys	Gly	Glu 45	Arg	Gly	Phe
Phe	Tyr 50	Thr	Pro	Lys	Thr	Arg 55	Arg	Glu	Ala	Glu	Asp	Leu	Gln	Val	Gly
Gln 65	Val	Glu	Leu	Gly	Gly 70	Gly	Pro	Gly	Ala	Gly 75	Ser	Leu	Gln	Pro	Leu 80
Ala	Leu	Glu	Gly	Ser 85	Leu	Gln	Lys	Arg	Gly 90	Ile	Val	Glu	Gln	Сув 95	Cys
Thr	Ser	Ile	Сув 100	Ser	Leu	Tyr	Gln	Leu 105	Glu	Asn	Tyr	CAa	Asn 110		

-continued

<210> SEQ ID NO 7 <211> LENGTH: 1065 <212> TYPE: PRT <213 > ORGANISM: Homo sapiens <400> SEQUENCE: 7 Met Lys Ile Leu Ile Leu Gly Ile Phe Leu Phe Leu Cys Ser Thr Pro Ala Trp Ala Lys Glu Lys His Tyr Tyr Ile Gly Ile Ile Glu Thr Thr Trp Asp Tyr Ala Ser Asp His Gly Glu Lys Lys Leu Ile Ser Val Asp Thr Glu His Ser Asn Ile Tyr Leu Gln Asn Gly Pro Asp Arg Ile Gly 50Arg Leu Tyr Lys Lys Ala Leu Tyr Leu Gln Tyr Thr Asp Glu Thr Phe 65 70 75 80 Arg Thr Thr Ile Glu Lys Pro Val Trp Leu Gly Phe Leu Gly Pro Ile Ile Lys Ala Glu Thr Gly Asp Lys Val Tyr Val His Leu Lys Asn Leu 105 Ala Ser Arg Pro Tyr Thr Phe His Ser His Gly Ile Thr Tyr Tyr Lys 120 Glu His Glu Gly Ala Ile Tyr Pro Asp Asn Thr Thr Asp Phe Gln Arg Ala Asp Asp Lys Val Tyr Pro Gly Glu Gln Tyr Thr Tyr Met Leu Leu Ala Thr Glu Glu Gln Ser Pro Gly Glu Gly Asp Gly Asn Cys Val Thr Arg Ile Tyr His Ser His Ile Asp Ala Pro Lys Asp Ile Ala Ser Gly 185 Leu Ile Gly Pro Leu Ile Ile Cys Lys Lys Asp Ser Leu Asp Lys Glu Lys Glu Lys His Ile Asp Arg Glu Phe Val Val Met Phe Ser Val Val Asp Glu Asn Phe Ser Trp Tyr Leu Glu Asp Asn Ile Lys Thr Tyr Cys Ser Glu Pro Glu Lys Val Asp Lys Asp Asn Glu Asp Phe Gln Glu Ser Asn Arg Met Tyr Ser Val Asn Gly Tyr Thr Phe Gly Ser Leu Pro Gly Leu Ser Met Cys Ala Glu Asp Arg Val Lys Trp Tyr Leu Phe Gly Met Gly Asn Glu Val Asp Val His Ala Ala Phe Phe His Gly Gln Ala Leu Thr Asn Lys Asn Tyr Arg Ile Asp Thr Ile Asn Leu Phe Pro Ala Thr Leu Phe Asp Ala Tyr Met Val Ala Gln Asn Pro Gly Glu Trp Met Leu Ser Cys Gln Asn Leu Asn His Leu Lys Ala Gly Leu Gln Ala Phe Phe Gln Val Gln Glu Cys Asn Lys Ser Ser Ser Lys Asp Asn Ile Arg Gly 360 Lys His Val Arg His Tyr Tyr Ile Ala Ala Glu Glu Ile Ile Trp Asn

385	Ala	Pro	Ser	Gly	Ile 390	Asp	Ile	Phe	Thr	Lys 395	Glu	Asn	Leu	Thr	Ala 400
Pro	Gly	Ser	Asp	Ser 405	Ala	Val	Phe	Phe	Glu 410	Gln	Gly	Thr	Thr	Arg 415	Ile
Gly	Gly	Ser	Tyr 420	Lys	Lys	Leu	Val	Tyr 425	Arg	Glu	Tyr	Thr	Asp 430	Ala	Ser
Phe	Thr	Asn 435	Arg	Lys	Glu	Arg	Gly 440	Pro	Glu	Glu	Glu	His 445	Leu	Gly	Ile
Leu	Gly 450	Pro	Val	Ile	Trp	Ala 455	Glu	Val	Gly	Asp	Thr 460	Ile	Arg	Val	Thr
Phe 465	His	Asn	Lys	Gly	Ala 470	Tyr	Pro	Leu	Ser	Ile 475	Glu	Pro	Ile	Gly	Val 480
Arg	Phe	Asn	Lys	Asn 485	Asn	Glu	Gly	Thr	Tyr 490	Tyr	Ser	Pro	Asn	Tyr 495	Asn
Pro	Gln	Ser	Arg 500	Ser	Val	Pro	Pro	Ser 505	Ala	Ser	His	Val	Ala 510	Pro	Thr
Glu	Thr	Phe 515	Thr	Tyr	Glu	Trp	Thr 520	Val	Pro	ГЛа	Glu	Val 525	Gly	Pro	Thr
Asn	Ala 530	Asp	Pro	Val	CAa	Leu 535	Ala	Lys	Met	Tyr	Tyr 540	Ser	Ala	Val	Asp
Pro 545	Thr	Lys	Asp	Ile	Phe 550	Thr	Gly	Leu	Ile	Gly 555	Pro	Met	Lys	Ile	560 560
Lys	Lys	Gly	Ser	Leu 565	His	Ala	Asn	Gly	Arg 570	Gln	Lys	Asp	Val	Asp 575	Lys
Glu	Phe	Tyr	Leu 580	Phe	Pro	Thr	Val	Phe 585	Asp	Glu	Asn	Glu	Ser 590	Leu	Leu
Leu	Glu	Asp 595	Asn	Ile	Arg	Met	Phe 600	Thr	Thr	Ala	Pro	Asp 605	Gln	Val	Asp
Lys	Glu 610	Asp	Glu	Asp	Phe	Gln 615	Glu	Ser	Asn	Lys	Met 620	His	Ser	Met	Asn
Gly 625	Phe	Met	Tyr	Gly	Asn 630	Gln	Pro	Gly	Leu	Thr 635	Met	CAa	Lys	Gly	Asp 640
Ser	Val														
		Val	Trp	Tyr 645	Leu	Phe	Ser	Ala	Gly 650	Asn	Glu	Ala	Asp	Val 655	His
Gly	Ile			645					650					655	
	Ile Thr	Tyr	Phe 660	645 Ser	Gly	Asn	Thr	Tyr 665	650 Leu	Trp	Arg	Gly	Glu 670	655 Arg	Arg
Asp		Tyr Ala 675	Phe 660 Asn	645 Ser Leu	Gly Phe	Asn Pro	Thr Gln 680	Tyr 665 Thr	650 Leu Ser	Trp Leu	Arg Thr	Gly Leu 685	Glu 670 His	655 Arg Met	Arg Trp
Asp Pro	Thr Asp	Tyr Ala 675 Thr	Phe 660 Asn Glu	645 Ser Leu Gly	Gly Phe Thr	Asn Pro Phe 695	Thr Gln 680 Asn	Tyr 665 Thr Val	650 Leu Ser Glu	Trp Leu Cys	Arg Thr Leu 700	Gly Leu 685 Thr	Glu 670 His	Arg Met Asp	Arg Trp His
Asp Pro Tyr 705	Thr Asp 690	Tyr Ala 675 Thr	Phe 660 Asn Glu	645 Ser Leu Gly Met	Gly Phe Thr Lys 710	Asn Pro Phe 695 Gln	Thr Gln 680 Asn Lys	Tyr 665 Thr Val	650 Leu Ser Glu Thr	Trp Leu Cys Val 715	Arg Thr Leu 700 Asn	Gly Leu 685 Thr	Glu 670 His Thr	Arg Met Asp	Arg Trp His Arg 720
Asp Pro Tyr 705 Gln	Thr Asp 690 Thr	Tyr Ala 675 Thr Gly	Phe 660 Asn Glu Gly	645 Ser Leu Gly Met	Gly Phe Thr Lys 710	Asn Pro Phe 695 Gln Phe	Thr Gln 680 Asn Lys Tyr	Tyr 665 Thr Val Tyr Leu	650 Leu Ser Glu Thr	Trp Leu Cys Val 715 Glu	Arg Thr Leu 700 Asn	Gly Leu 685 Thr Gln	Glu 670 His Thr Cys	Arg Met Asp Arg Tyr 735	Arg Trp His Arg 720 Ile
Asp Pro Tyr 705 Gln	Thr Asp 690 Thr	Tyr Ala 675 Thr Gly Glu Val	Phe 660 Asn Glu Gly Asp Glu 740	645 Ser Leu Gly Met Ser 725 Val	Gly Phe Thr Lys 710 Thr	Asn Pro Phe 695 Gln Phe Trp	Thr Gln 680 Asn Lys Tyr	Tyr 665 Thr Val Tyr Leu	650 Leu Ser Glu Thr Gly 730 Ser	Trp Leu Cys Val 715 Glu Pro	Arg Thr Leu 700 Asn Arg	Gly Leu 685 Thr Gln Thr	Glu 670 His Thr Cys Tyr	Arg Met Asp Arg Tyr 735	Arg Trp His Arg 720 Ile Glu
Asp Pro Tyr 705 Gln Ala	Thr Asp 690 Thr Ser	Tyr Ala 675 Thr Gly Glu Val Leu 755	Phe 660 Asn Glu Gly Asp Glu 740	645 Ser Leu Gly Met Ser 725 Val	Gly Phe Thr Lys 710 Thr Glu Leu	Asn Pro Phe 695 Gln Phe	Thr Gln 680 Asn Lys Tyr Asp Glu 760	Tyr 665 Thr Val Tyr Leu Tyr 745 Gln	650 Leu Ser Glu Thr Gly 730 Ser Asn	Trp Leu Cys Val 715 Glu Pro Val	Arg Thr Leu 700 Asn Arg Gln Ser	Gly Leu 685 Thr Gln Thr Arg Asn 765	Glu 670 His Thr Cys Tyr Glu 750 Ala	Asp Arg Tyr 735 Trp Phe	Arg Trp His Arg 720 Ile Glu Leu

-continued

Glu Glu Glu His Leu Gly Ile Leu Gly Pro Gln Leu His Ala Asp Val 805 810 Gly Asp Lys Val Lys Ile Ile Phe Lys Asn Met Ala Thr Arg Pro Tyr Ser Ile His Ala His Gly Val Gln Thr Glu Ser Ser Thr Val Thr Pro Thr Leu Pro Gly Glu Thr Leu Thr Tyr Val Trp Lys Ile Pro Glu Arg Ser Gly Ala Gly Thr Glu Asp Ser Ala Cys Ile Pro Trp Ala Tyr Tyr Ser Thr Val Asp Gln Val Lys Asp Leu Tyr Ser Gly Leu Ile Gly Pro Leu Ile Val Cys Arg Arg Pro Tyr Leu Lys Val Phe Asn Pro Arg Arg Lys Leu Glu Phe Ala Leu Leu Phe Leu Val Phe Asp Glu Asn Glu Ser Trp Tyr Leu Asp Asp Asn Ile Lys Thr Tyr Ser Asp His Pro Glu Lys Val Asn Lys Asp Asp Glu Glu Phe Ile Glu Ser Asn Lys Met His Ala 955 Ile Asn Gly Arg Met Phe Gly Asn Leu Gln Gly Leu Thr Met His Val 970 Gly Asp Glu Val Asn Trp Tyr Leu Met Gly Met Gly Asn Glu Ile Asp 980 985 Leu His Thr Val His Phe His Gly His Ser Phe Gln Tyr Lys His Arg 1000 Gly Val Tyr Ser Ser Asp Val Phe Asp Ile Phe Pro Gly Thr Tyr 1015 1020 Gln Thr Leu Glu Met Phe Pro Arg Thr Pro Gly Ile Trp Leu Leu 1030 His Cys $\,$ His Val Thr Asp His $\,$ Ile His Ala Gly Met $\,$ Glu Thr Thr $\,$ 1045 Tyr Thr Val Leu Gln Asn Glu Asp Thr Lys Ser Gly 1055 1060 <210> SEQ ID NO 8 <211> LENGTH: 471 <212> TYPE: PRT <213 > ORGANISM: Escherichia coli <400> SEQUENCE: 8 Met Lys Gln Ser Thr Ile Ala Leu Ala Leu Leu Pro Leu Leu Phe Thr Pro Val Thr Lys Ala Arg Thr Pro Glu Met Pro Val Leu Glu Asn Arg Ala Ala Gln Gly Asp Ile Thr Ala Pro Gly Gly Ala Arg Arg Leu Thr 40 Gly Asp Gln Thr Ala Ala Leu Arg Asp Ser Leu Ser Asp Lys Pro Ala Lys Asn Ile Ile Leu Leu Ile Gly Asp Gly Met Gly Asp Ser Glu Ile Thr Ala Ala Arg Asn Tyr Ala Glu Gly Ala Gly Gly Phe Phe Lys Gly 90 Ile Asp Ala Leu Pro Leu Thr Gly Gln Tyr Thr His Tyr Ala Leu Asn 105

Lys Lys Thr Gly Lys Pro Asp Tyr Val Thr Asp Ser Ala Ala Ser Ala 120 Thr Ala Trp Ser Thr Gly Val Lys Thr Tyr Asn Gly Ala Leu Gly Val 135 Asp Ile His Glu Lys Asp His Pro Thr Ile Leu Glu Met Ala Lys Ala Ala Gly Leu Ala Thr Gly Asn Val Ser Thr Ala Glu Leu Gln Asp Ala Thr Pro Ala Ala Leu Val Ala His Val Thr Ser Arg Lys Cys Tyr Gly Pro Ser Ala Thr Ser Glu Lys Cys Pro Gly Asn Ala Leu Glu Lys Gly Gly Lys Gly Ser Ile Thr Glu Gln Leu Leu Asn Ala Arg Ala Asp Val Thr Leu Gly Gly Gly Ala Lys Thr Phe Ala Glu Thr Ala Thr Ala Gly 230 235 Glu Trp Gln Gly Lys Thr Leu Arg Glu Gln Ala Gln Ala Arg Gly Tyr 250 Gln Leu Val Ser Asp Ala Ala Ser Leu Asn Ser Val Thr Glu Ala Asn 265 Gln Gln Lys Pro Leu Leu Gly Leu Phe Ala Asp Gly Asn Met Pro Val 280 Arg Trp Leu Gly Pro Lys Ala Thr Tyr His Gly Asn Ile Asp Lys Pro 295 Ala Val Thr Cys Thr Pro Asn Pro Gln Arg Asn Asp Ser Val Pro Thr 310 315 Leu Ala Gln Met Thr Asp Lys Ala Ile Glu Leu Leu Ser Lys Asn Glu Lys Gly Phe Phe Leu Gln Val Glu Gly Ala Ser Ile Asp Lys Gln Asp 345 His Ala Ala Asn Pro Cys Gly Gln Ile Gly Glu Thr Val Asp Leu Asp Glu Ala Val Gln Arg Ala Leu Glu Phe Ala Lys Lys Glu Gly Asn Thr Leu Val Ile Val Thr Ala Asp His Ala His Ala Ser Gln Ile Val Ala Pro Asp Thr Lys Ala Pro Gly Leu Thr Gln Ala Leu Asn Thr Lys Asp Gly Ala Val Met Val Met Ser Tyr Gly Asn Ser Glu Glu Asp Ser Gln Glu His Thr Gly Ser Gln Leu Arg Ile Ala Ala Tyr Gly Pro His Ala Ala Asn Val Val Gly Leu Thr Asp Gln Thr Asp Leu Phe Tyr Thr Met Lys Ala Ala Leu Gly Leu Lys <210> SEQ ID NO 9 <211> LENGTH: 1024 <212> TYPE: PRT <213 > ORGANISM: Escherichia coli

<400> SEQUENCE: 9

Met Thr Met Ile Thr Asp Ser Leu Ala Val Val Leu Gln Arg Arg Asp

1	5			10			15	
Trp Glu As	n Pro Gly 20	Val Thr	Gln Leu 25	Asn Arg	Leu Al	a Ala 30	His	Pro
Pro Phe Al	a Ser Trp	Arg Asn	Ser Glu 40	Glu Ala	Arg Th	r Asp	Arg	Pro
Ser Gln Gl 50	n Leu Arg	Ser Leu 55	Asn Gly	Glu Trp	Arg Ph	e Ala	Trp	Phe
Pro Ala Pr 65	o Glu Ala	Val Pro 70	Glu Ser	Trp Leu 75	Glu Cy	s Asp	Leu	Pro 80
Glu Ala As	p Thr Val 85	Val Val	Pro Ser	Asn Trp 90	Gln Me	t His	Gly 95	Tyr
Asp Ala Pr	o Ile Tyr 100	Thr Asn	Val Thr 105		Ile Th	r Val 110	Asn	Pro
Pro Phe Va		Glu Asn	Pro Thr 120	Gly Cys	Tyr Se		Thr	Phe
Asn Val As	p Glu Ser	Trp Leu 135	Gln Glu	Gly Gln	Thr Ar	g Ile	Ile	Phe
Asp Gly Va 145	l Asn Ser	Ala Phe 150	His Leu	Trp Cys 155	Asn Gl	y Arg	Trp	Val 160
Gly Tyr Gl	y Gln Asp 165	_	Leu Pro	Ser Glu 170	Phe As	p Leu	Ser 175	Ala
Phe Leu Ar	g Ala Gly 180	Glu Asn	Arg Leu 185		Met Va	l Leu 190	Arg	Trp
Ser Asp Gl		Leu Glu	Asp Gln 200	Asp Met	Trp Ar 20		Ser	Gly
Ile Phe Ar 210	g Asp Val	Ser Leu 215	Leu His	Lys Pro	Thr Th	r Gln	Ile	Ser
Asp Phe Hi 225	s Val Ala	Thr Arg 230	Phe Asn	Asp Asp 235	Phe Se	r Arg	Ala	Val 240
Leu Glu Al	a Glu Val 245		Cys Gly	Glu Leu 250	Arg As	p Tyr	Leu 255	Arg
Val Thr Va	1 Ser Leu 260	Trp Gln	Gly Glu 265		Val Al	a Ser 270	Gly	Thr
Ala Pro Ph 27		Glu Ile	Ile Asp 280	Glu Arg	Gly Gl 28		Ala	Asp
Arg Val Th	r Leu Arg	Leu Asn 295		Asn Pro	Lys Le	u Trp	Ser	Ala
Glu Ile Pr 305	o Asn Leu	Tyr Arg 310	Ala Val	Val Glu 315	Leu Hi	s Thr	Ala	Asp 320
Gly Thr Le	u Ile Glu 325		Ala Cys	Asp Val	Gly Ph	e Arg	Glu 335	Val
Arg Ile Gl	u Asn Gly 340	Leu Leu	Leu Leu 345		Lys Pr	o Leu 350	Leu	Ile
Arg Gly Va	_	His Glu	His His 360	Pro Leu	His Gl 36	_	Val	Met
Asp Glu Gl 370	n Thr Met	Val Gln 375	_	Leu Leu	Met Ly 380	s Gln	Asn	Asn
Phe Asn Al 385	a Val Arg	Cys Ser 390	His Tyr	Pro Asn 395	His Pr	o Leu	Trp	Tyr 400
Thr Leu Cy	s Asp Arg 405		Leu Tyr	Val Val 410	Asp Gl	u Ala	Asn 415	Ile
Glu Thr Hi	s Gly Met 420	Val Pro	Met Asn 425	_	Thr As	p Asp 430	Pro	Arg

Trp	Leu	Pro 435	Ala	Met	Ser	Glu	Arg 440	Val	Thr	Arg	Met	Val	Gln	Arg	Asp
Arg	Asn 450	His	Pro	Ser	Val	Ile 455		Trp	Ser	Leu	Gly 460		Glu	Ser	Gly
His 465	Gly	Ala	Asn	His	Asp 470	Ala	Leu	Tyr	Arg	Trp 475	Ile	Lys	Ser	Val	Asp 480
Pro	Ser	Arg	Pro	Val 485	Gln	Tyr	Glu	Gly	Gly 490	Gly	Ala	Asp	Thr	Thr 495	Ala
Thr	Asp	Ile	Ile 500	СЛа	Pro	Met	Tyr	Ala 505	Arg	Val	Asp	Glu	Asp 510	Gln	Pro
Phe	Pro	Ala 515	Val	Pro	Lys	Trp	Ser 520	Ile	Lys	Lys	Trp	Leu 525	Ser	Leu	Pro
Gly	Glu 530	Thr	Arg	Pro	Leu	Ile 535	Leu	Сув	Glu	Tyr	Ala 540	His	Ala	Met	Gly
Asn 545	Ser	Leu	Gly	Gly	Phe 550	Ala	Lys	Tyr	Trp	Gln 555	Ala	Phe	Arg	Gln	Tyr 560
Pro	Arg	Leu	Gln	Gly 565	Gly	Phe	Val	Trp	Asp 570	Trp	Val	Asp	Gln	Ser 575	Leu
Ile	Lys	Tyr	Asp 580	Glu	Asn	Gly	Asn	Pro 585	Trp	Ser	Ala	Tyr	Gly 590	Gly	Asp
Phe	Gly	Asp 595	Thr	Pro	Asn	Asp	Arg 600	Gln	Phe	Cys	Met	Asn 605	Gly	Leu	Val
Phe	Ala 610	Asp	Arg	Thr	Pro	His 615	Pro	Ala	Leu	Thr	Glu 620	Ala	Lys	His	Gln
Gln 625	Gln	Phe	Phe	Gln	Phe 630	Arg	Leu	Ser	Gly	Gln 635	Thr	Ile	Glu	Val	Thr 640
Ser	Glu	Tyr	Leu	Phe 645	Arg	His	Ser	Asp	Asn 650	Glu	Leu	Leu	His	Trp 655	Met
Val	Ala	Leu	Asp 660	Gly	ГÀЗ	Pro	Leu	Ala 665	Ser	Gly	Glu	Val	Pro 670	Leu	Asp
Val	Ala	Pro 675	Gln	Gly	Lys	Gln	Leu 680	Ile	Glu	Leu	Pro	Glu 685	Leu	Pro	Gln
Pro	Glu 690	Ser	Ala	Gly	Gln	Leu 695	Trp	Leu	Thr	Val	Arg 700	Val	Val	Gln	Pro
Asn 705	Ala	Thr	Ala	Trp	Ser 710	Glu	Ala	Gly	His	Ile 715	Ser	Ala	Trp	Gln	Gln 720
Trp	Arg	Leu	Ala	Glu 725	Asn	Leu	Ser	Val	Thr 730	Leu	Pro	Ala	Ala	Ser 735	His
Ala	Ile	Pro	His 740	Leu	Thr	Thr	Ser	Glu 745	Met	Asp	Phe	Cys	Ile 750	Glu	Leu
Gly	Asn	Lys 755	Arg	Trp	Gln	Phe	Asn 760	Arg	Gln	Ser	Gly	Phe 765	Leu	Ser	Gln
Met	Trp 770	Ile	Gly	Asp	Lys	Lys 775	Gln	Leu	Leu	Thr	Pro 780	Leu	Arg	Asp	Gln
Phe 785	Thr	Arg	Ala	Pro	Leu 790	Asp	Asn	Asp	Ile	Gly 795	Val	Ser	Glu	Ala	Thr 800
Arg	Ile	Asp	Pro	Asn 805	Ala	Trp	Val	Glu	Arg 810	Trp	ГЛа	Ala	Ala	Gly 815	His
Tyr	Gln	Ala	Glu 820	Ala	Ala	Leu	Leu	Gln 825	Сув	Thr	Ala	Asp	Thr 830	Leu	Ala
Asp	Ala	Val 835	Leu	Ile	Thr	Thr	Ala 840	His	Ala	Trp	Gln	His 845	Gln	Gly	Lys

-continued

_															
Thr	Leu 850	Phe	Ile	Ser	Arg	Lys 855	Thr	Tyr	Arg	Ile	860	Gly	Ser	Gly	Gln
Met 865	Ala	Ile	Thr	Val	Asp 870	Val	Glu	Val	Ala	Ser 875	Asp	Thr	Pro	His	Pro 880
Ala	Arg	Ile	Gly	Leu 885	Asn	CAa	Gln	Leu	Ala 890	Gln	Val	Ala	Glu	Arg 895	Val
Asn	Trp	Leu	Gly 900	Leu	Gly	Pro	Gln	Glu 905	Asn	Tyr	Pro	Asp	Arg 910	Leu	Thr
Ala	Ala	Сув 915	Phe	Asp	Arg	Trp	Asp 920	Leu	Pro	Leu	Ser	Asp 925	Met	Tyr	Thr
Pro	Tyr 930	Val	Phe	Pro	Ser	Glu 935	Asn	Gly	Leu	Arg	Cys 940	Gly	Thr	Arg	Glu
Leu 945	Asn	Tyr	Gly	Pro	His 950	Gln	Trp	Arg	Gly	Asp 955	Phe	Gln	Phe	Asn	Ile 960
Ser	Arg	Tyr	Ser	Gln 965	Gln	Gln	Leu	Met	Glu 970	Thr	Ser	His	Arg	His 975	Leu
Leu	His	Ala	Glu 980	Glu	Gly	Thr	Trp	Leu 985	Asn	Ile	Asp	Gly	Phe 990	His	Met
Gly	Ile	Gly 995	Gly	Asp	Asp	Ser	Trp		r Pro	Se:	r Val	l Se:		la G	lu Phe
Gln	Leu 1010		r Al	a Gl	y Ar	g Ty: 10:		is Ty	yr G	ln L		al ' 020	Irp (Cys (3ln
Lys															

What is claimed is:

1. A microtube comprising:

an electrospun shell,

an electrospun coat polymer over an internal surface of said shell and a molecule-of-interest attached to the microtube,

wherein said electrospun shell comprises pores,

wherein said electrospun shell is formed of a first polymeric solution comprising a first solvent and said electrospun coat is formed of a second polymeric solution comprising a second solvent,

wherein said second solvent of said second polymeric solution is incapable of dissolving the polymer of said 45 first polymeric solution,

wherein said first polymeric solution solidifies faster than said second polymeric solution,

wherein said second polymeric solution is capable of wetting said internal surface of said shell during or 50 following solidification of said first polymeric solution,

wherein said electrospun shell comprises a polymer selected from the group consisting of poly (e-caprolactone) (PCL), polyamide, poly(siloxane), poly(silicone), poly(ethylene), poly(vinyl pyrrolidone), poly(2-hy- 55 droxy ethylmethacrylate), poly(N-vinyl pyrrolidone), poly(methyl methacrylate), poly(vinyl alcohol), poly (acrylic acid), poly(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(methacrylic acid), polylactide, polyglycolide, poly(lactide-cogly- 60 colide), polyanhydride, polyorthoester, poly(carbonate), poly(acrylo nitrile), polyaniline, polyvinyl carbaphenol), polystyrene, poly(vinyl polyhydroxyacid, poly(caprolactone), polyanhydride, polyhydroxyalkanoate, and polyurethane, and whereas 65 said electrospun coat comprises a polymer selected from the group consisting of poly(acrylic acid), poly

(vinyl acetate), polyacrylamide, poly(ethylene-co-vinyl acetate), poly(ethylene glycol), poly(methacrylic acid), polylactide polyglycolide, poly(lactide-coglycolide), polyanhydride, polyorthoester, poly(carbonate), poly (ethylene oxide), polyaniline, polyvinyl carbazole, polystyrene, poly(vinyl phenol), and polyhydroxyacid,

wherein said polymer of said first polymeric solution and said polymer of said second polymeric solution are different.

wherein said molecule-of-interest is selected from the group consisting of: a polypeptide, a polynucleotide, a carbohydrate, a polysaccharide, a lipid, a drug molecule, and a small molecule, and

wherein said small molecule is selected from the group consisting of a nucleotide base, an amino acid, a nucleotide, an antibiotic, and a vitamin.

- 2. The microtube of claim 1, wherein said first solvent of said first polymeric solution evaporates faster than said second solvent of said second polymeric solution.
- 3. The microtube of claim 1, wherein said second solvent of said second polymeric solution is capable of evaporating through said internal surface of said shell.
- **4**. The microtube of claim **1**, wherein a thickness of said shell is from about 100 nm to about 20 micrometer.
- 5. The microtube of claim 1, wherein an internal diameter of the microtube is from about 50 nm to about 20 micrometer.
- 6. The microtube of claim 1, wherein said first and said second polymeric solutions are selected from the group consisting of: 10% poly (e-caprolactone) (PCL) in chloroform (CHCl₃) and dimethylforamide (DMF) (80:20 by weight) as said first polymeric solution and 4% poly(ethylene oxide) (PEO) in water (H₂O) and ethanol (60:40 by weight) as said second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as said first polymeric

solution and 6% PEO in $\rm H_2O$ and ethanol (60:40 by weight) as said second polymeric solution, 9% PCL in CHCl₃ and DMF (90:10 by weight) as said first polymeric solution and 7% PEO in $\rm H_2O$ as said second polymeric solution, 10% PCL in CHCl₃ and DMF (80:20 by weight) as said first 5 polymeric solution and 9% poly(vinyl alcohol) (PVA) in water and ethanol (50:50 by weight) as said second polymeric solution, and 10% PCL in CHCl₃ and DMF (90:10 by weight) as the first polymeric solution and 4% (by weight) PEO in ethanol: $\rm H_2O$ (26:74 by weight) as a second polymeric solution.

- 7. The microtube of claim 1, wherein said microtube is filled with a liquid.
- **8**. The microtube of claim **1**, wherein said molecule-of-interest is attached to said coat over said internal surface of $_{15}$ said shell.
- **9**. The microtube of claim **1**, wherein said cell or said membrane-coated particle-of-interest is attached to said shell of the microtube.

86

- 10. The microtube of claim 1, wherein said first polymeric solution further comprises polyethylene glycol (PEG).
- 11. The microtube of claim 1, wherein said shell prevents diffusion of the molecule-of-interest therethrough.
- 12. A microfluidic device comprising a plurality of the microtubes of claim 1.
- 13. The microtube of claim 1, wherein said microtube is a single or separated microtube.
- 14. The microtube of claim 1, wherein said microtube comprises a plurality of microtubes.
- 15. The microtube of claim 14, wherein said plurality of microtubes form an aligned array.
 - 16. The microtube of claim 7, wherein said liquid is blood.
- 17. The microtube of claim 1, wherein said polymer comprises a co-polymer.
- 18. The microtube of claim 1, wherein said polymer comprises a blend of polymers.

* * * * *